

Transition-Metallocarbene Mediated Tandem Cyclization Reactions and Polynuclear Gold Clusters Synthesis

Dr. Jin-Ming Yang (杨金明)

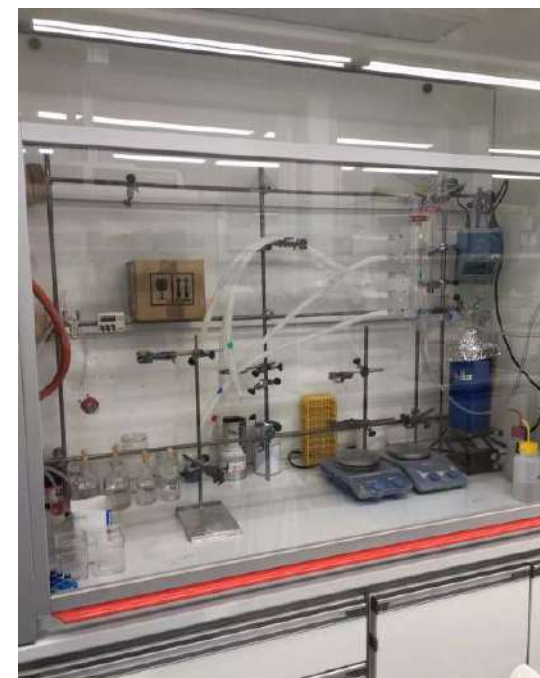
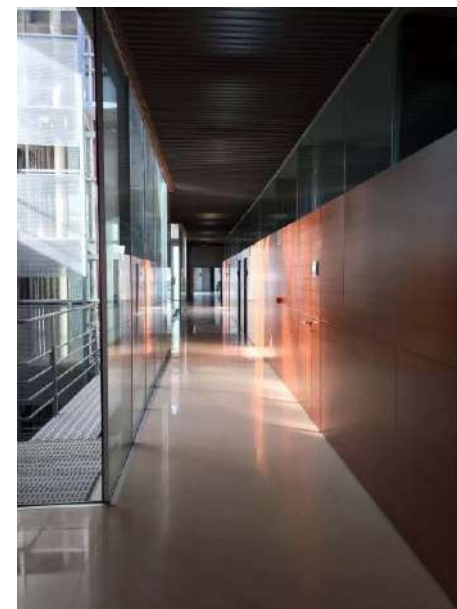
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博后合作导师: Antonio M Echavarren (ICIQ)

05-12-2018

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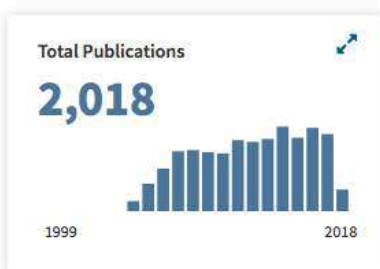
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Computational
Chemistry



CO₂ recycling



Renewable Fuels



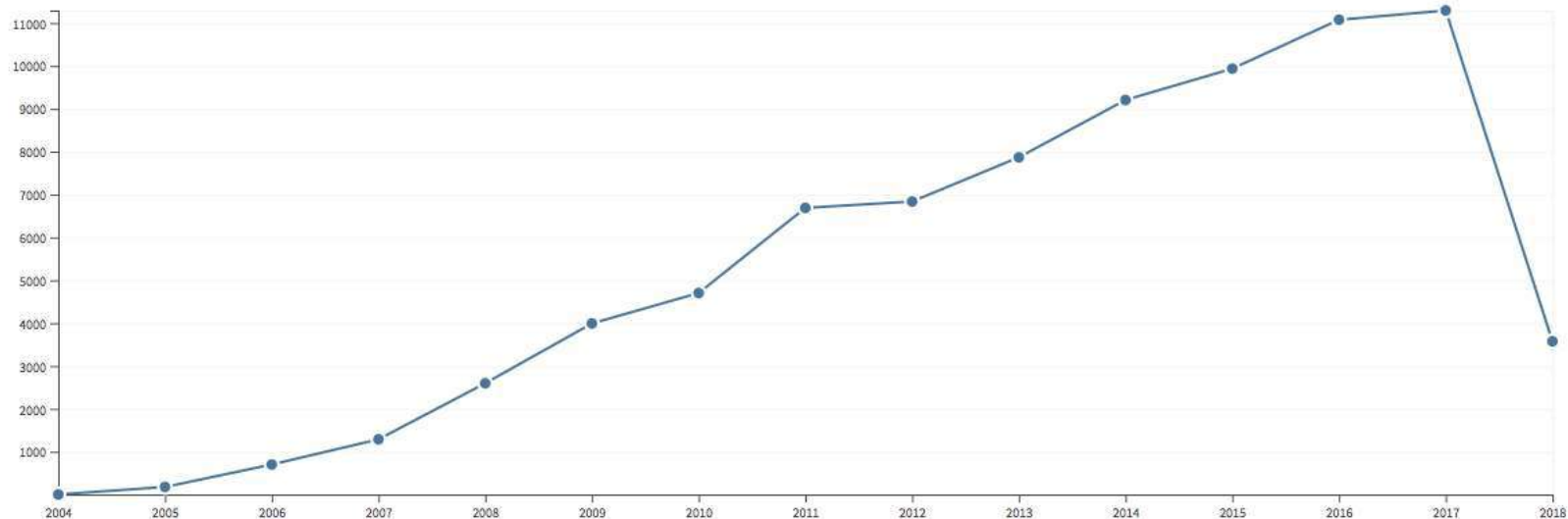
Catalysis



Artificial
Photosynthesis

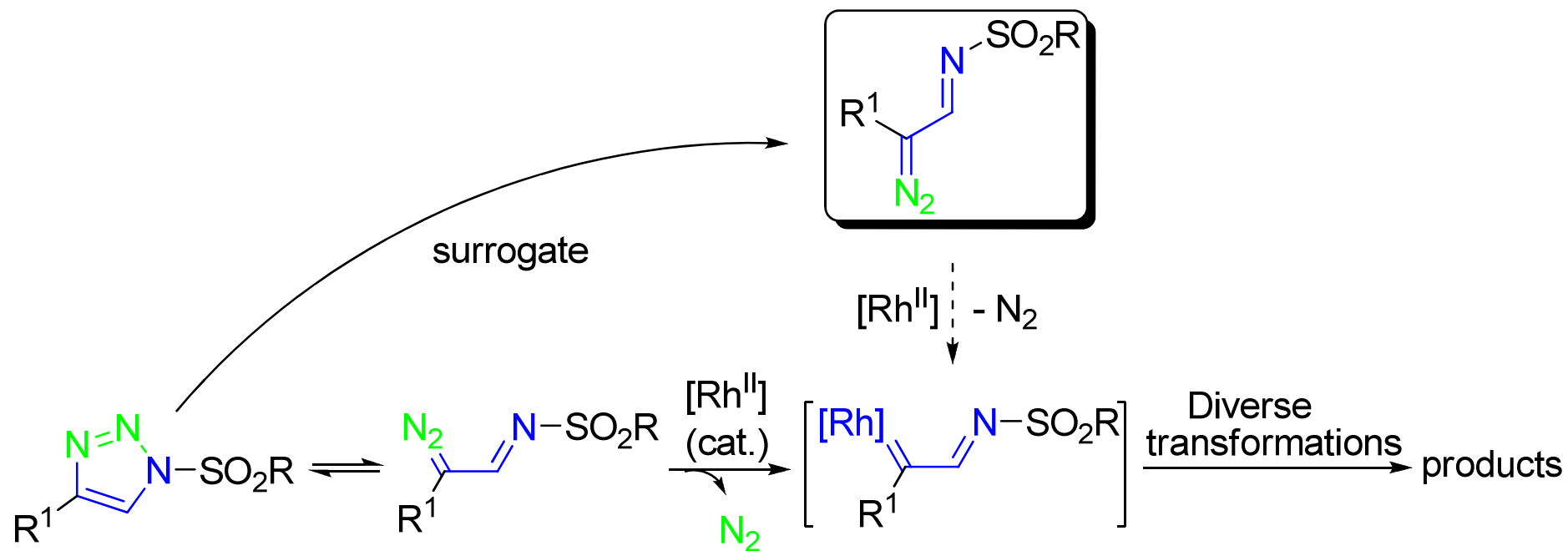


Sum of Times Cited per Year



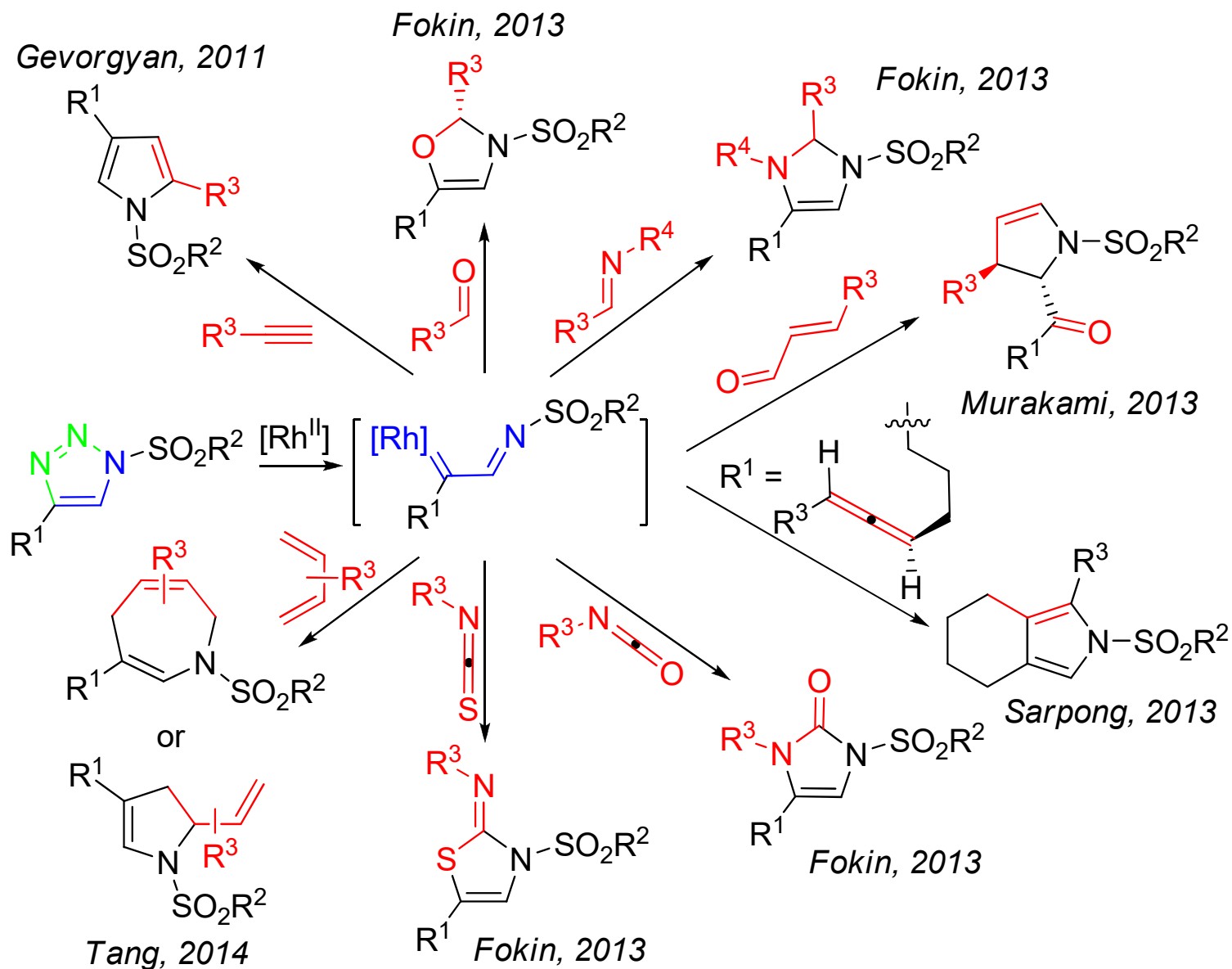
- **Background**
- **1) Rhodium-carbene Mediated Heterocycles Synthesis**
 - **Rhodium(II)-catalyzed intramolecular annulation of 1-sulfonyl-1,2,3-triazoles with pyrrole and indole rings**
- **2) Gold-carbene Mediated Tandem Cyclization Reactions**
 - **Gold(I)-Catalyzed Highly Stereoselective Synthesis of Polycyclic Indolines**
 - **Gold(I)-catalyzed intramolecular cycloisomerization of propargylic esters with furan rings**
- **3) Polynuclear Gold Clusters Synthesis**
- **Research Summary**
- **Acknowledgements**

Triazoles



Gulevich, A. V.; Gevorgyan, V. *Angew. Chem. Int. Ed.* **2013**, *52*, 1371-1373.

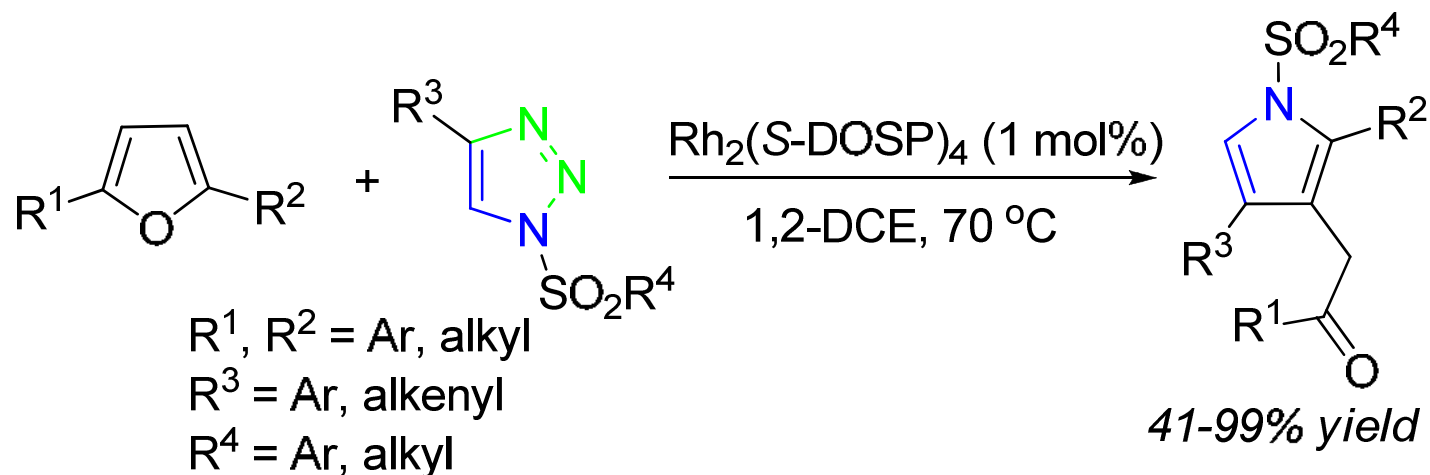
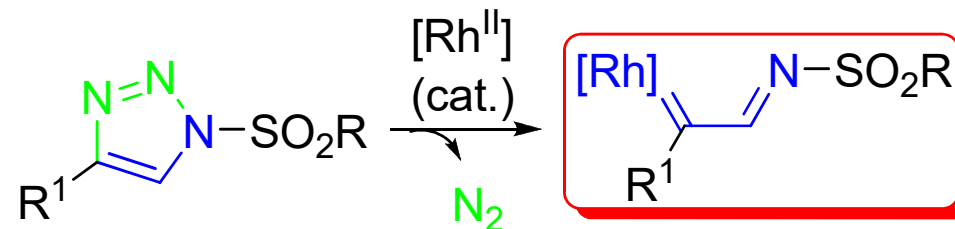
Transformations of Triazoles



(a) Gulevich, A. V.; Gevorgyan, V. *Angew. Chem. Int. Ed.* **2013**, *52*, 1371-1373. (b) Davies, H. M. L.; Alford, J. S. *Chem. Soc. Rev.* **2014**, *43*, 5151-5162.

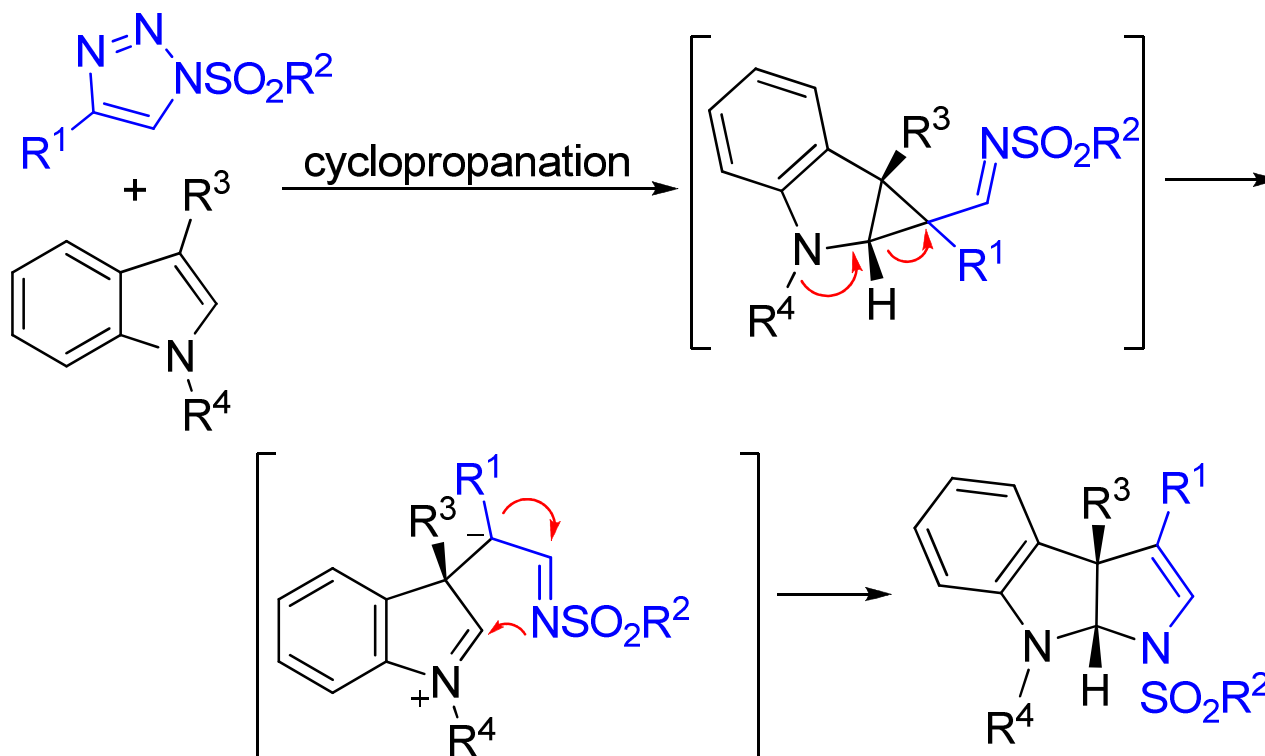
1) Rhodium-carbene Mediated Heterocycles Synthesis

Rh is one of the most popular catalysts for cycloaddition, Pauson-Khand reaction and C–H activation



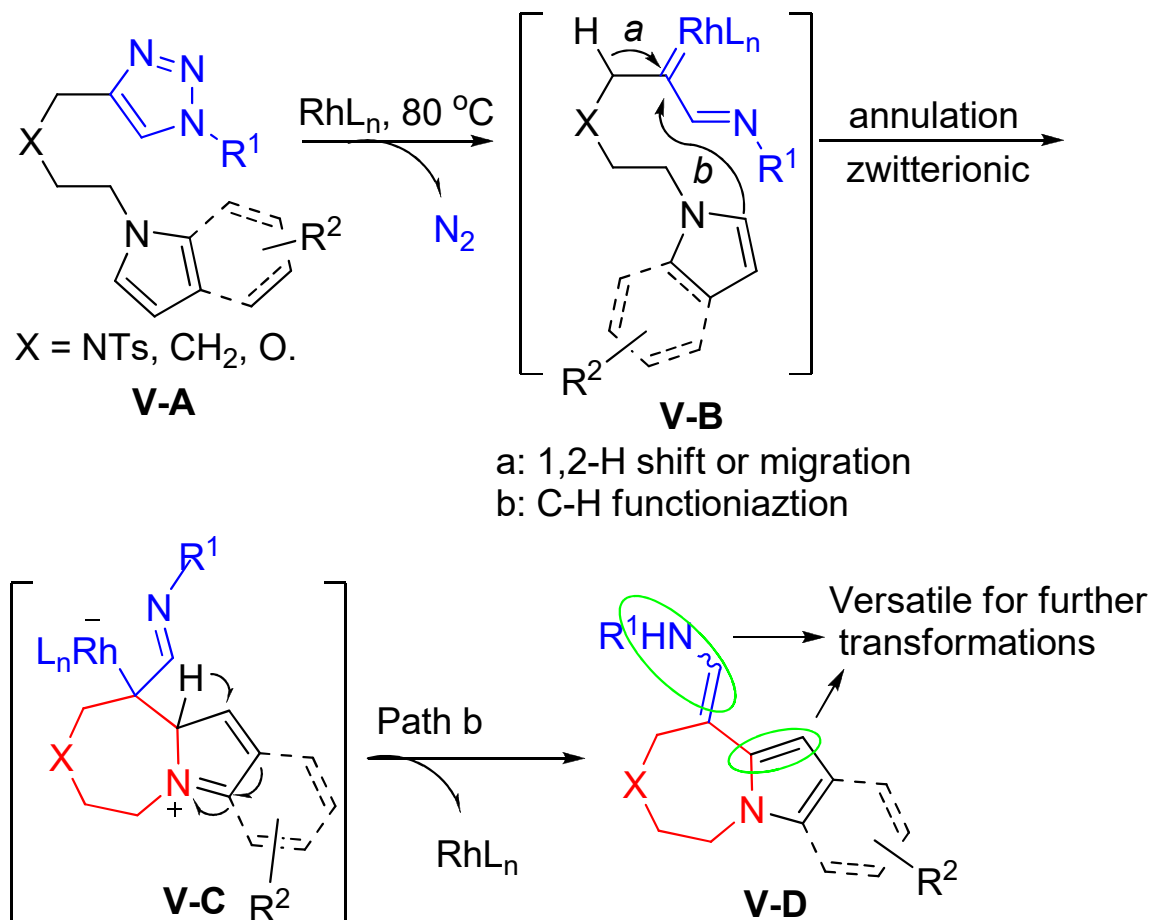
1) Rhodium-carbene Mediated Heterocycles Synthesis

Intermolecular annulation of indoles (Davies's work)



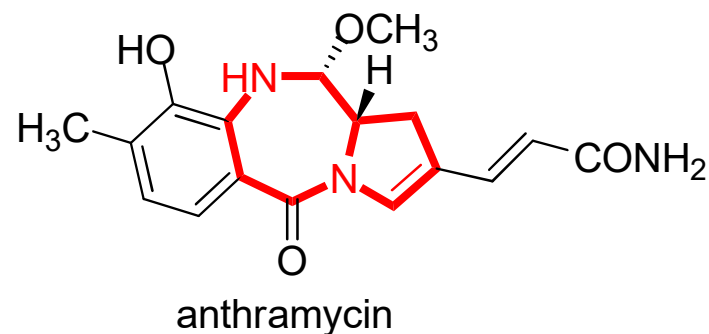
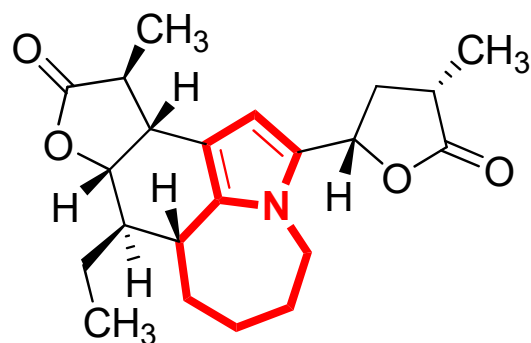
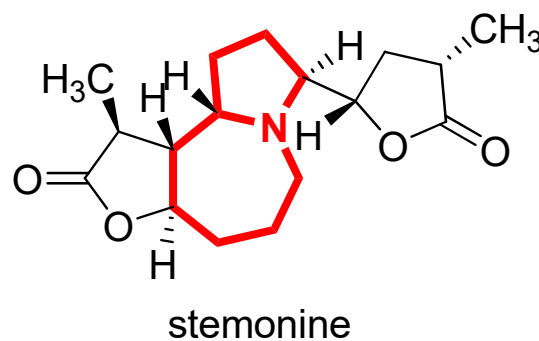
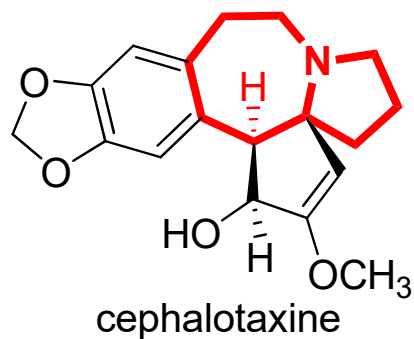
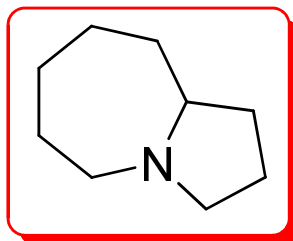
1) Rhodium-carbene Mediated Heterocycles Synthesis

Intramolecular annulation of pyrroles and indoles (This work)



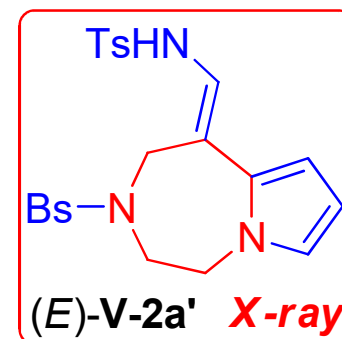
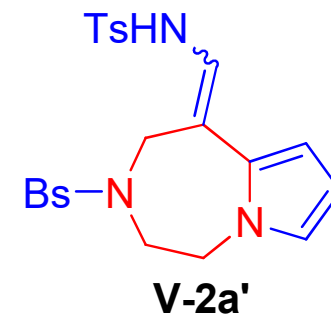
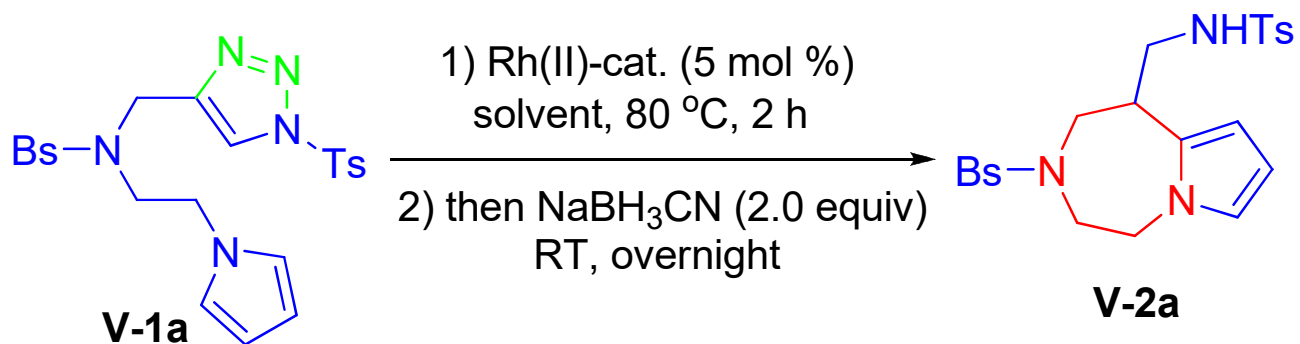
High chemoselectivity (path b only)!
Easy-to-handle functional groups!
Wide azepine ring flexibility (N, O, C)!

1) Rhodium-carbene Mediated Heterocycles Synthesis



bisdehydrotuberostemonine
 bisdehydrotuberostemonine (H-11 α , H-12 α)
 bisdehydrotuberostemonine B (H-10 β)
 bisdehydrotuberostemonine C (H-9 α , H-10 β)

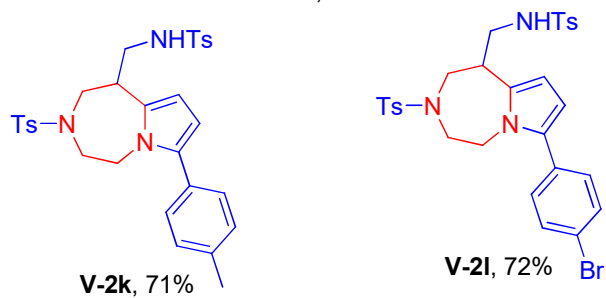
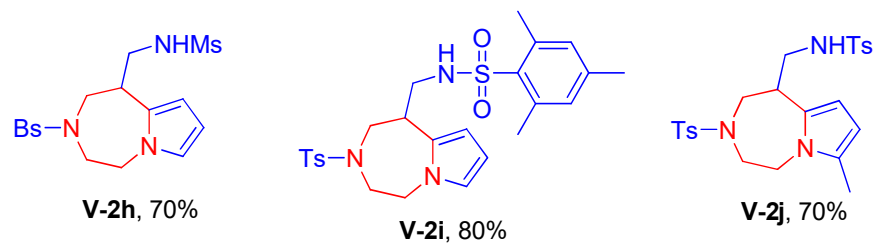
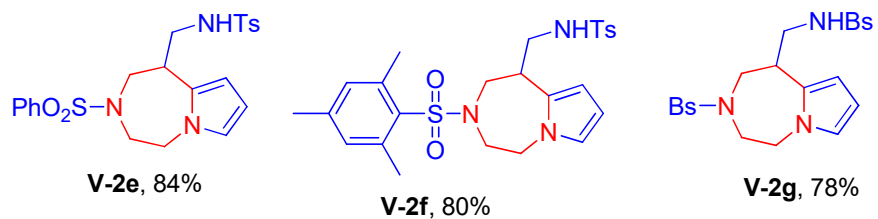
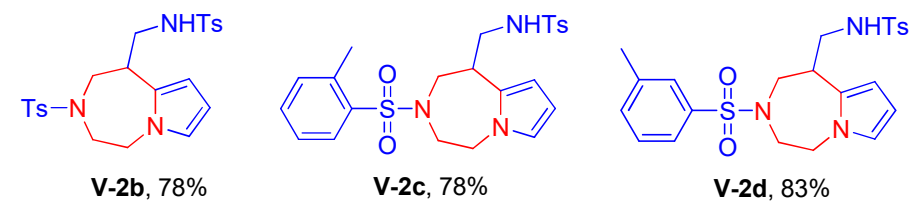
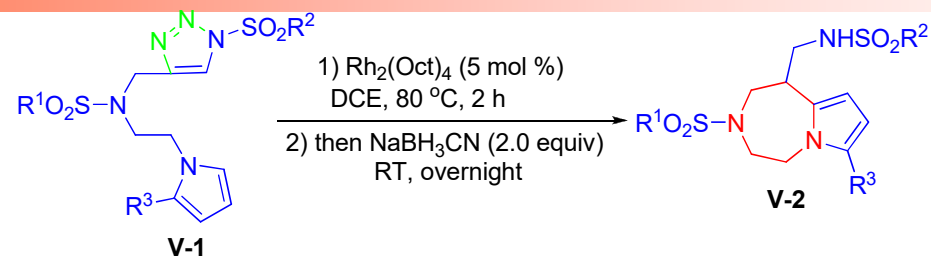
Optimization of the reaction conditions



entry ^a	Rh(II)-cat.	solvent	yield (%) ^b
1	Rh₂(Oct)₄	DCE	86
2	Rh ₂ (Piv) ₄	DCE	80
3	Rh ₂ (esp) ₂	DCE	77
4	Rh ₂ (OAc) ₄	DCE	78
5	Rh ₂ (Adc) ₄	DCE	80
6	Rh ₂ (tfa) ₄	DCE	0
7	Rh ₂ (S-NTTL) ₄	DCE	70
8	Rh ₂ (Oct) ₄	toluene	78
9	Rh ₂ (Oct) ₄	cyclohexane	- ^c
10	Rh ₂ (Oct) ₄	CHCl ₃	- ^c

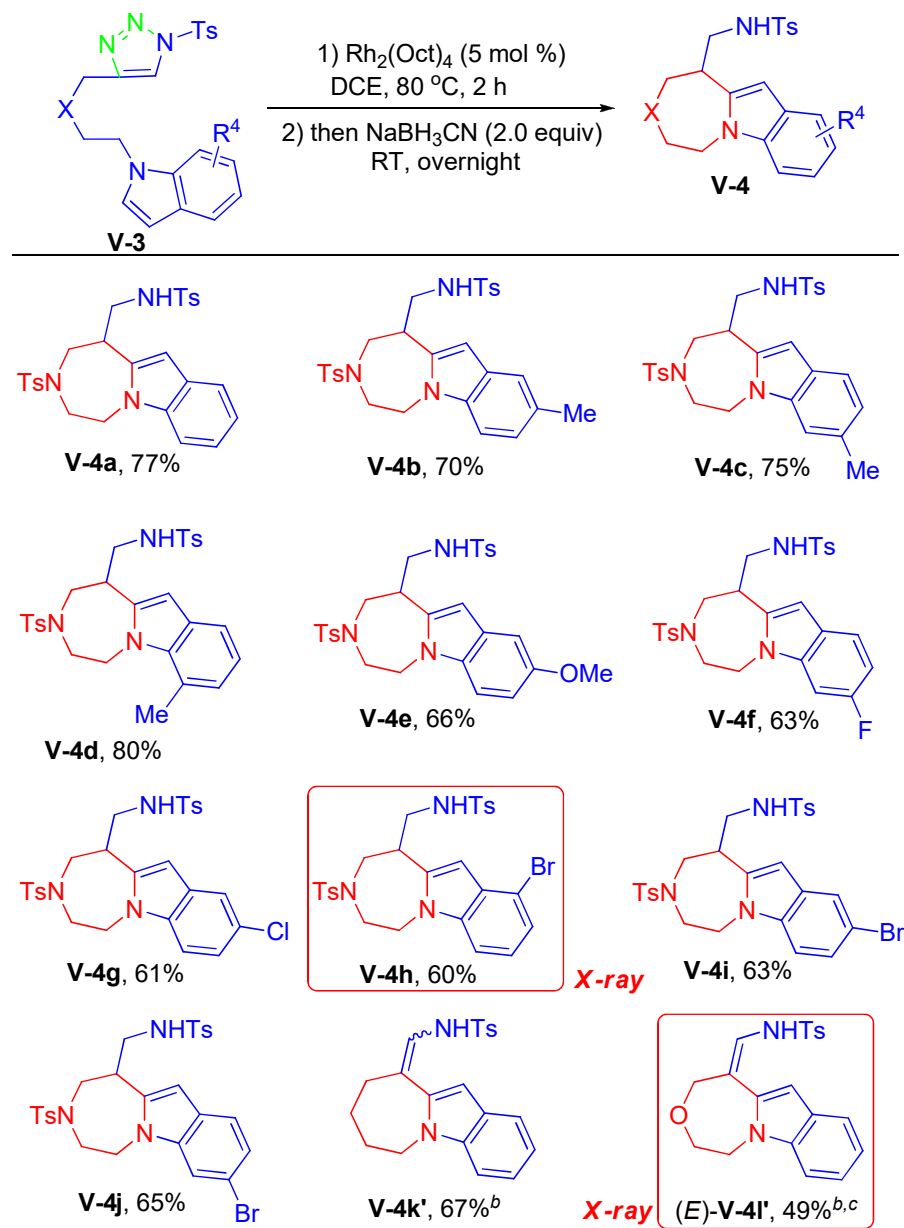
^a Reaction conditions: 0.1 mmol of **V-1a**; 5 mol% of cat.; 1.0 mL of dry solvent. ^b Isolated yields. ^c undetermined. DCE = 1,2-dichloroethane.

Substrate scope



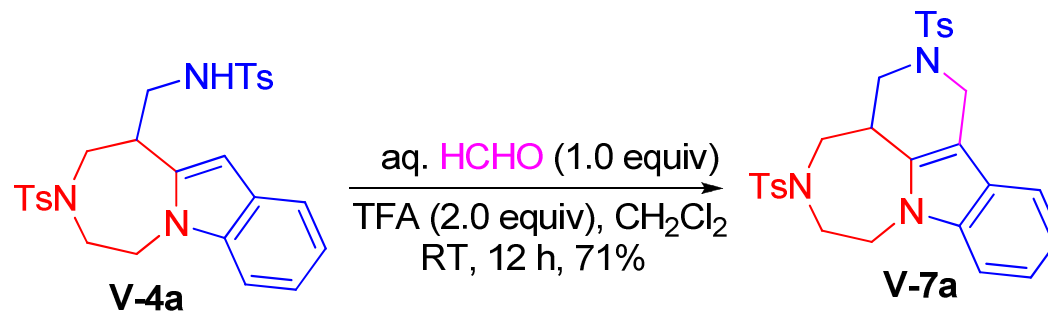
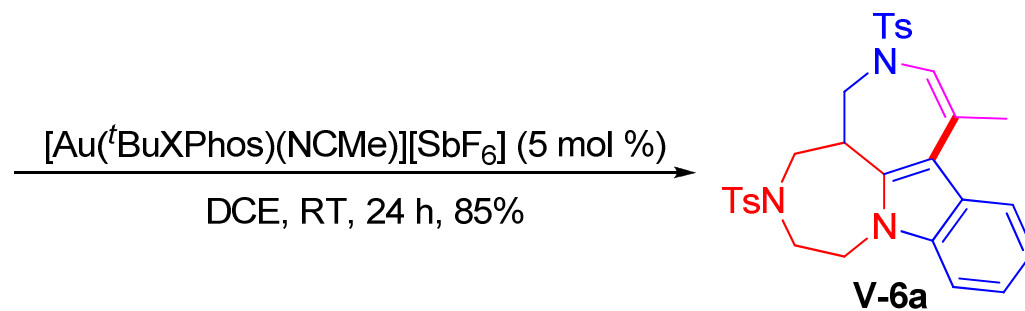
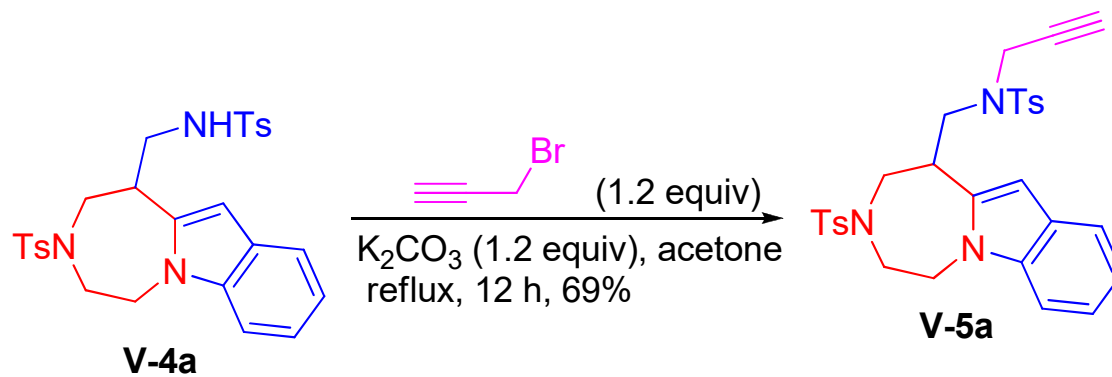
Reaction conditions: 0.1 mmol of **V-1**; 5 mol % of $\text{Rh}_2(\text{Oct})_4$; 1.0 mL anhydrous DCE. Isolated yields.

Substrate scope

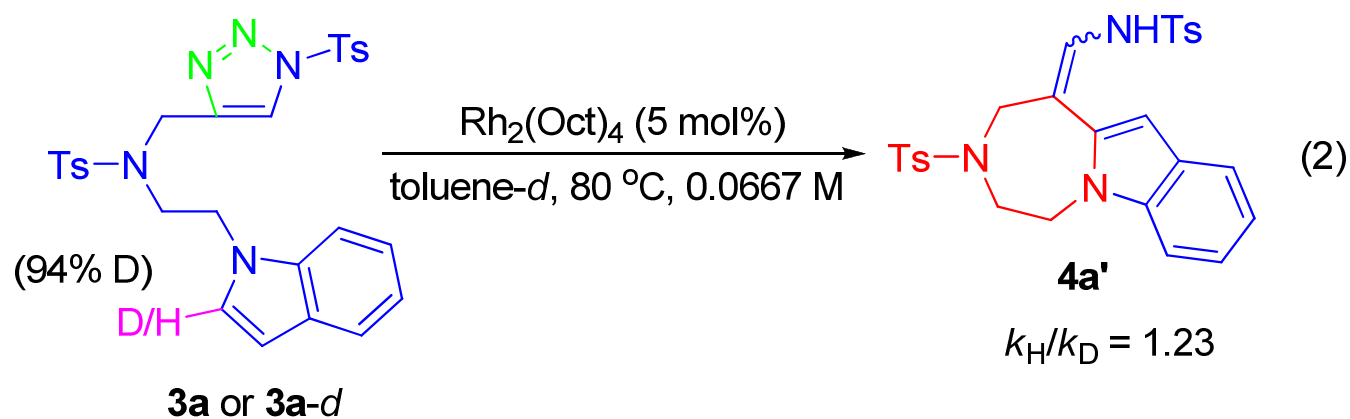
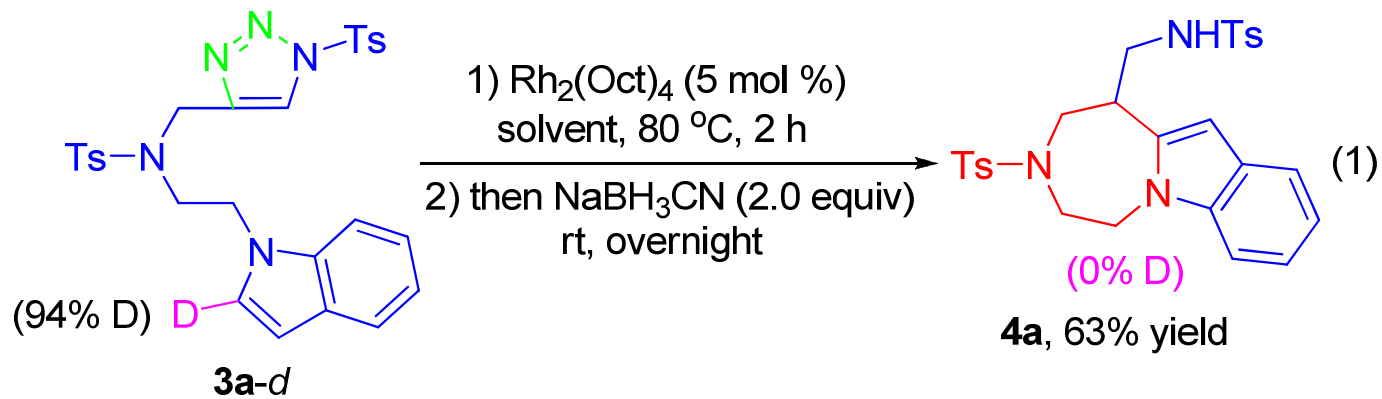


^a Reaction conditions: 0.1 mmol of **V-3**; 5 mol % of $\text{Rh}_2(\text{Oct})_4$; 1.0 mL anhydrous DCE. Isolated yields. ^b Substrates were performed only in the first step, and the two isomers were not reduced. ^c **V-4l''** was obtained in 43% yield.

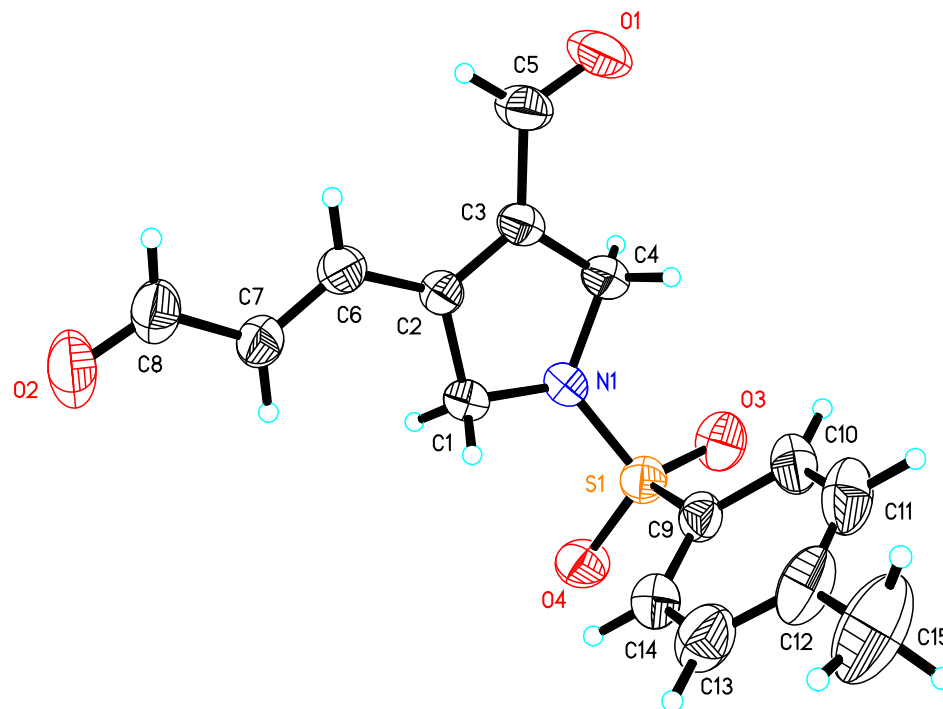
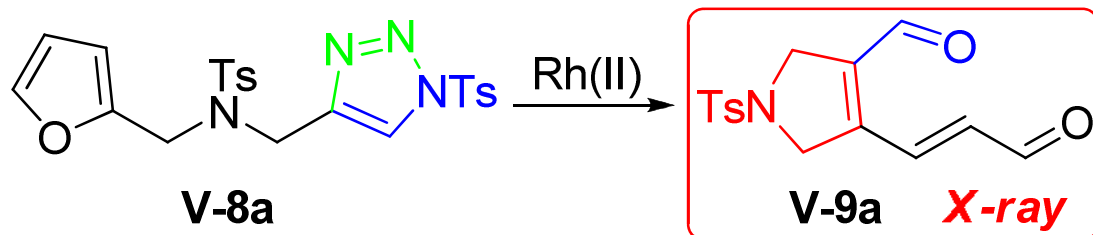
Further transformations of the products



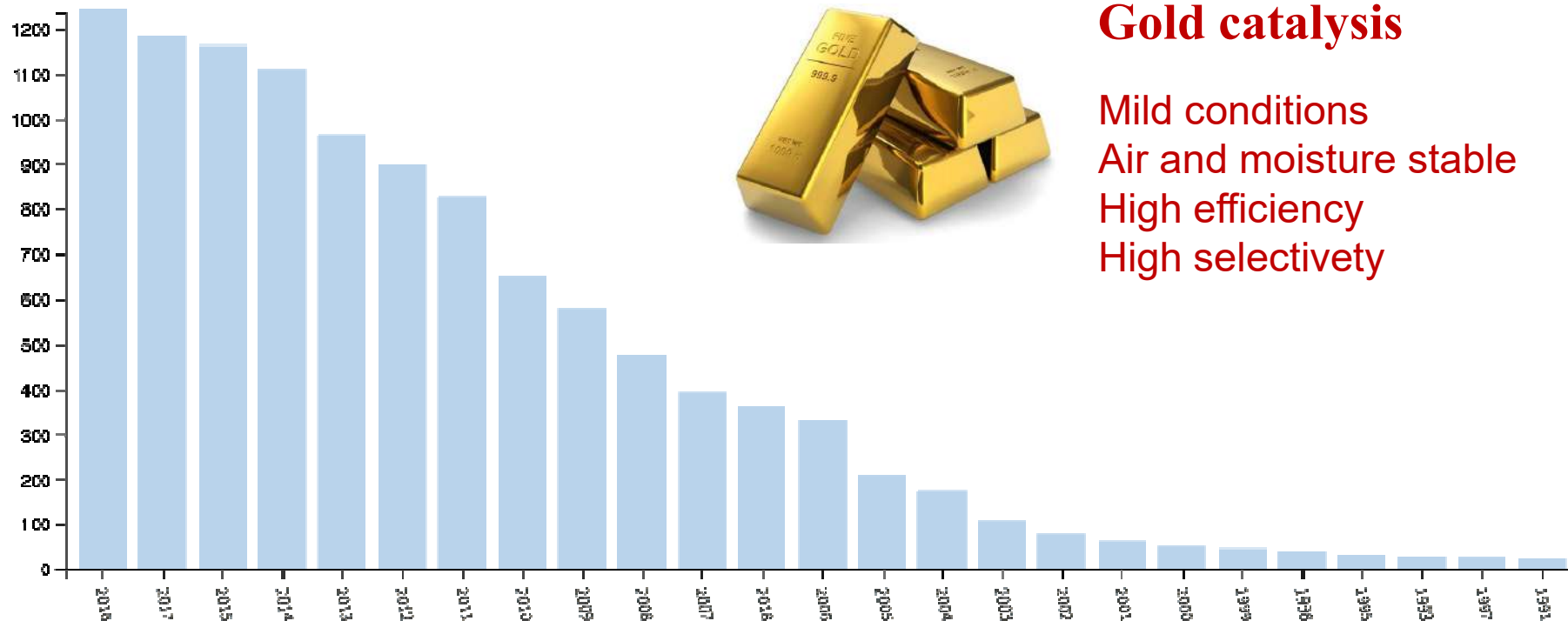
Kinetic isotope effect study



Substrate scope



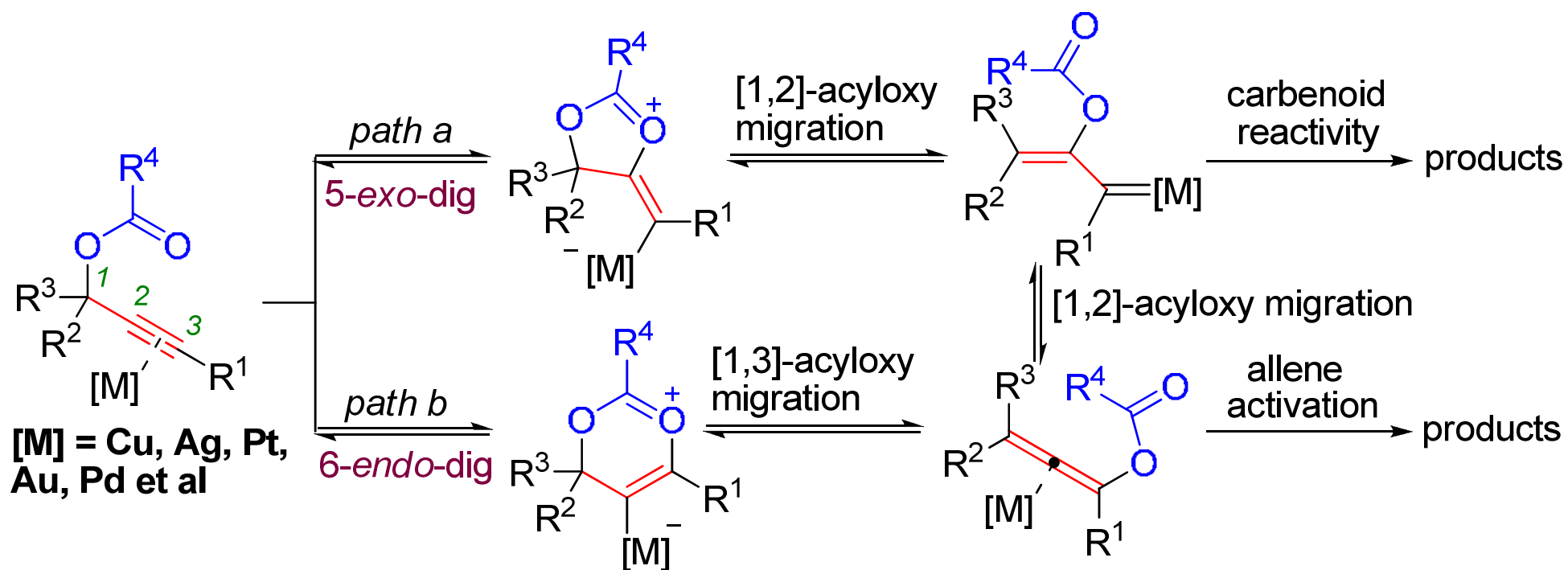
Gold catalysis



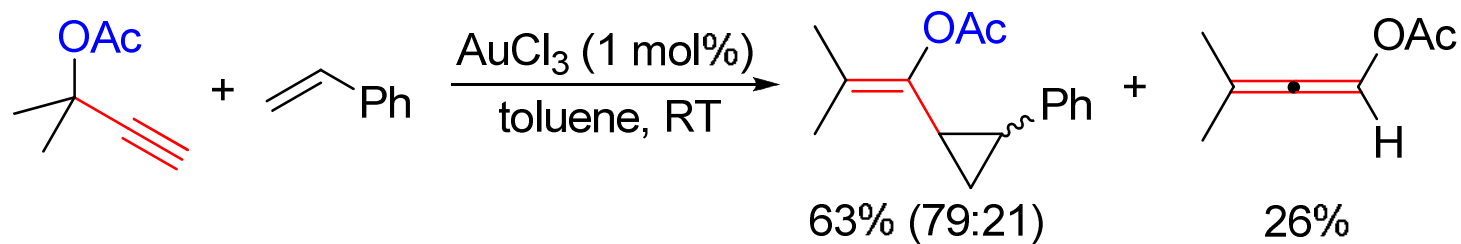
Gold catalysis

- Mild conditions
- Air and moisture stable
- High efficiency
- High selectivity

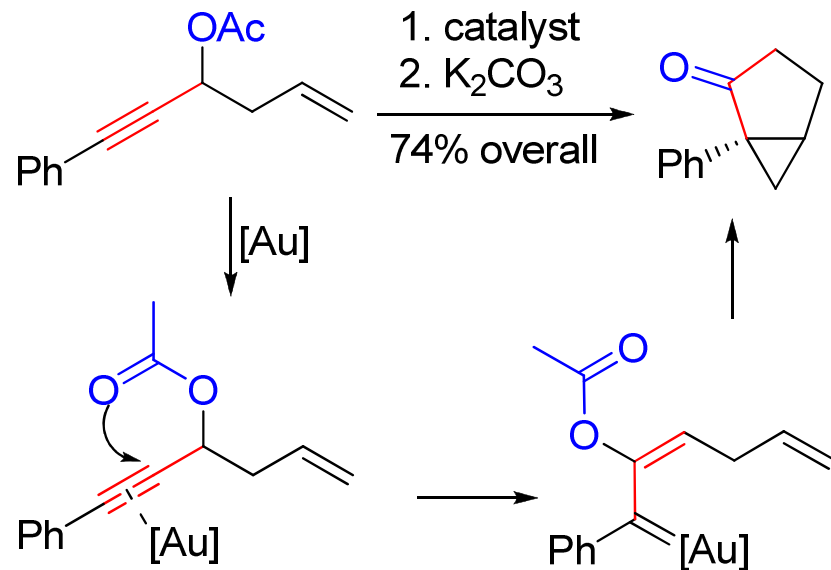
Propargylic Esters



[1,2]-Migration

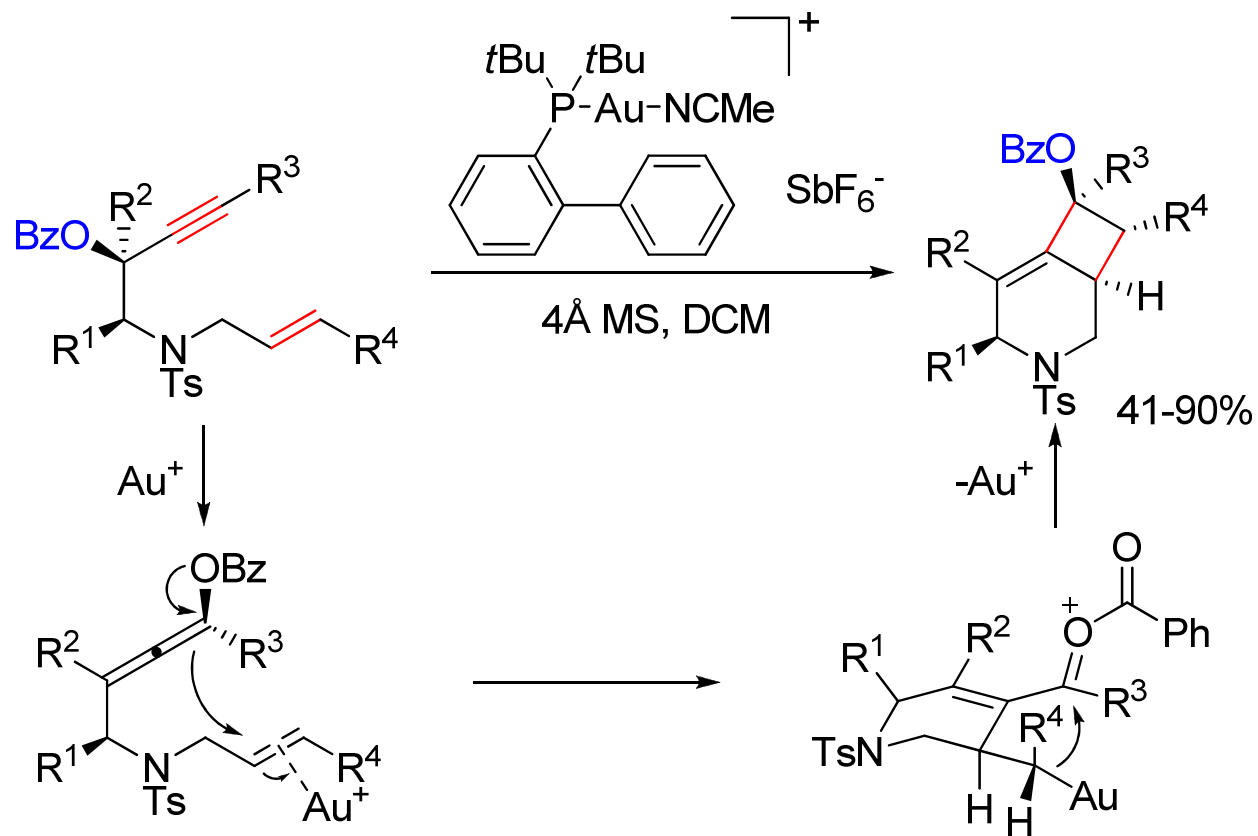


(a) Miki, K.; Ohe, K.; Uemura, S. *Tetrahedron Lett.* **2003**, *44*, 2019-2022. (b) Miki, K.; Ohe, K.; Uemura, S. *J. Org. Chem.* **2003**, *68*, 8505-8513.



Mamane, V.; Gress, T.; Krause, H.; Fürstner, J. *Am. Chem. Soc.* **2004**, *126*, 8654-8655.

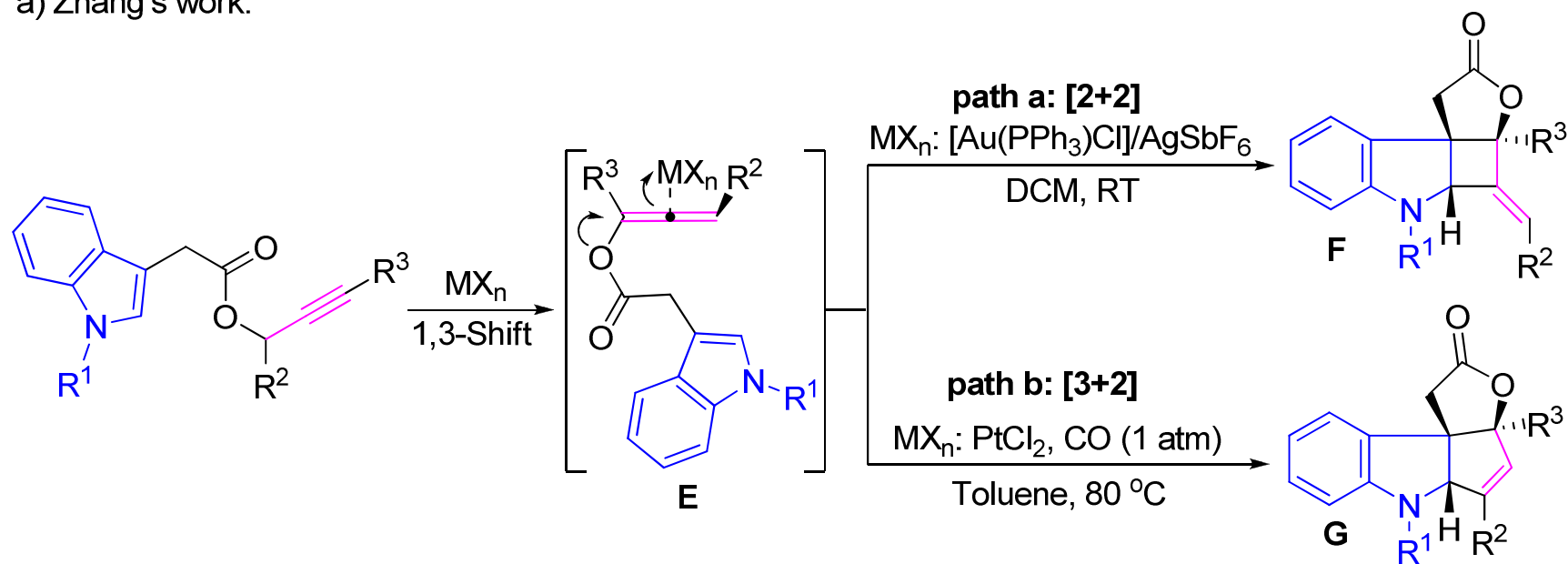
[1,3]-Migration



Rao, W.; Susanti, D.; Chan, P. W. H. *J. Am. Chem. Soc.* **2011**, *133*, 15248-15251.

Gold(I)-Catalyzed Highly Stereoselective Synthesis of Polycyclic Indolines

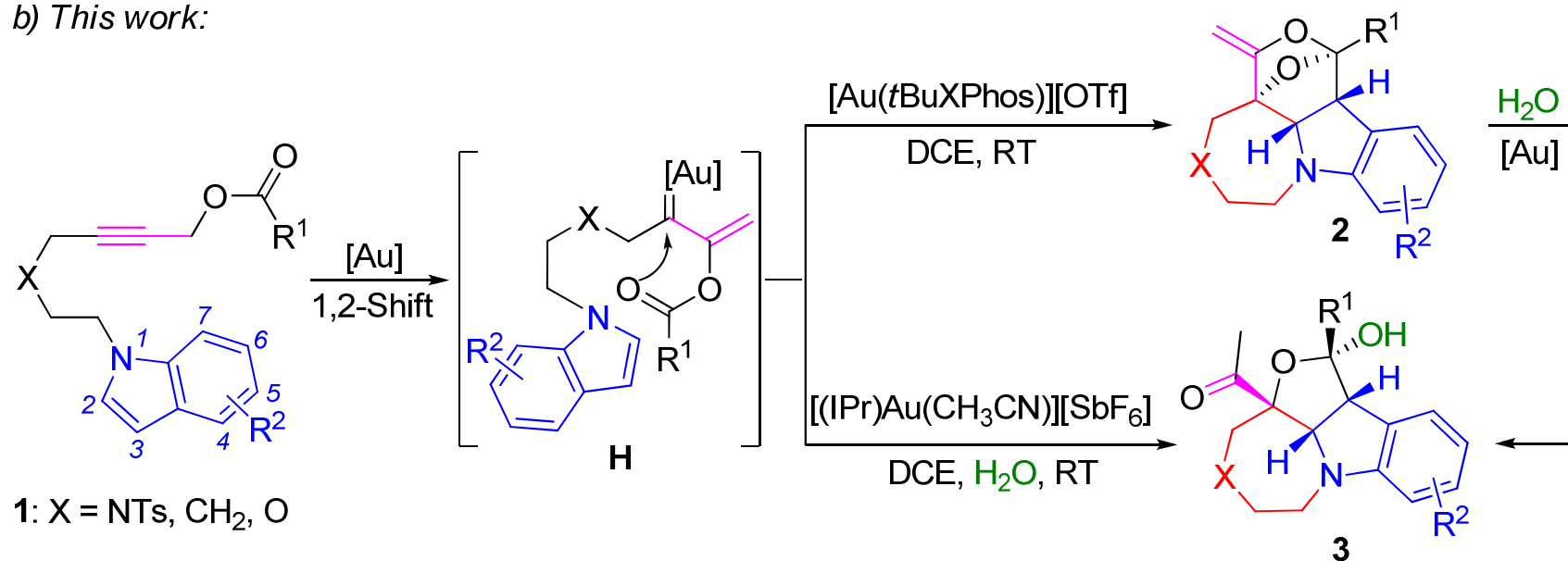
a) Zhang's work:



(a) Zhang, L. *J. Am. Chem. Soc.* **2005**, *127*, 16804-16805. (b) Zhang, G.; Catalano, V. J.; Zhang, L. *J. Am. Chem. Soc.* **2007**, *129*, 11358-11359.

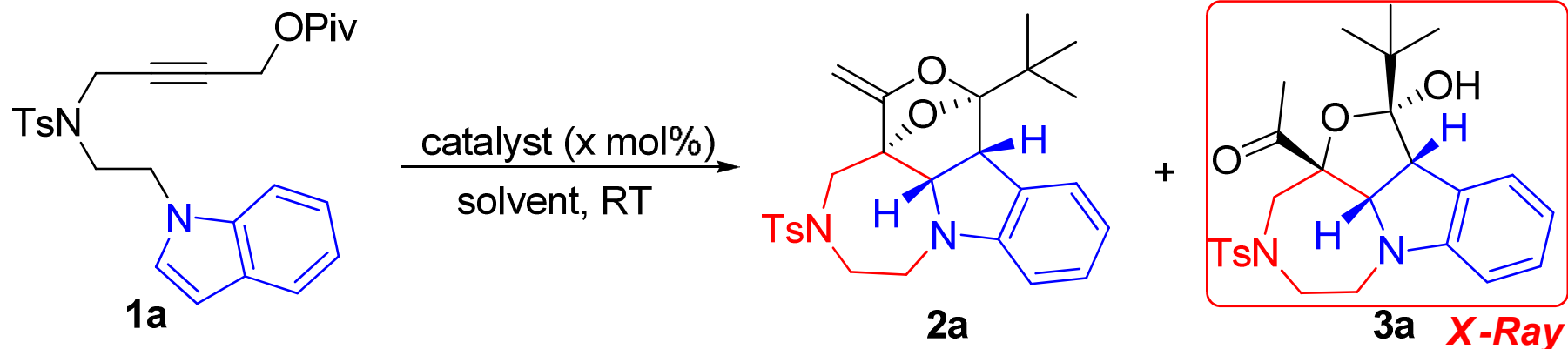
Gold(I)-Catalyzed Highly Stereoselective Synthesis of Polycyclic Indolines

b) This work:



Yang, J.-M.; Li, P.-H.; Wei, Y.; Tang, X.-Y.*; Shi, M.* *Chem. Commun.* **2016**, 52, 346-349.

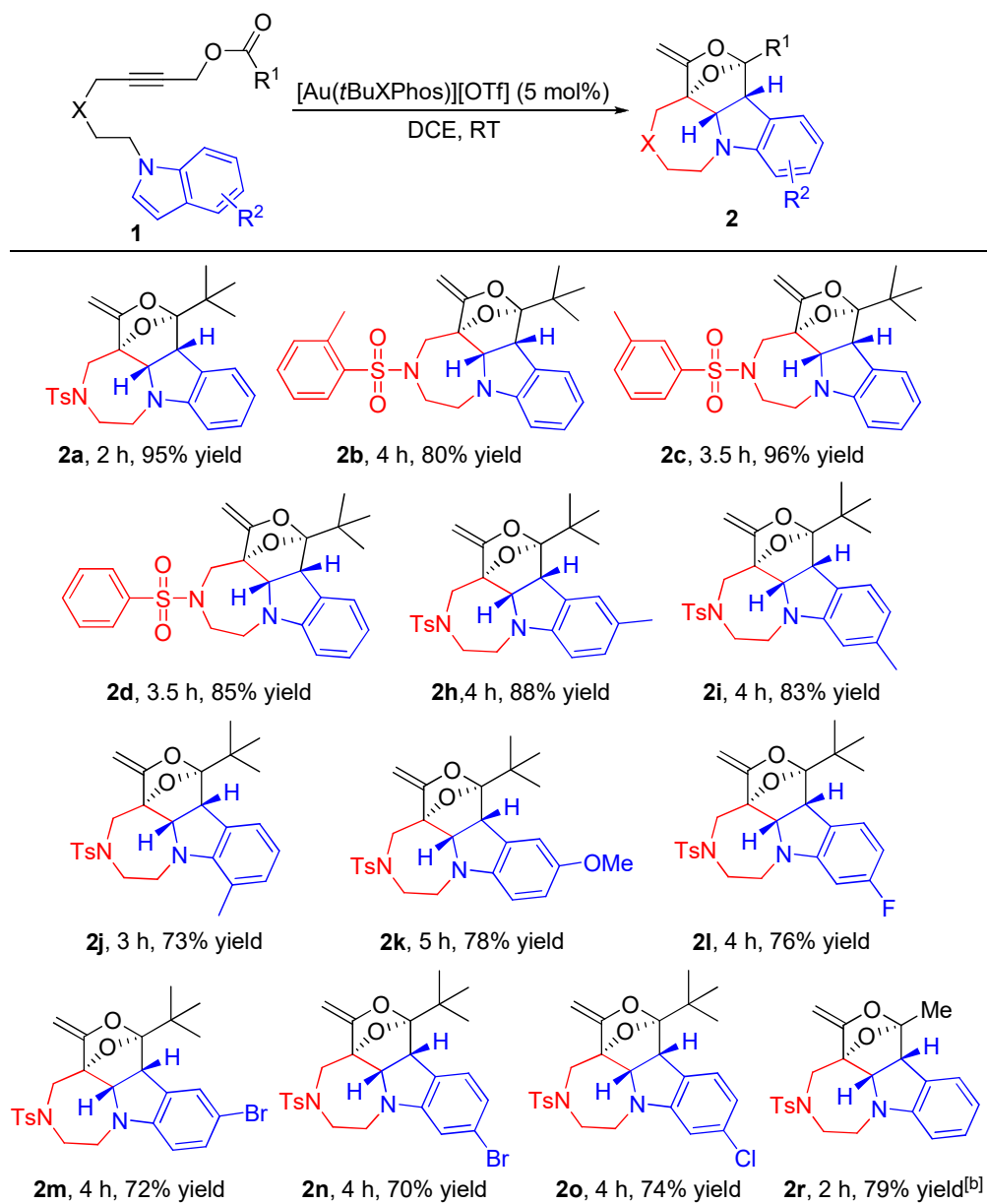
Optimization of the reaction conditions



entry ^[a]	catalyst (x mol%)	solvent	H ₂ O (y eq.)	time	yield (%) ^[b]	
					2a	3a
1	[Au(<i>t</i>BuXPhos)][OTf] (5)	DCE	-	2 h	95	0
2	[Au(Me ₄ tBuXPhos)(CH ₃ CN)][SbF ₆] (5)	DCE	-	6 h	0	0
3	[(IPr)Au(CH ₃ CN)][SbF ₆] (5)	DCE	-	2 h	0	70
4	[Au(<i>n</i> BuPAd ₂)(CH ₃ CN)][SbF ₆] (5)	DCE	-	4 h	0	89
5	[(ArO) ₃ PAu][NTf ₂] (5)	DCE	-	4 h	0	70
6	[(IPr)Au(CH ₃ CN)][SbF ₆] (5)	DCE	1.0	2 h	0	85
7	[Ph ₃ PAuCl] (5)/AgNTf ₂ (5)	DCE	1.0	1 h	0	74
8	[(IPr)Au(CH₃CN)][SbF₆] (2.5)	DCE	1.0	5 h	0	86

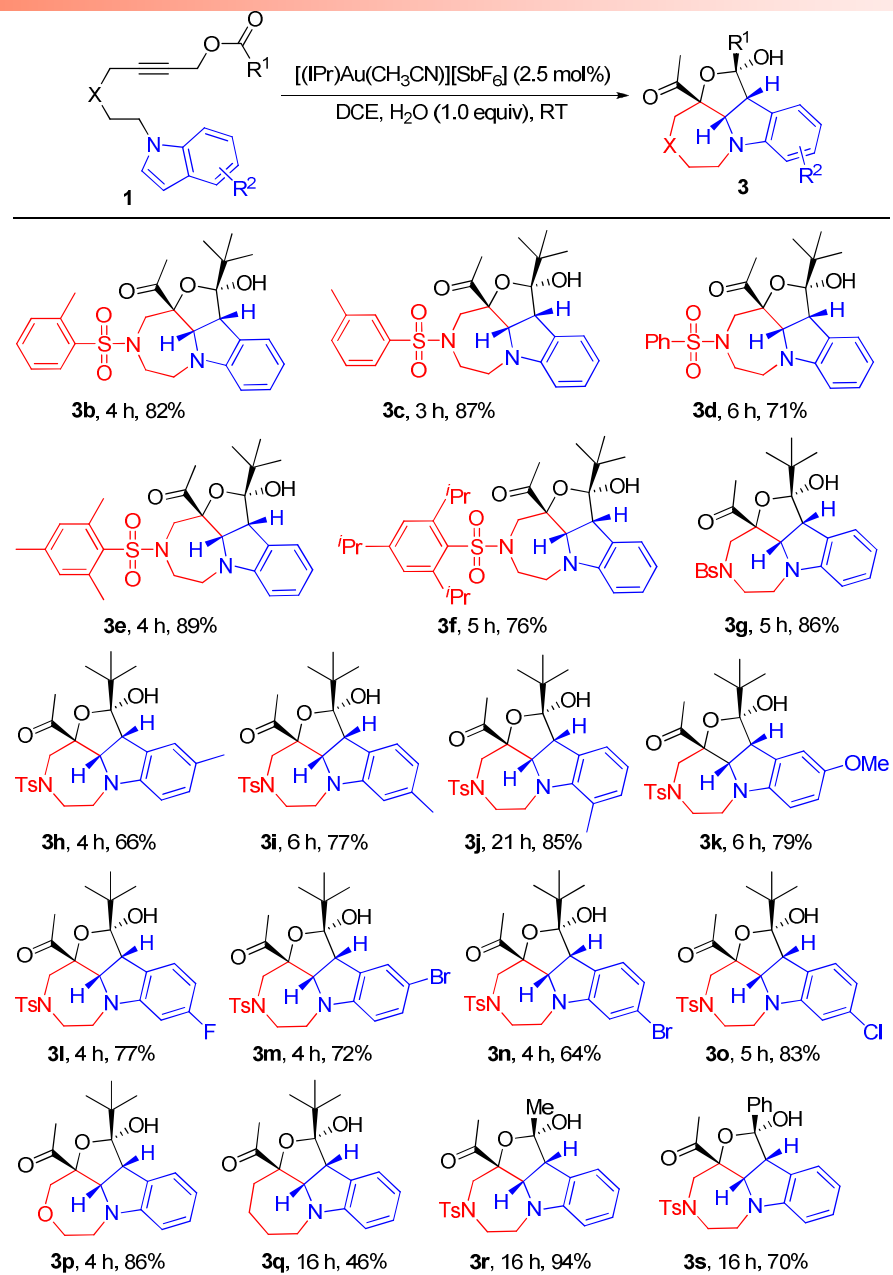
[a] All reactions were carried out using **1a** (0.1 mmol) in the presence of catalyst (x mol%) in DCE (1.0 mL) unless otherwise specified. [b] Isolated yields. Ar = 2,4-di-tert-butylphenyl. DCE = 1,2-dichloroethane.

Substrate scope

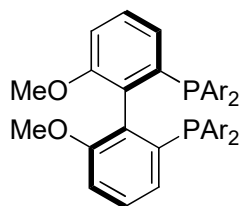


[a] Reaction conditions: **1** (0.1 mmol); $[\text{Au}(\text{tBuXPhos})][\text{OTf}]$ (5 mol%); anhydrous DCE (1.0 mL). Yields are those of the isolated yields. [b] 2.5 mol% of $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$ was used as the catalyst.

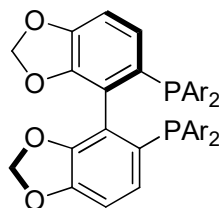
Substrate scope



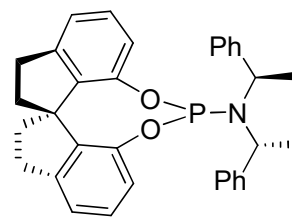
Ligands for asymmetric synthesis



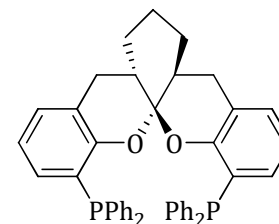
Ar = 3,5-(tBu)₂-4-MeOC₆H₂;
(*R*)-MeO-DTBM-BIPHEP (**L1**)



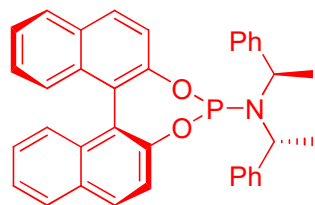
Ar = 3,5-(Me)₂C₆H₃;
(*R*)-DM-SEGPHOS (**L2**)



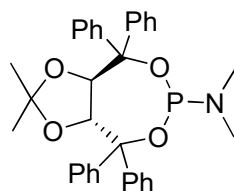
L3



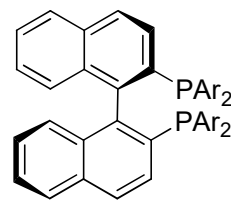
L4



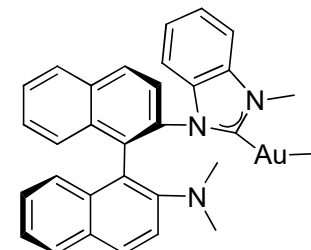
L5



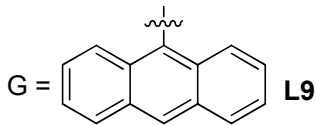
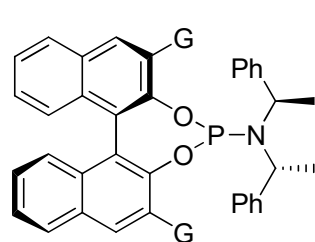
L6



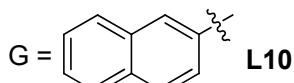
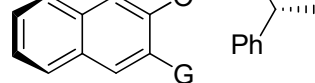
Ar = 3,5-(Me)₂C₆H₃;
(*R*)-Xylyl-BINAP (**L7**)



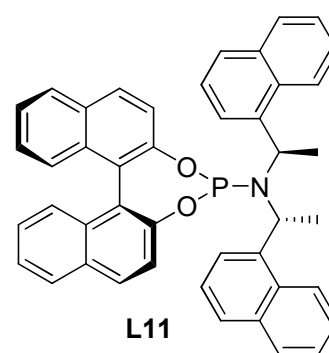
(*aR*)-**8**



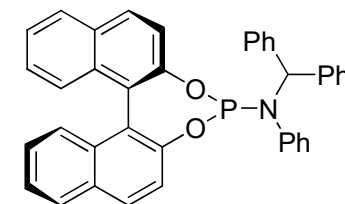
L9



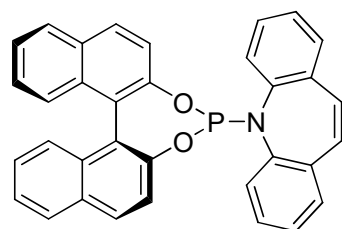
L10



L11

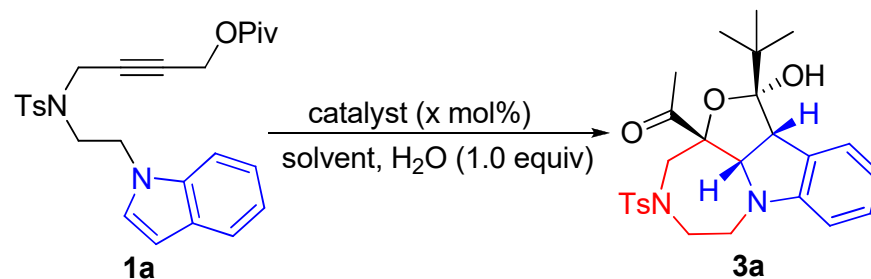


L12



L13

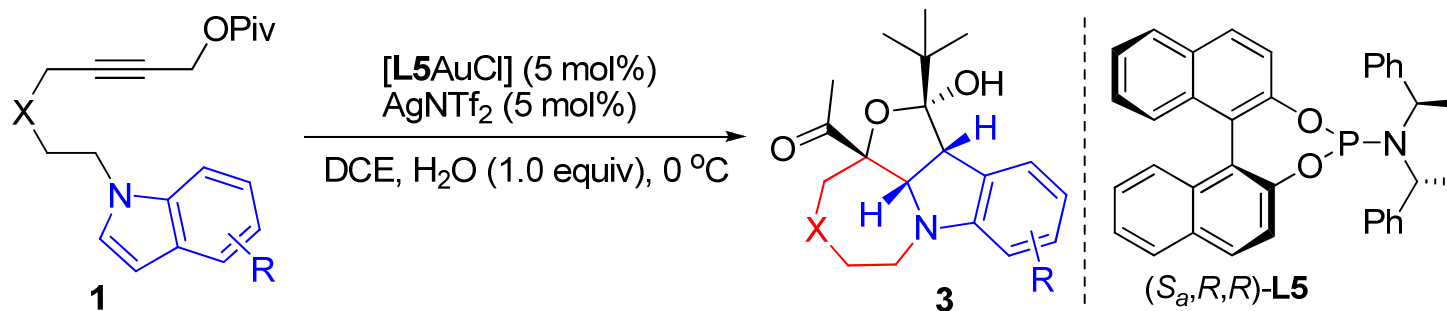
Optimization of the reaction conditions



entry ^[a]	catalyst (x mol%)	solvent	T (°C)	time	yield (%) ^[b]	ee (%) ^[c]
1	[L1AuCl] (5)/AgSbF ₆ (5)	DCE	RT	1 h	74	0
2	[L2Au ₂ Cl ₂] (5)/AgOPNB (5)	DCE	RT	2 d	trace	_[d]
3	[L2Au ₂ Cl ₂] (5)/AgSbF ₆ (5)	DCE	RT	2 h	72	0
4	[L3Au(CH ₃ CN)][SbF ₆] (5)	DCE	RT	3 h	85	40
5	[L4Au ₂ (CH ₃ CN) ₂][(SbF ₆) ₂] (5)	DCE	RT	3 d	67	15
6	[L5Au(CH ₃ CN)][SbF ₆] (5)	DCE	RT	1 h	86	71
7	[L5AuCl] (5)/AgNTf ₂ (5)	DCE	RT	3.5 h	79	71
8	[L5AuCl] (5)/AgNTf ₂ (5)	Toluene	RT	4 h	23	47
9	[L5Au(CH ₃ CN)][SbF ₆] (5)	DCM	RT	1 h	87	67
10	[L5Au(CH ₃ CN)][SbF ₆] (5)	CHCl ₃	RT	5 h	85	50
11	[L5AuCl] (5)/AgBF ₄ (5)	DCE	RT	45 min	65	31
12	[L5AuCl] (5)/AgSbF ₆ (5)	DCE	RT	4 h	77	14
13	[L5AuCl] (5)/AgOTf (5)	DCE	RT	4 h	trace	_[d]
14	[L5AuCl] (5)/AgOONB (5)	DCE	RT	4 d	trace	_[d]
15	[L6Au(CH ₃ CN)][SbF ₆] (5)	DCE	RT	6 h	55	-7
16	[L7Au ₂ Cl ₂] (5)/AgSbF ₆ (10)	DCE	RT	5 h	67	2
17	(aR)- 8 (5)/AgSbF ₆ (5)	DCE	RT	30 min	79	0
18	[L9AuCl] (5)/AgNTf ₂ (5)	DCE	RT	4 h	78	-17
19	[L10AuCl] (5)/AgNTf ₂ (5)	DCE	RT	22 h	77	43
20	[L11AuCl] (5)/AgNTf ₂ (5)	DCE	RT	2 h	84	52
21	[L12AuCl] (5)/AgNTf ₂ (5)	DCE	RT	18 h	69	-46
22	[L13AuCl] (5)/AgNTf ₂ (5)	DCE	RT	1.5 h	72	27
23	[L5AuCl] (5)/AgNTf₂ (5)	DCE	0	16 h	72	77

[a] All reactions were carried out using **1a** (0.1 mmol) in the presence of catalyst (x mol%) in various solvents (1.0 mL) unless otherwise specified. [b] Yield of isolated product. [c] Determined by HPLC on a chiral stationary phase. [d] Not determined. OPNB = *p*-nitrobenzoate, OONB = *o*-nitrobenzoate

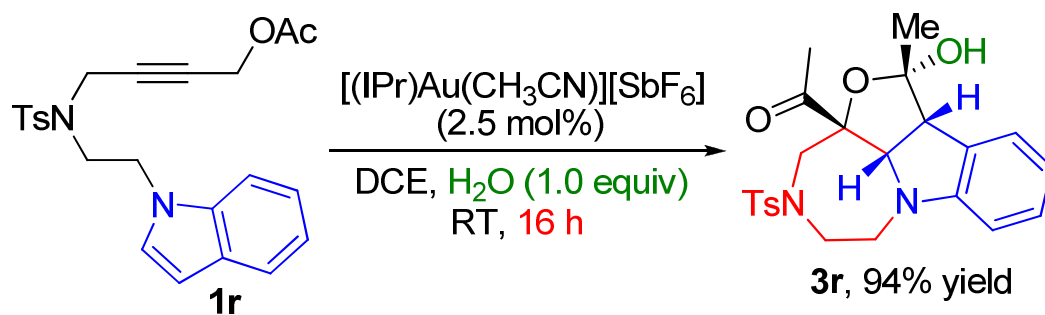
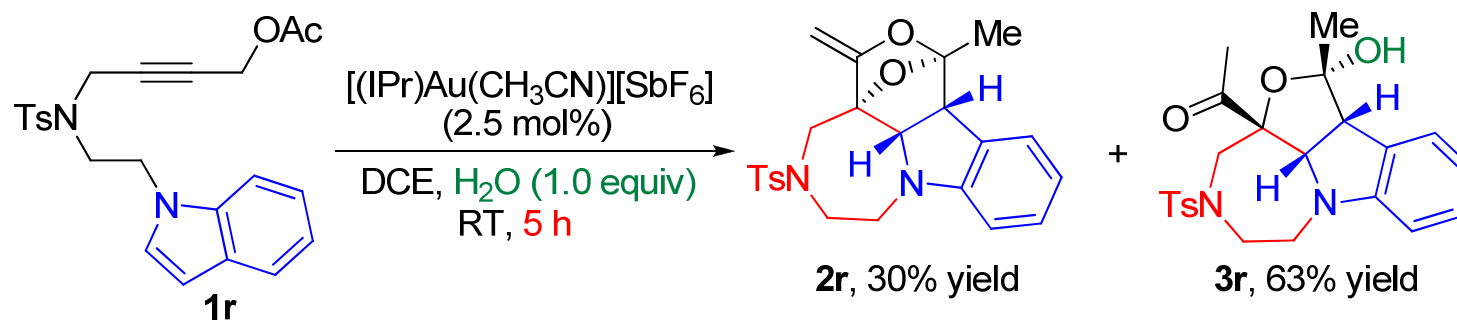
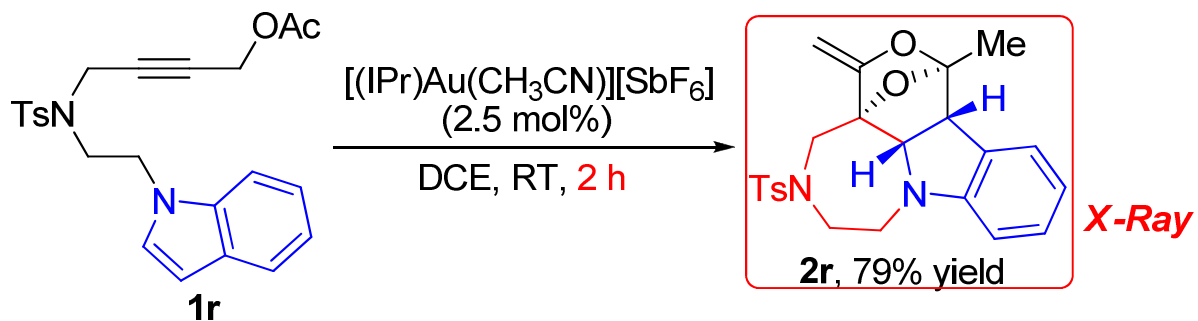
Substrate scope



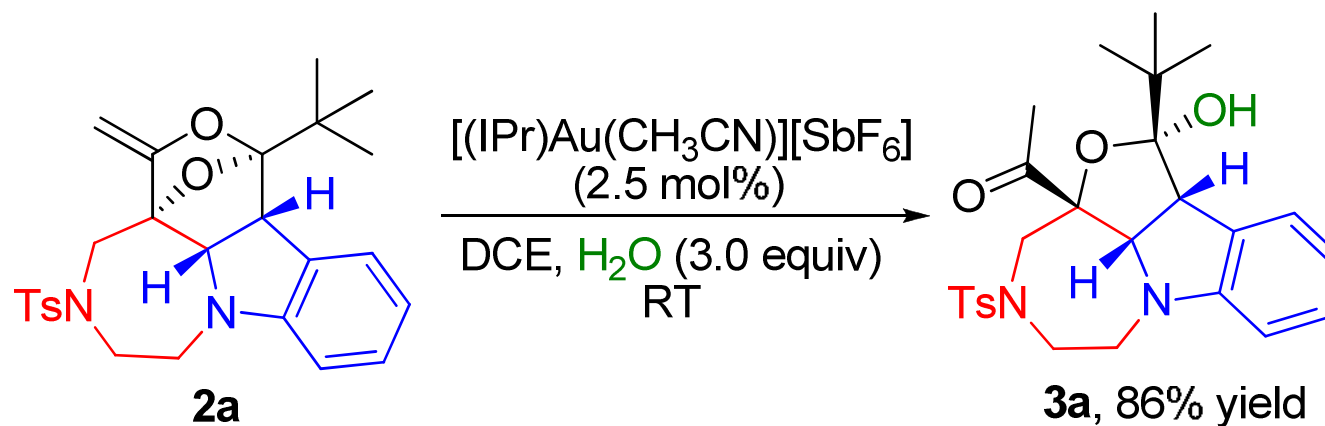
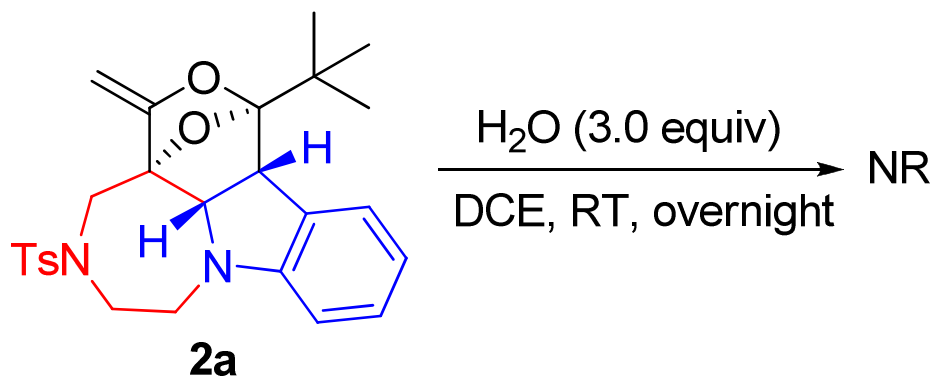
entry ^[a]	1	X	R	time [days]	3	yield [%] ^[b]	ee [%] ^[c]
1	1b	2-MeC ₆ H ₄ SO ₂ N	H	3	3b	30	74
2	1c	3-MeC ₆ H ₄ SO ₂ N	H	2	3c	60	80
3	1d	PhSO ₂ N	H	2	3d	43	90
4	1e	MesSO ₂ N	H	2	3e	51	83
5	1f	2,4,6- <i>i</i> Pr ₃ C ₆ H ₂ SO ₂ N	H	2	3f	40	91
6	1h	TsN	5-Me	2	3h	50	74
7	1i	TsN	6-Me	0.5	3i	46	71
8	1j	TsN	7-Me	2	3j	88	81
9	1k	TsN	5-OMe	2	3k	62	82
10	1l	TsN	6-F	0.5	3l	72	82
11 ^[d]	1m	TsN	5-Br	2	3m	66	62
12	1n	TsN	6-Br	1	3n	85	82
13	1o	TsN	6-Cl	3		_ ^[e]	_ ^[e]
14	1p	O	H	3	3p	30	72
15	1q	CH ₂	H	3	3q	25	60

[a] Reaction conditions: **1** (0.1 mmol), $[L5AuCl]$ (5 mol%), $AgNTf_2$ (5 mol%), H_2O (1.0 equiv), anhydrous DCE (1.0 mL). [b] Yields are those of the isolated yields. [c] Determined by HPLC on a chiral stationary phase. [d] Reaction performed at room temperature. [e] Complex mixtures, not determined. DCE=1,2-dichloroethane.

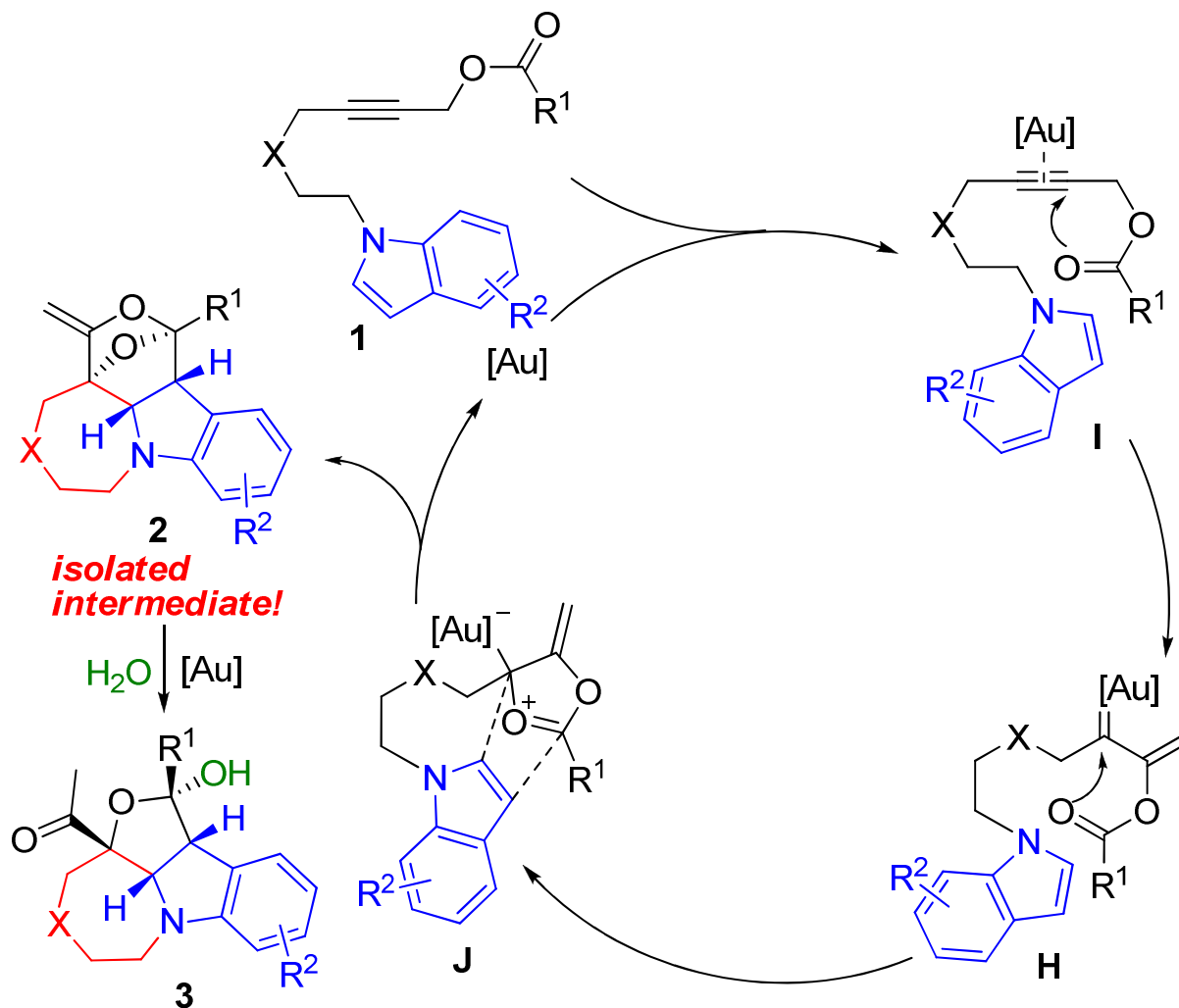
The control experiments



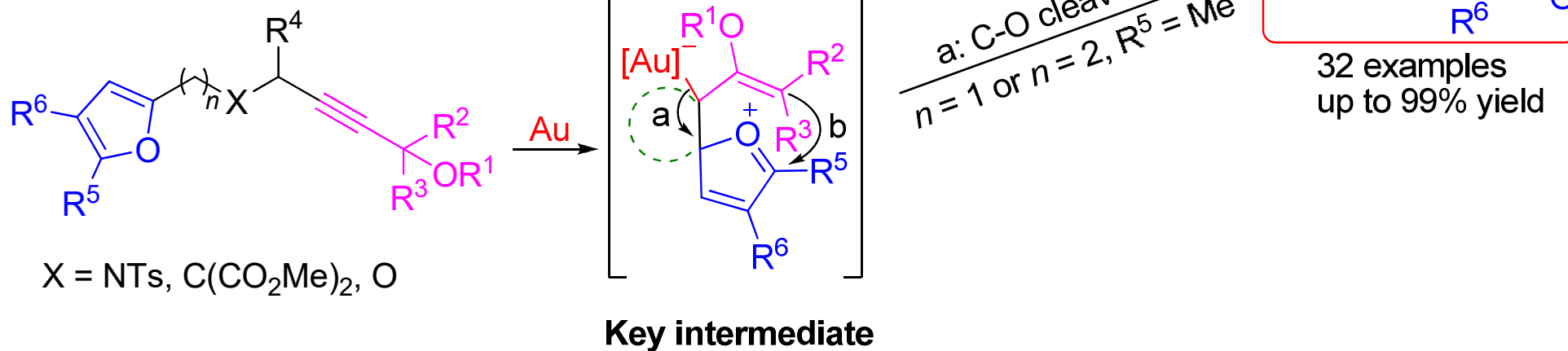
The control experiments



A plausible mechanism

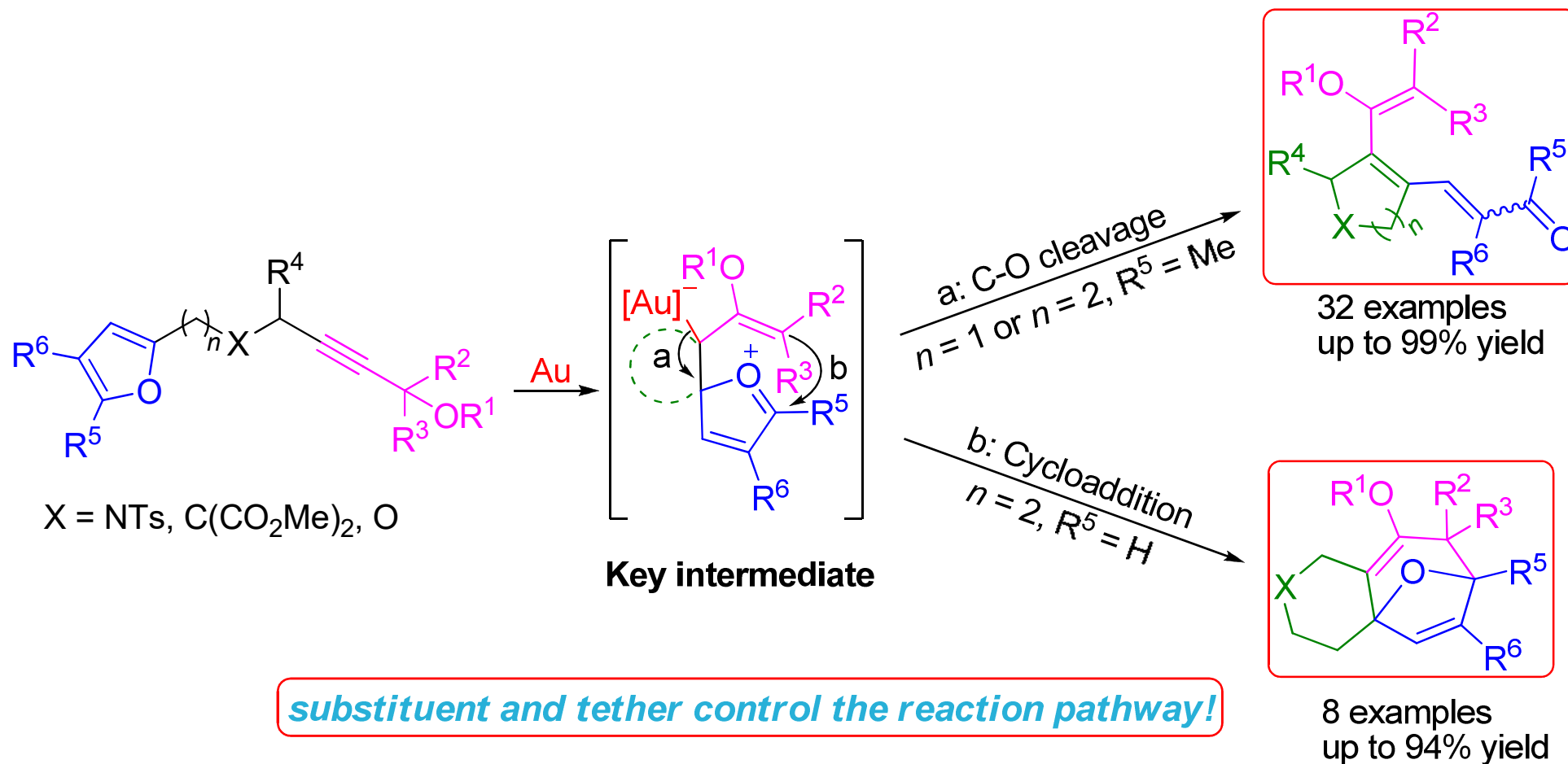


Gold(I)-catalyzed intramolecular cycloisomerization of α -yne furans

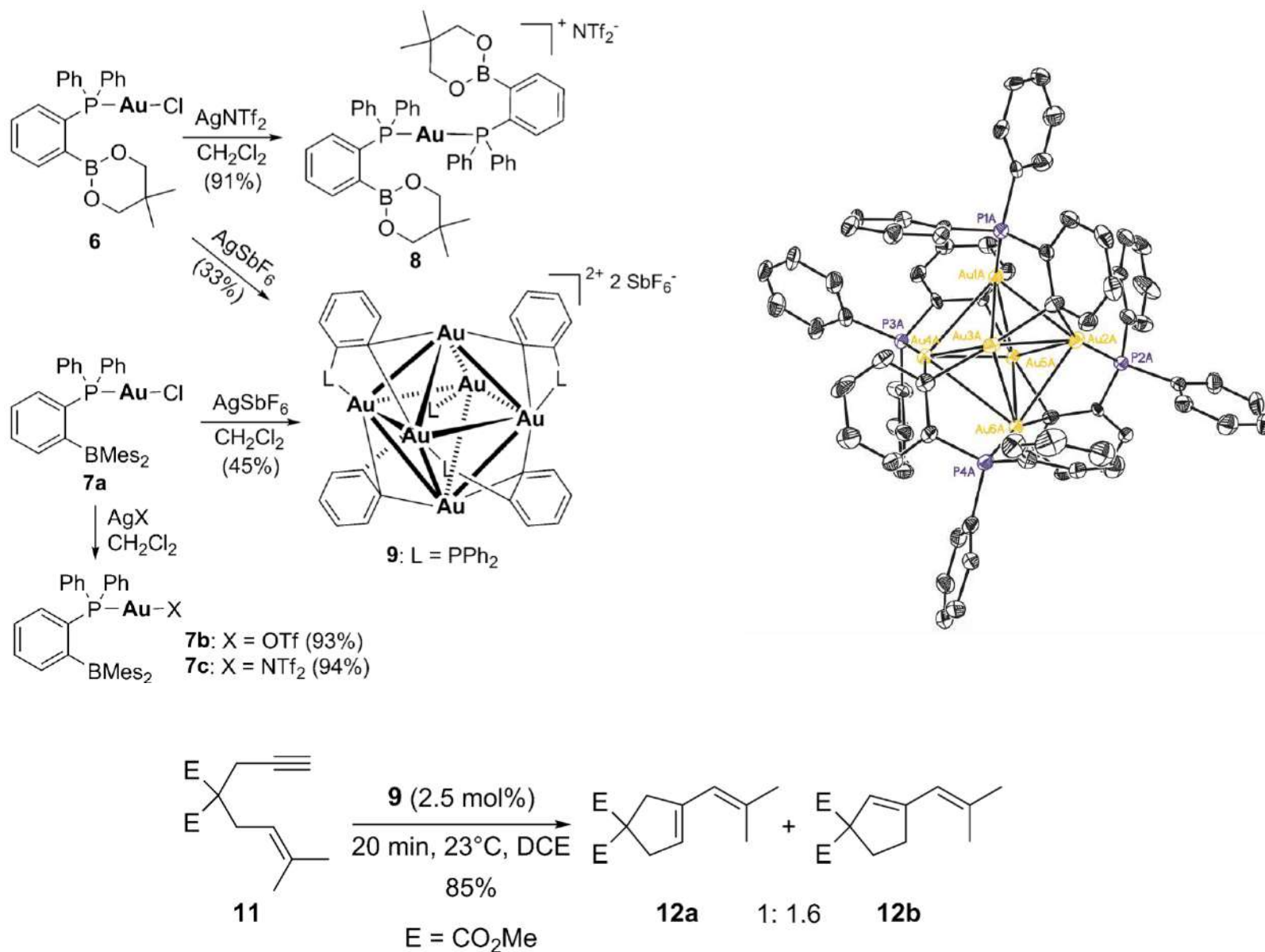


substituent and tether control the reaction pathway!

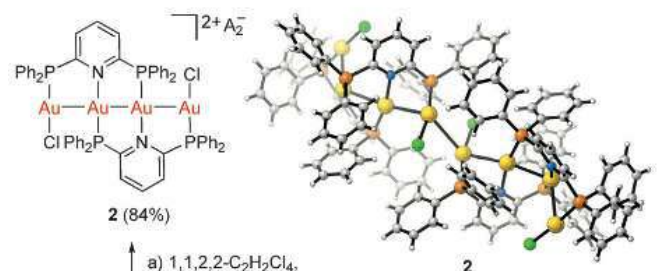
Gold(I)-catalyzed intramolecular cycloisomerization of α -yne furans



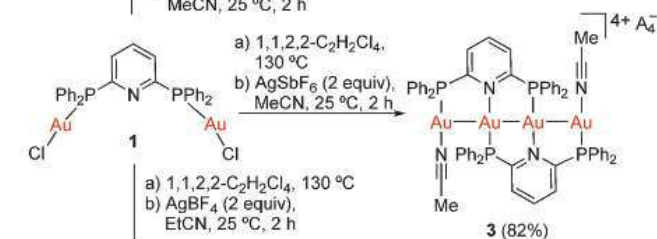
Polynuclear Gold Clusters Synthesis



Polynuclear Gold Clusters Synthesis

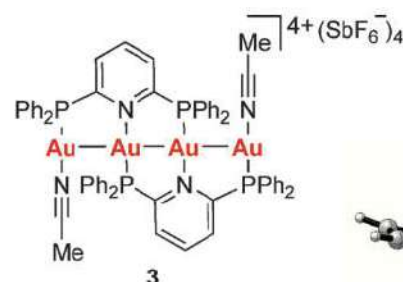
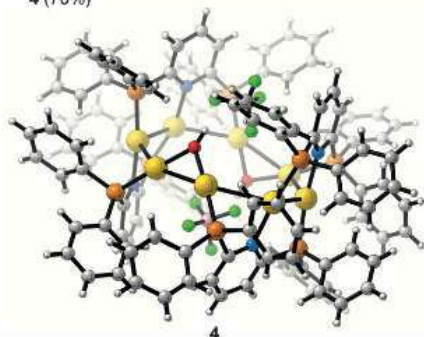
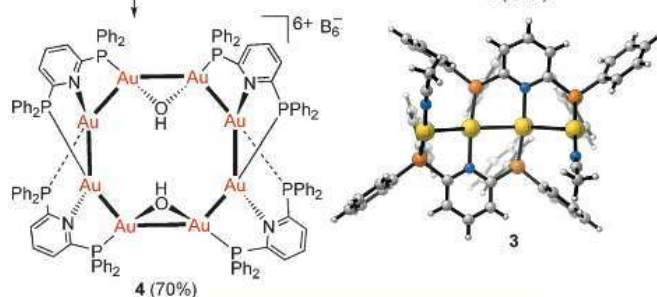


a) 1,1,2,2-C₂H₂Cl₄,
130 °C
b) AgSbF₆ (1 equiv),
MeCN, 25 °C, 2 h

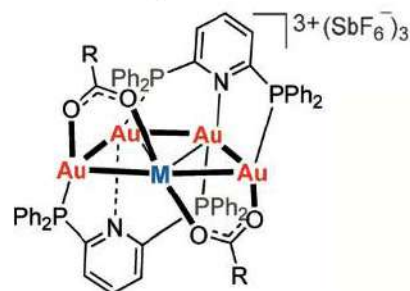
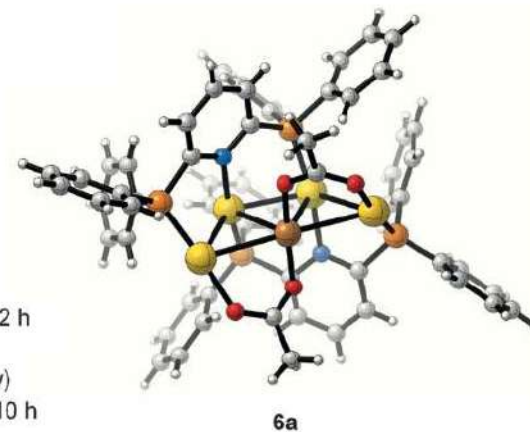


a) 1,1,2,2-C₂H₂Cl₄,
130 °C
b) AgSbF₆ (2 equiv),
MeCN, 25 °C, 2 h

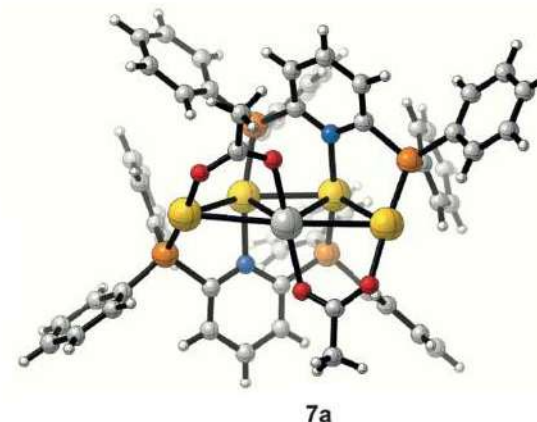
a) 1,1,2,2-C₂H₂Cl₄, 130 °C
b) AgBF₄ (2 equiv),
EtCN, 25 °C, 2 h



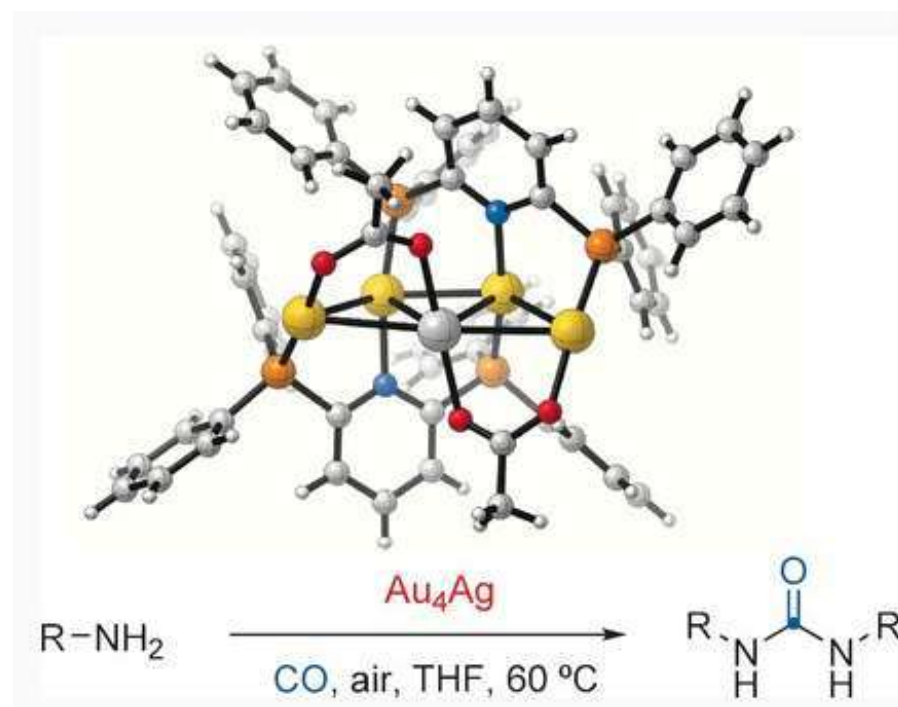
Method A:
M₂O (2 equiv)
MeCN, 25 °C, 12 h
Method B:
MO₂CR (2 equiv)
CH₂Cl₂, 25 °C, 10 h



6a: M = Cu, R = Me
(68%^A, 70%^B)
6b: M = Cu, R = Et (75%^A)
7a: M = Ag, R = Me
(75%^A, 83%^B)
7b: M = Ag, R = Et (80%^A)
7c: M = Ag, R = Ph (85%^B)

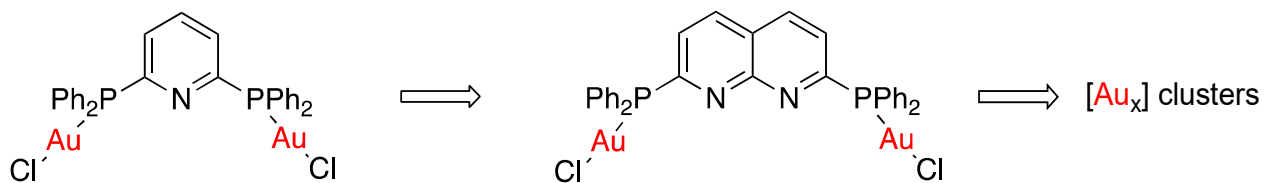


Polynuclear Gold Clusters Synthesis

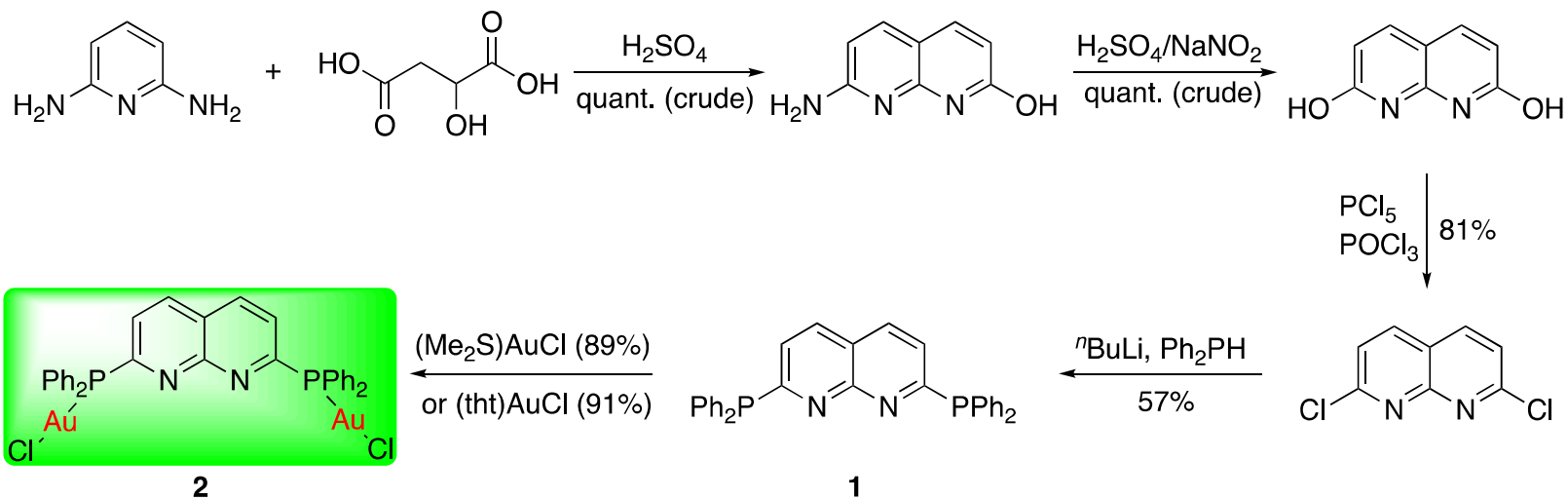
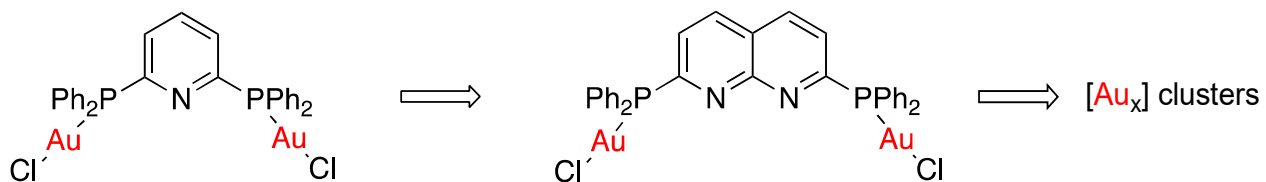


Smirnova, E. S.; Molina, J. M. M.; Johnson, A.; Bandeira, N. A. G.; Bo, C., Echavarren, A. M. *Angew. Chem. Int. Ed.* **2016**, *55*, 7487.

Ligand synthesis

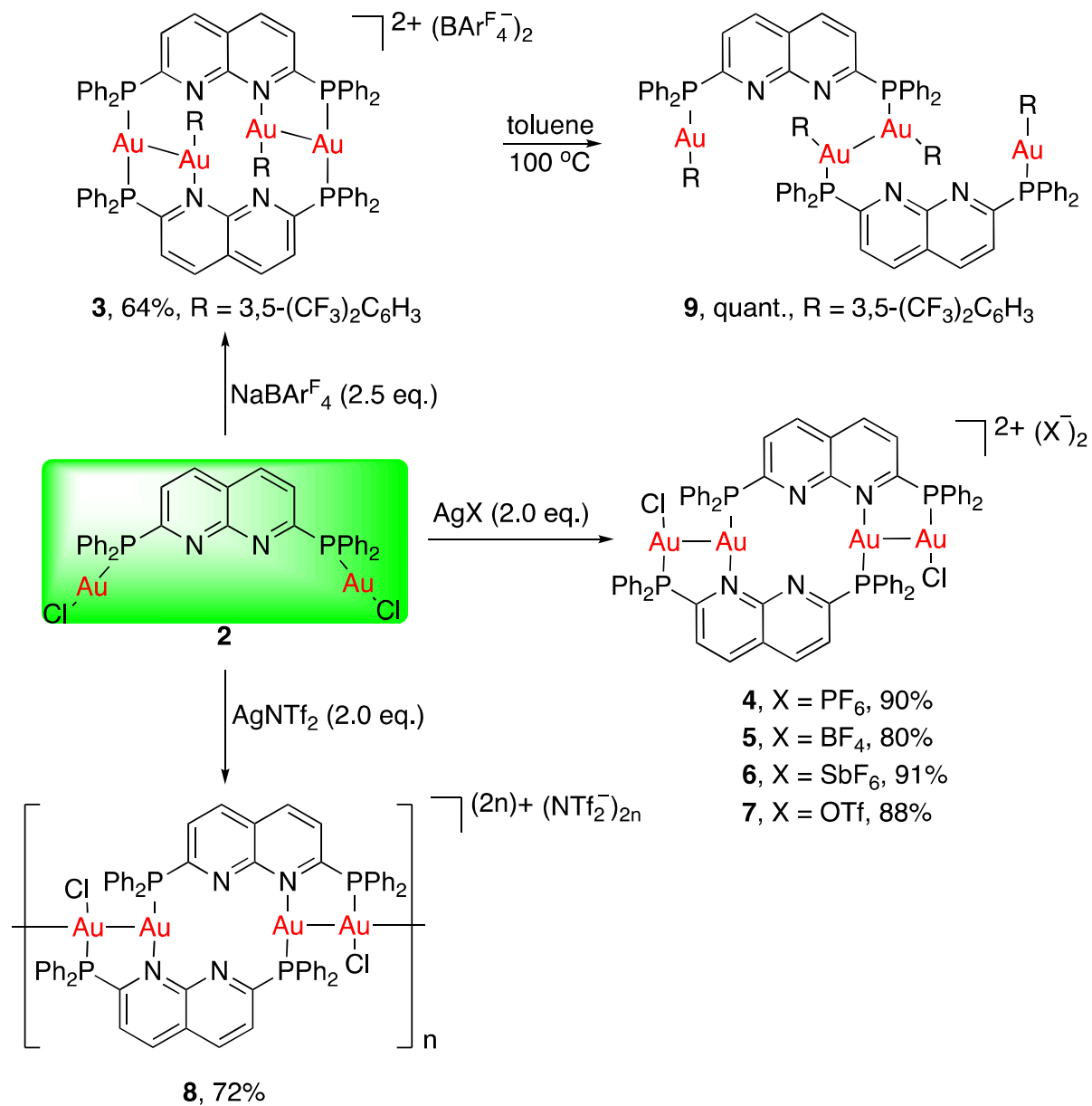


Ligand synthesis

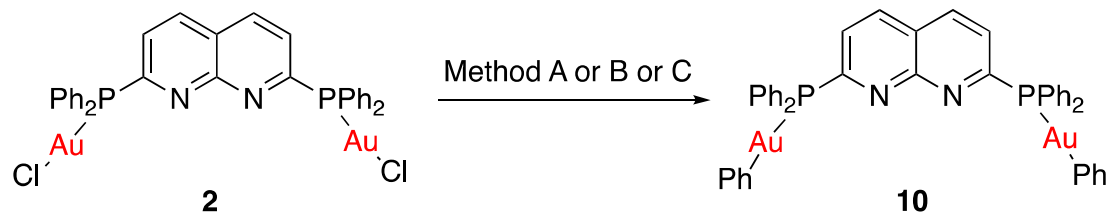


(a) Ziessel, R. *Tetrahedron Lett.* **1989**, *30*, 463-466. (b) Catalano, V. J.; Kar, H. M.; Bennett, B. L. *Inorg. Chem.* **2000**, *39*, 121-127.

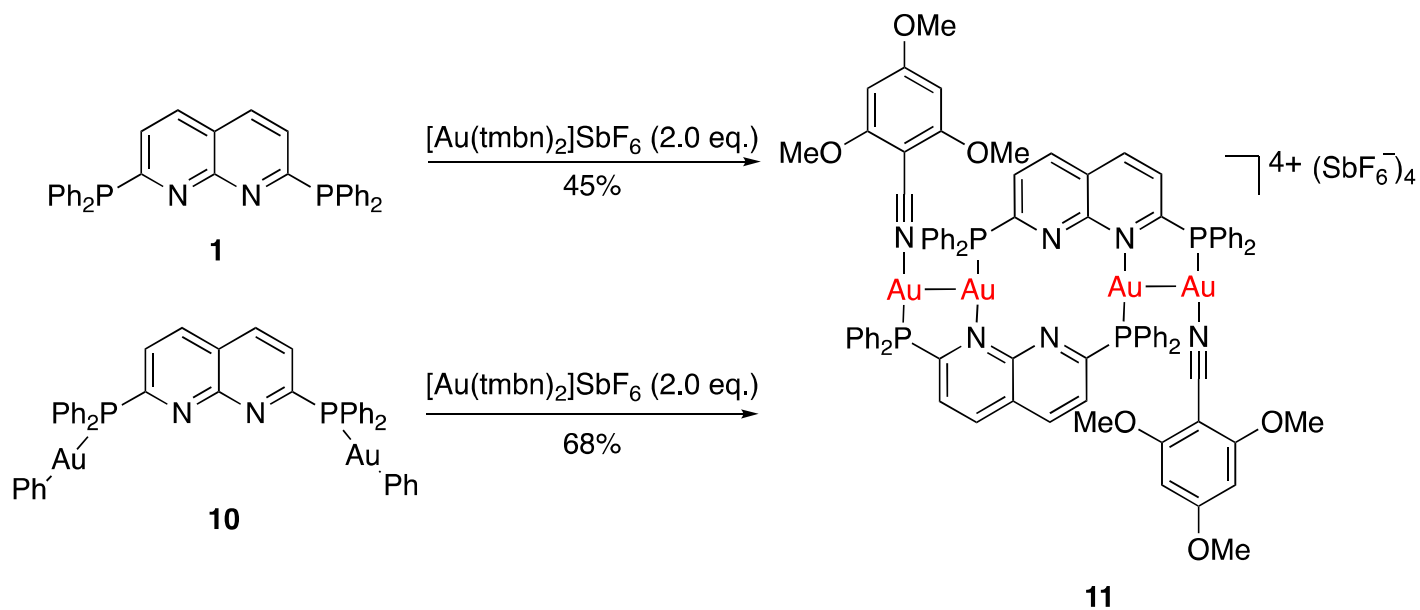
Polynuclear gold clusters synthesis



Polynuclear gold clusters synthesis

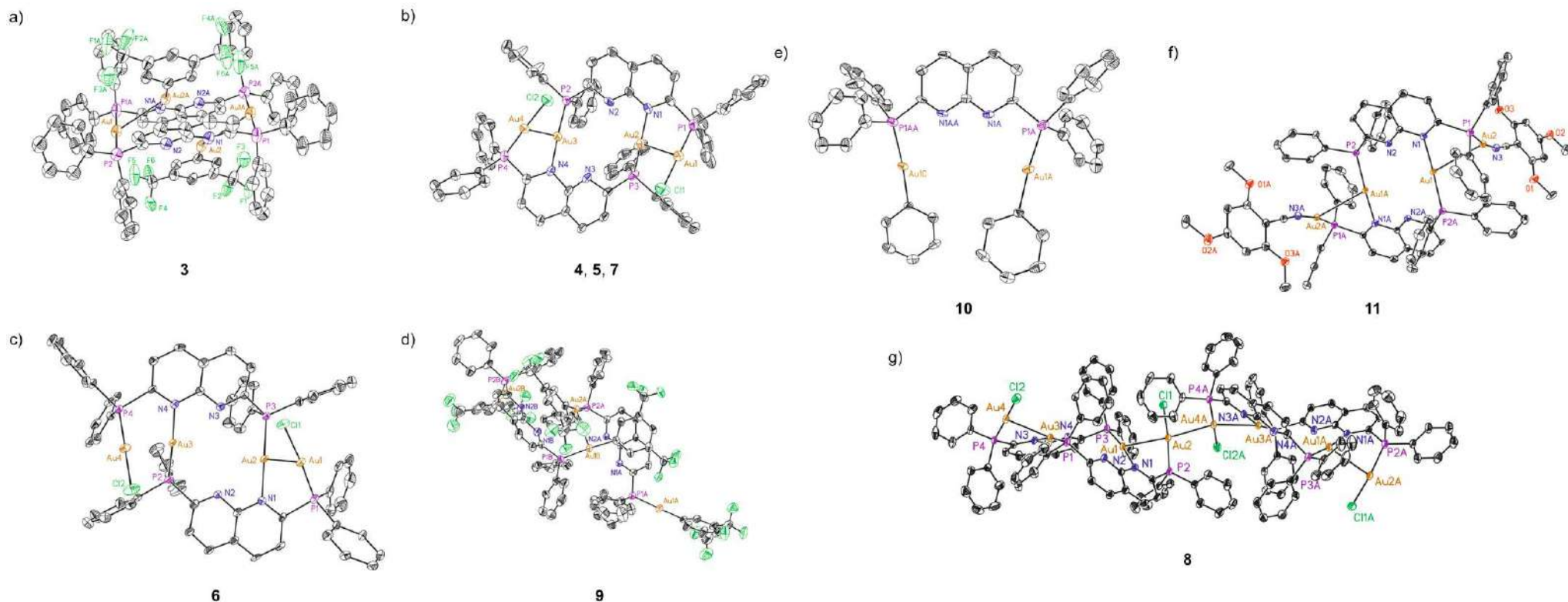


A: NaBPh_4 (2.5 eq.) rt, 98%; B: PhB(OH)_2 (2.5 eq.), Cs_2CO_3 (2.0 eq.), rt, 98%;
 C: Me_3SnPh (4.0 eq.), Cs_2CO_3 (2.0 eq.), 50 °C, 83%



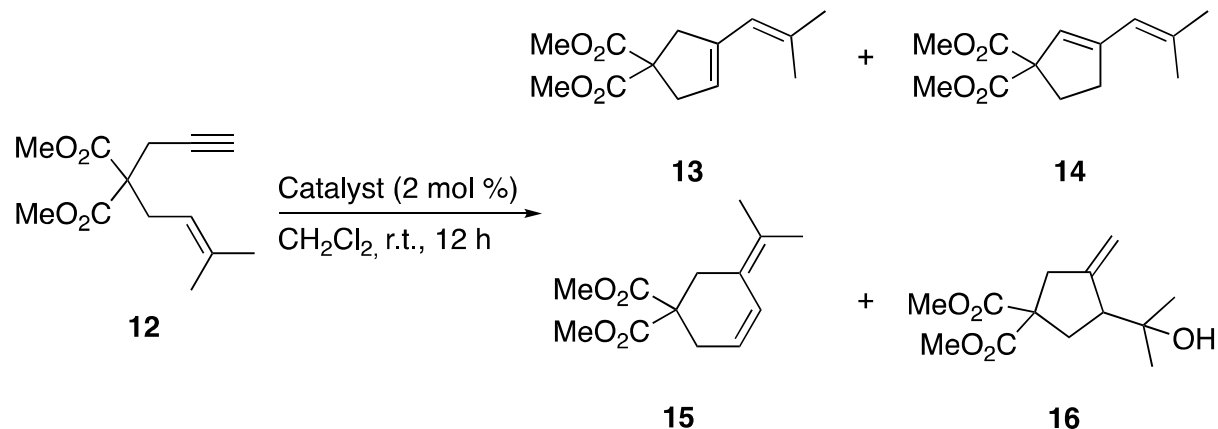
Yang, J.-M.; Echavarren, A. M.* 2018, *To be submitted*.

Polynuclear gold clusters synthesis



Yang, J.-M.; Echavarren, A. M.* 2018, *To be submitted*.

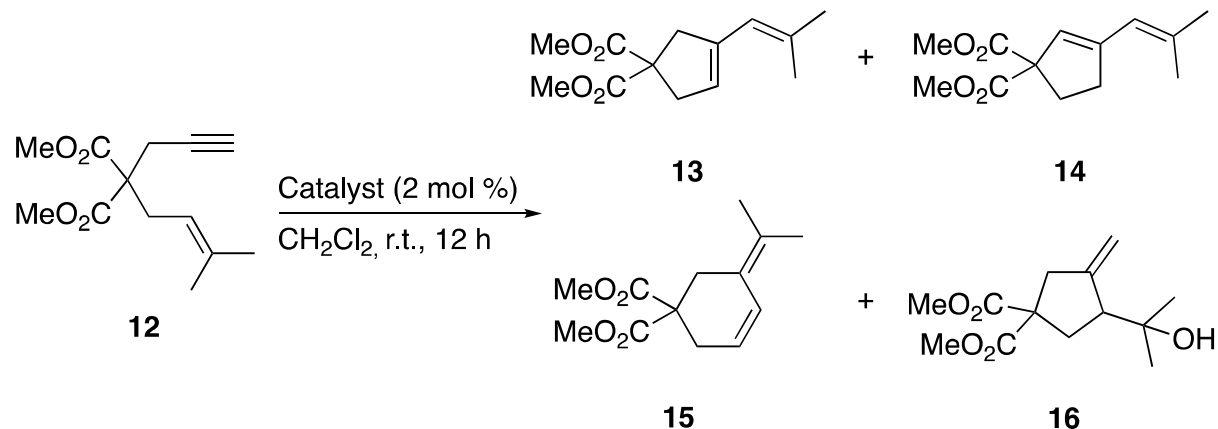
Catalytic studies



Entry ^[a]	Catalyst	Conv. [%] ^[b]	Yield [%] ^[b]			
			13	14	15	16
1	3 , L ₂ Au ₄ R ₂ (BAR ^F ₄) ₂	100	85 (83)	0	10	1
2 ^[c]	4 , L ₂ Au ₄ Cl ₂ (PF ₆) ₂	35	21	0	5	1
3	4 , L ₂ Au ₄ Cl ₂ (PF ₆) ₂	100	58	0	18	0
4	5 , L ₂ Au ₄ Cl ₂ (BF ₄) ₂	42	28	0	3	9
5	6 , L ₂ Au ₄ Cl ₂ (SbF ₆) ₂	100	74	0	17	4
6	7 , L ₂ Au ₄ Cl ₂ (OTf) ₂	100	22	0	23	0
7	8 , [L ₂ Au ₄ Cl ₂ (NTf ₂) ₂] _n	100	14	27	29	0
8	10 , LAu ₂ Ph ₂	0	0	0	0	0
9	11 , L ₂ Au ₄ (tmbn) ₂ (SbF ₆) ₄	100	63	0	35	0
10 ^[d]	11 , L ₂ Au ₄ (tmbn) ₂ (SbF ₆) ₄	90	54	0	20	2

[a] Reaction conditions : **12** (0.1 mmol), cat. (2 mol %), CH₂Cl₂ (1.0 mL). [b] Conversions and yields were determined by ¹H NMR spectroscopy using 1,4-diacetylbenzene as internal standard. Value within parentheses is that of the yield of the isolated product after column chromatography. [c] 1 mol % of **4**, reaction time: 3 h. [d] 0.5 mol % of **11**. L = 2,9-bis(diphenylphosphino)-1,8-naphthyridine (dppn). R = 3,5-(CF₃)₂C₆H₃. tmbn = 2,4,6-trimethoxybenzotrile.

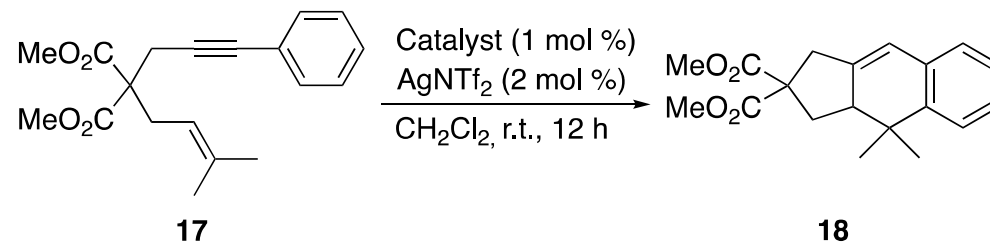
Catalytic studies



Entry ^[a]	Catalyst	Conv. [%] ^[b]	Yield [%] ^[b]			
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1	3 , L ₂ Au ₄ R ₂ (BAR ^F ₄) ₂	100	85 (83)	0	10	1
2 ^[c]	4 , L ₂ Au ₄ Cl ₂ (PF ₆) ₂	35	21	0	5	1
3	4 , L ₂ Au ₄ Cl ₂ (PF ₆) ₂	100	58	0	18	0
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5	6 , L ₂ Au ₄ Cl ₂ (SbF ₆) ₂	100	74	0	17	4
6	7 , L ₂ Au ₄ Cl ₂ (OTf) ₂	100	22	0	23	0
7	8 , [L ₂ Au ₄ Cl ₂ (NTf ₂) ₂] _n	100	14	27	29	0
8	10 , LAu ₂ Ph ₂	0	0	0	0	0
9	11 , L ₂ Au ₄ (tmbn) ₂ (SbF ₆) ₄	100	63	0	35	0
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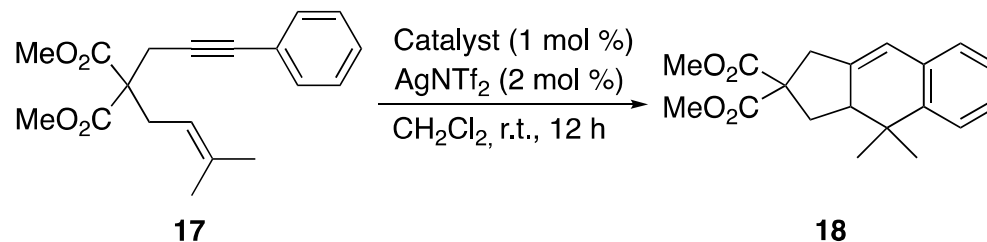
Catalytic studies



Entry ^[a]	Catalyst	Conv. [%] ^[b]	Yield [%] ^[b]
1	3 , $\text{L}_2\text{Au}_4\text{R}_2(\text{BARF}_4)_2$	35	23
2 ^[c]	3 , $\text{L}_2\text{Au}_4\text{R}_2(\text{BARF}_4)_2$	0	0
3 ^[d]	4 , $\text{L}_2\text{Au}_4\text{Cl}_2(\text{PF}_6)_2$	0	0
4	4 , $\text{L}_2\text{Au}_4\text{Cl}_2(\text{PF}_6)_2$	100	83 (77)
5	5 , $\text{L}_2\text{Au}_4\text{Cl}_2(\text{BF}_4)_2$	100	81 (76)
6	6 , $\text{L}_2\text{Au}_4\text{Cl}_2(\text{SbF}_6)_2$	100	75
7	7 , $\text{L}_2\text{Au}_4\text{Cl}_2(\text{OTf})_2$	100	81 (74)
8	8 , $[\text{L}_2\text{Au}_4\text{Cl}_2(\text{NTf}_2)_2]_n$	100	68
9	10 , LAu_2Ph_2	28	25
10 ^[d]	11 , $\text{L}_2\text{Au}_4(\text{tmbn})_2(\text{SbF}_6)_4$	79	59
11	11 , $\text{L}_2\text{Au}_4(\text{tmbn})_2(\text{SbF}_6)_4$	100	60
12	-	39 ^[e]	0

[a] Reaction conditions : **17** (0.1 mmol), cat. (1 mol %), AgNTf₂ (2 mol %), CH₂Cl₂ (1.0 mL). [b] Conversions and yields were determined by ¹H NMR spectroscopy using 1,4-diacetylbenzene as internal standard. Value within parentheses is that of the yield of the isolated product after column chromatography. [c] NaBARF₄ (2 mol %) was added instead of AgNTf₂. [d] Absence of AgNTf₂. [e] Partial decomposition of **17**. L = 2,9-bis(diphenylphosphino)-1,8-naphthyridine (dppn). R = 3,5-(CF₃)₂C₆H₃. tmbn = 2,4,6-trimethoxybenzonitrile.

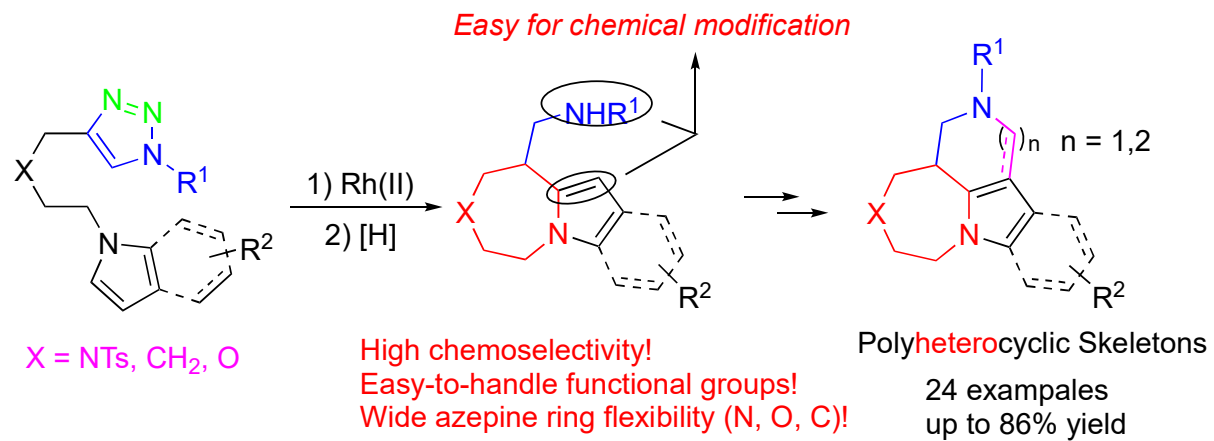
Catalytic studies



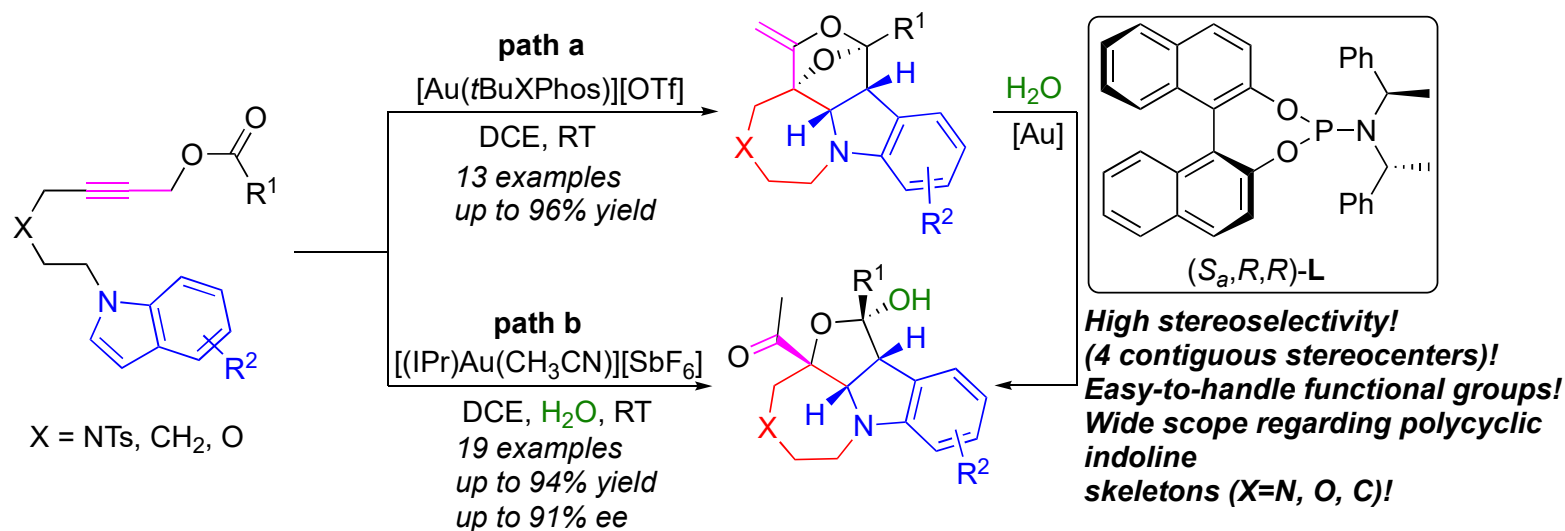
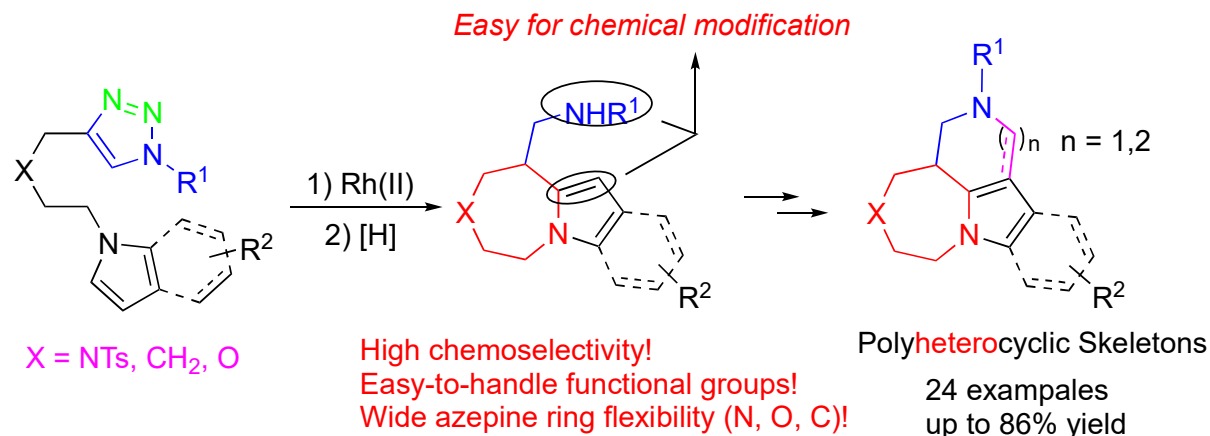
Entry ^[a]	Catalyst	Conv. [%] ^[b]	Yield [%] ^[b]
1	3 , L ₂ Au ₄ R ₂ (BAR ^F ₄) ₂	35	23
2 ^[c]	3 , L ₂ Au ₄ R ₂ (BAR ^F ₄) ₂	0	0
3 ^[d]	4 , L ₂ Au ₄ Cl ₂ (PF ₆) ₂	0	0
4	4 , L ₂ Au ₄ Cl ₂ (PF ₆) ₂	100	83 (77)
5	5 , L ₂ Au ₄ Cl ₂ (BF ₄) ₂	100	81 (76)
6	6 , L ₂ Au ₄ Cl ₂ (SbF ₆) ₂	100	75
7	7 , L ₂ Au ₄ Cl ₂ (OTf) ₂	100	81 (74)
8	8 , [L ₂ Au ₄ Cl ₂ (NTf ₂) ₂] _n	100	68
9	10 , LAu ₂ Ph ₂	28	25
10 ^[d]	11 , L ₂ Au ₄ (tmbn) ₂ (SbF ₆) ₄	79	59
11	11 , L ₂ Au ₄ (tmbn) ₂ (SbF ₆) ₄	100	60
12	-	39 ^[e]	0

[a] Reaction conditions : **17** (0.1 mmol), cat. (1 mol %), AgNTf₂ (2 mol %), CH₂Cl₂ (1.0 mL). [b] Conversions and yields were determined by ¹H NMR spectroscopy using 1,4-diacetylbenzene as internal standard. Value within parentheses is that of the yield of the isolated product after column chromatography. [c] NaBAR^F₄ (2 mol %) was added instead of AgNTf₂. [d] Absence of AgNTf₂. [e] Partial decomposition of **17**. L = 2,9-bis(diphenylphosphino)-1,8-naphthyridine (dppn). R = 3,5-(CF₃)₂C₆H₃. tmbn = 2,4,6-trimethoxybenzonitrile.

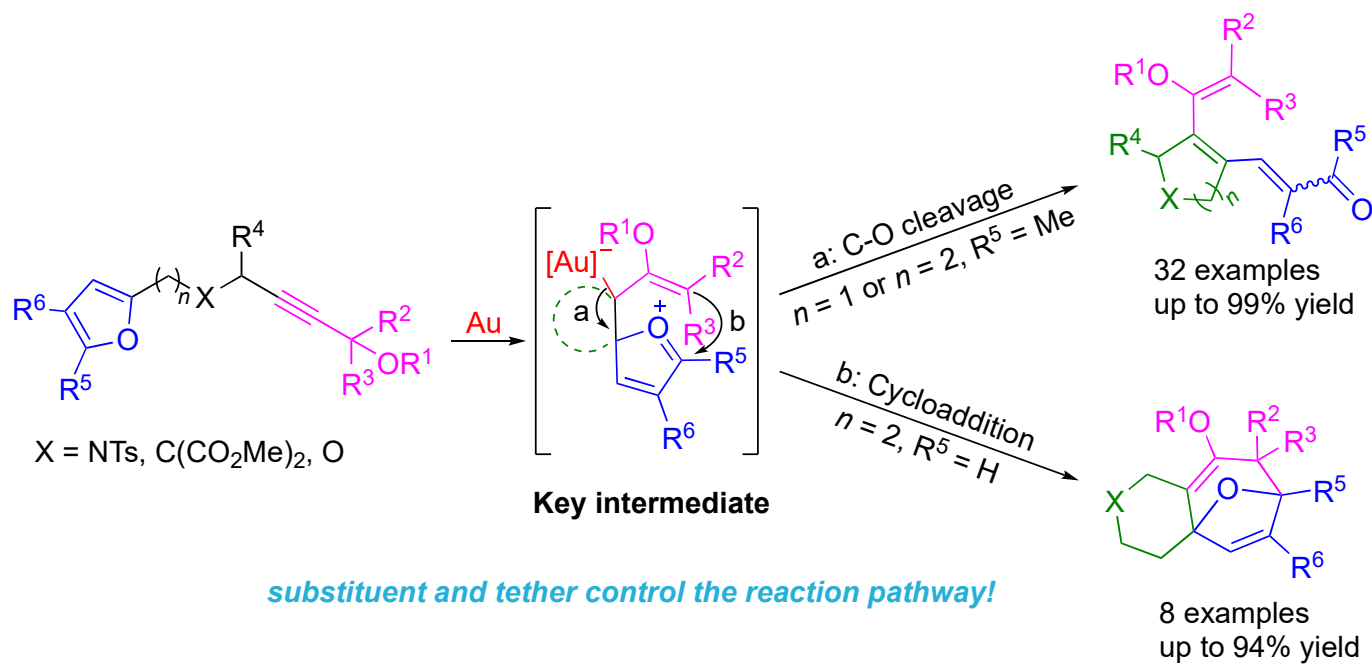
Research Summary



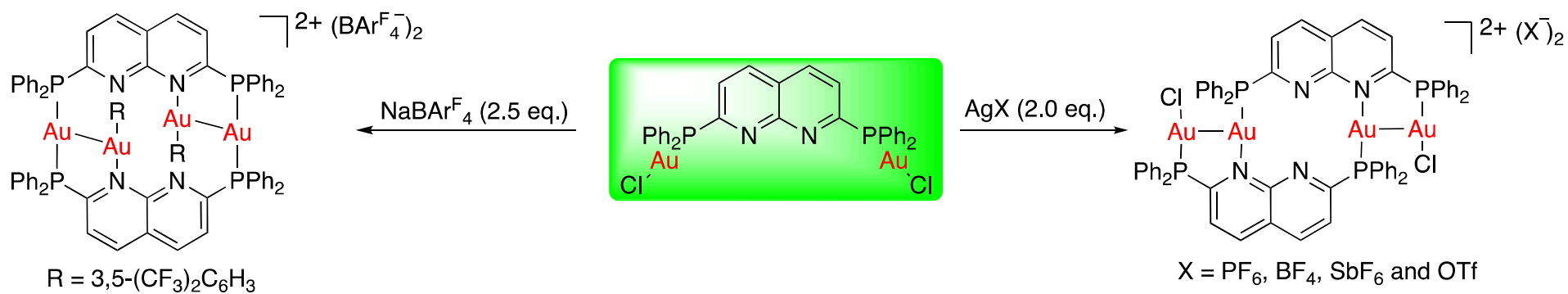
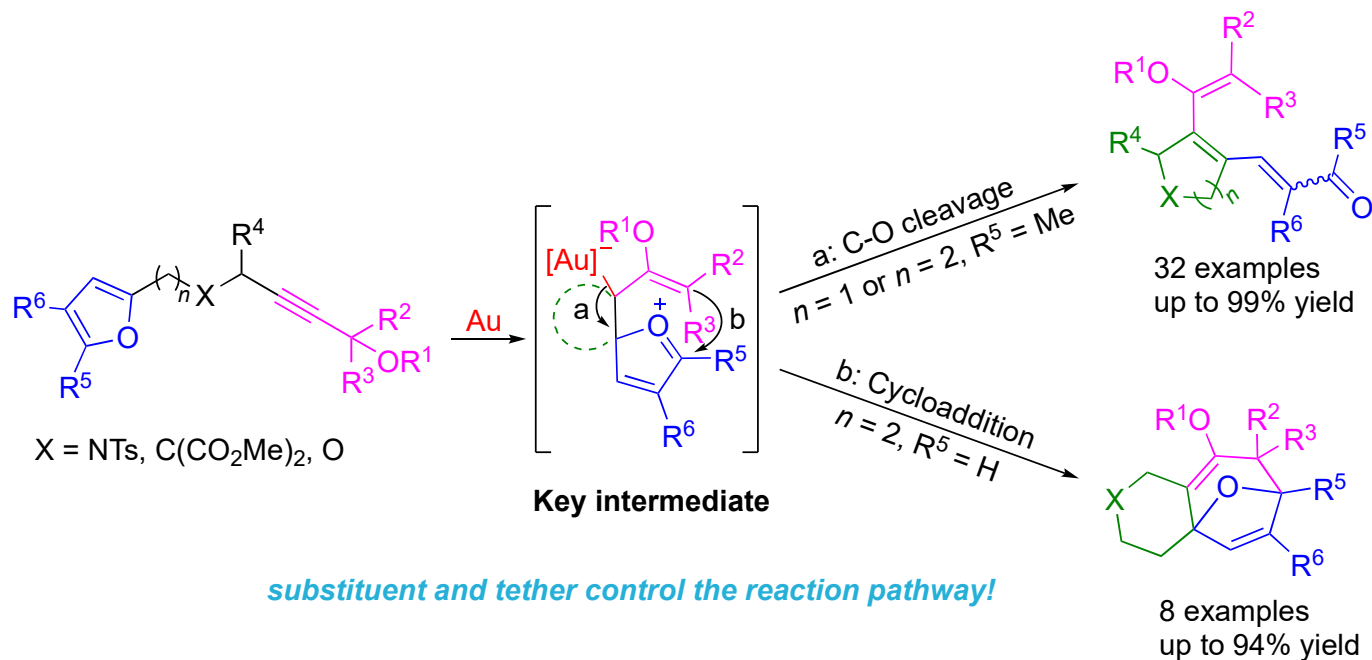
Research Summary



Research Summary



Research Summary

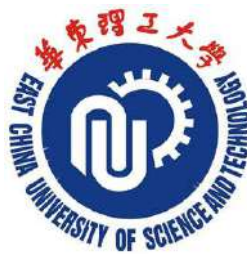


Acknowledgements



Prof. Dr. Min Shi

Prof. Dr. Antonio M Echavarren



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National Natural Science
Foundation of China



Thank you for your attention !