

# 福州大学第二届国际青年学者论坛

## 过渡金属卡宾参与的串联环化反应 及多金簇合物的合成

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博士导师: 施敏 研究员 (SIOC)  
博后合作导师: Antonio M Echavarren (ICIQ)

2017-12-24

# *Institute of Chemical Research of Catalonia (ICIQ)*



# Institute of Chemical Research of Catalonia (ICIQ)



## 主要内容

- 背景介绍
- 正文部分

### 一、铑卡宾参与的杂环合成

- 铑催化的1-磺酰基-1, 2, 3-三氮唑的分子内环化反应

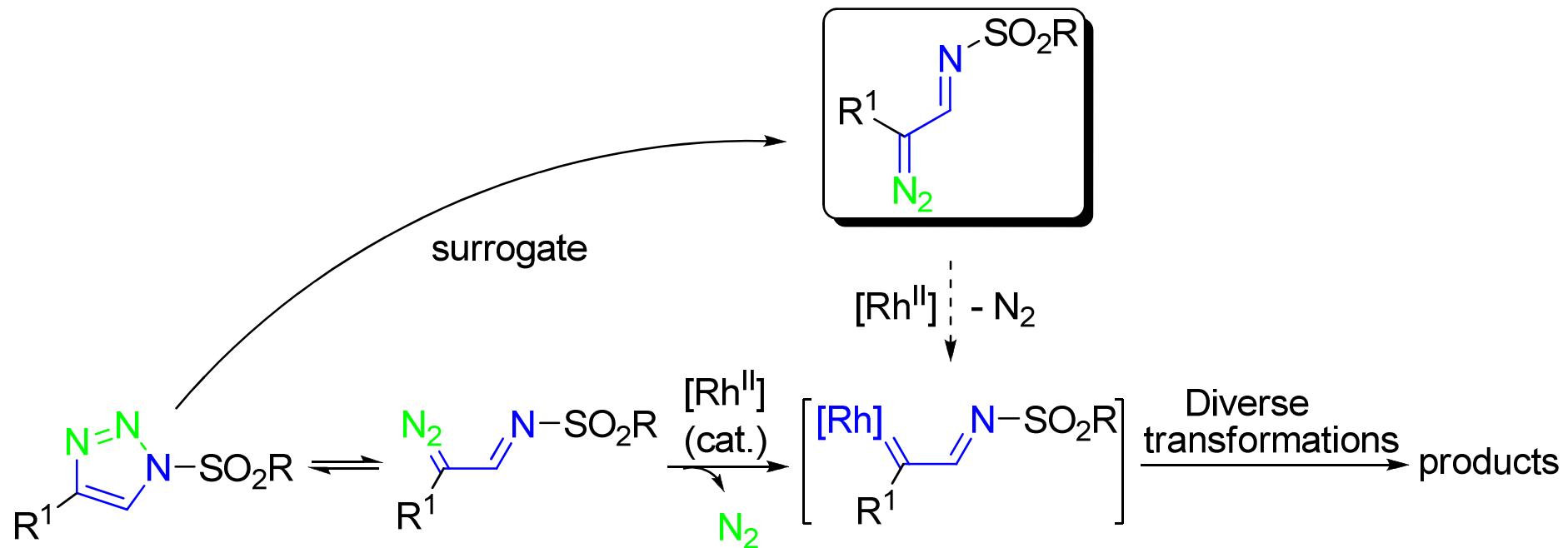
### 二、金卡宾参与的串联环化反应

- 金催化炔丙醇羧酸酯和呋喃分子内的环异构化反应
- 金催化炔丙醇羧酸酯和吲哚分子内的环化反应

### 三、多金簇合物的合成、表征以及催化活性研究

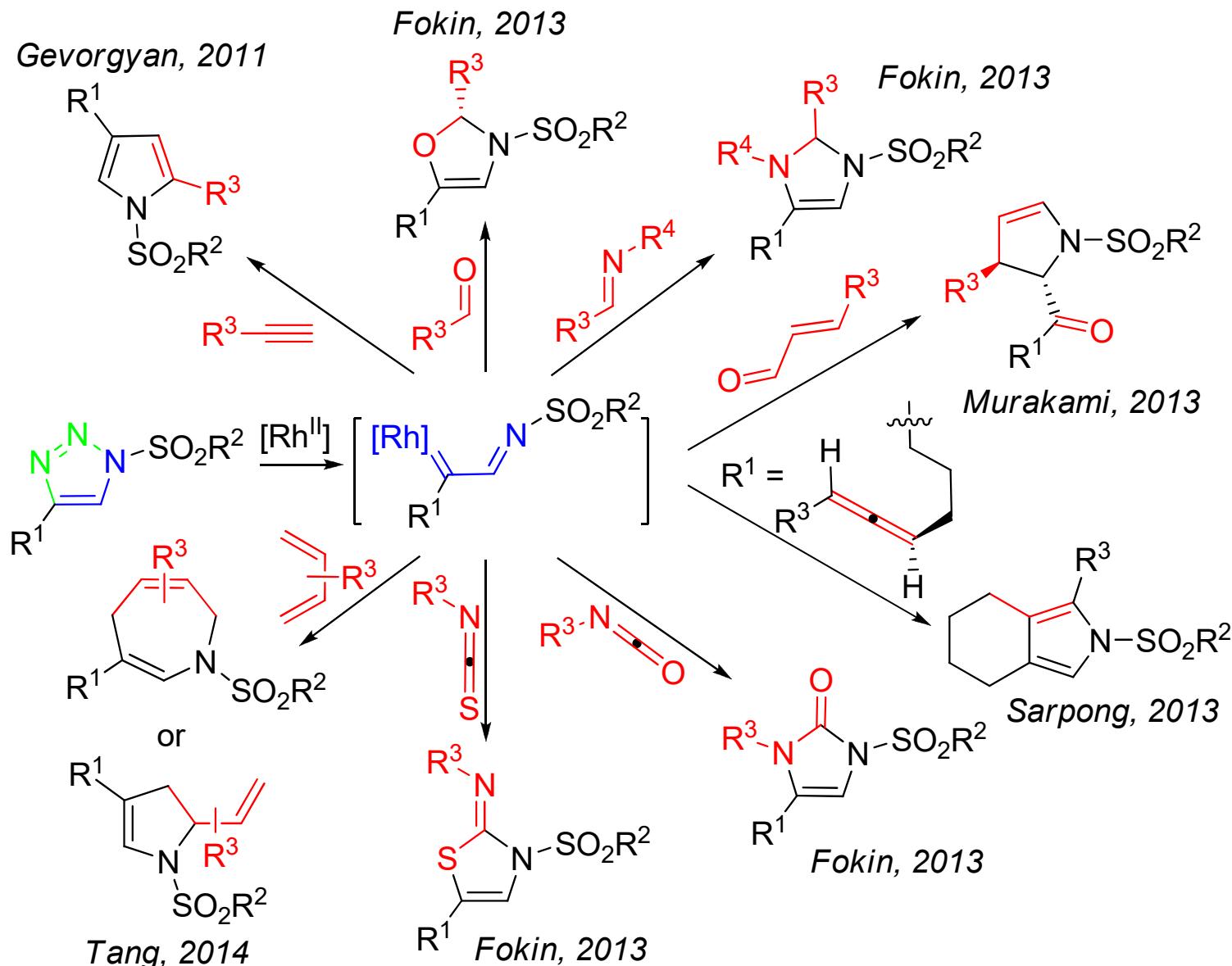
- 研究总结
- 致谢

## 三氮唑



Gulevich, A. V.; Gevorgyan, V. *Angew. Chem. Int. Ed.* **2013**, 52, 1371-1373.

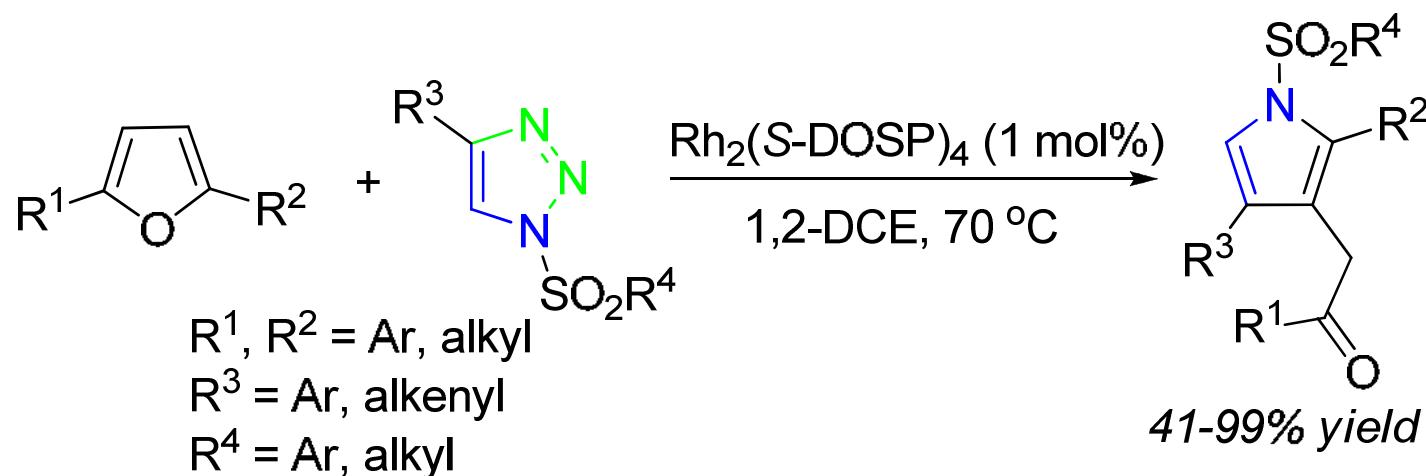
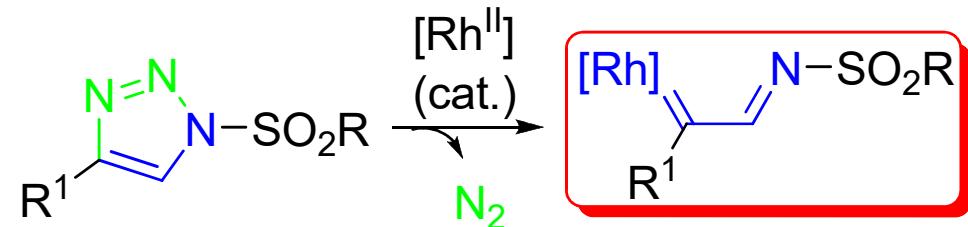
# 三氮唑产生铑卡宾后的转化



(a) Gulevich, A. V.; Gevorgyan, V. *Angew. Chem. Int. Ed.* **2013**, 52, 1371-1373. (b) Davies, H. M. L.; Alford, J. S. *Chem. Soc. Rev.* **2014**, 43, 5151-5162.

# 一、铑卡宾参与的杂环合成

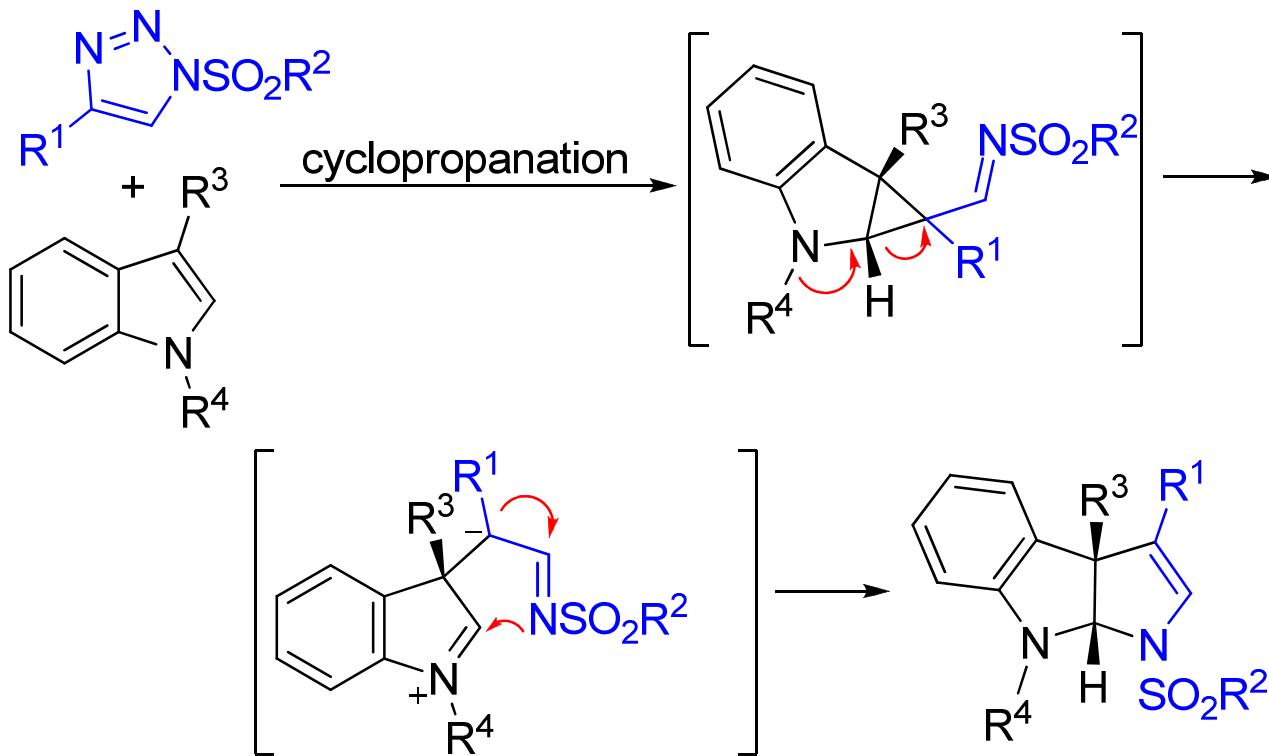
铑是应用最多、最广泛的催化剂之一  
环加成反应、PK反应、C–H键活化



Parr, B. T.; Green, S. A.; Davies, H. M. L. *J. Am. Chem. Soc.* 2013, 135, 4716-4718.

# 铑卡宾参与的杂环合成

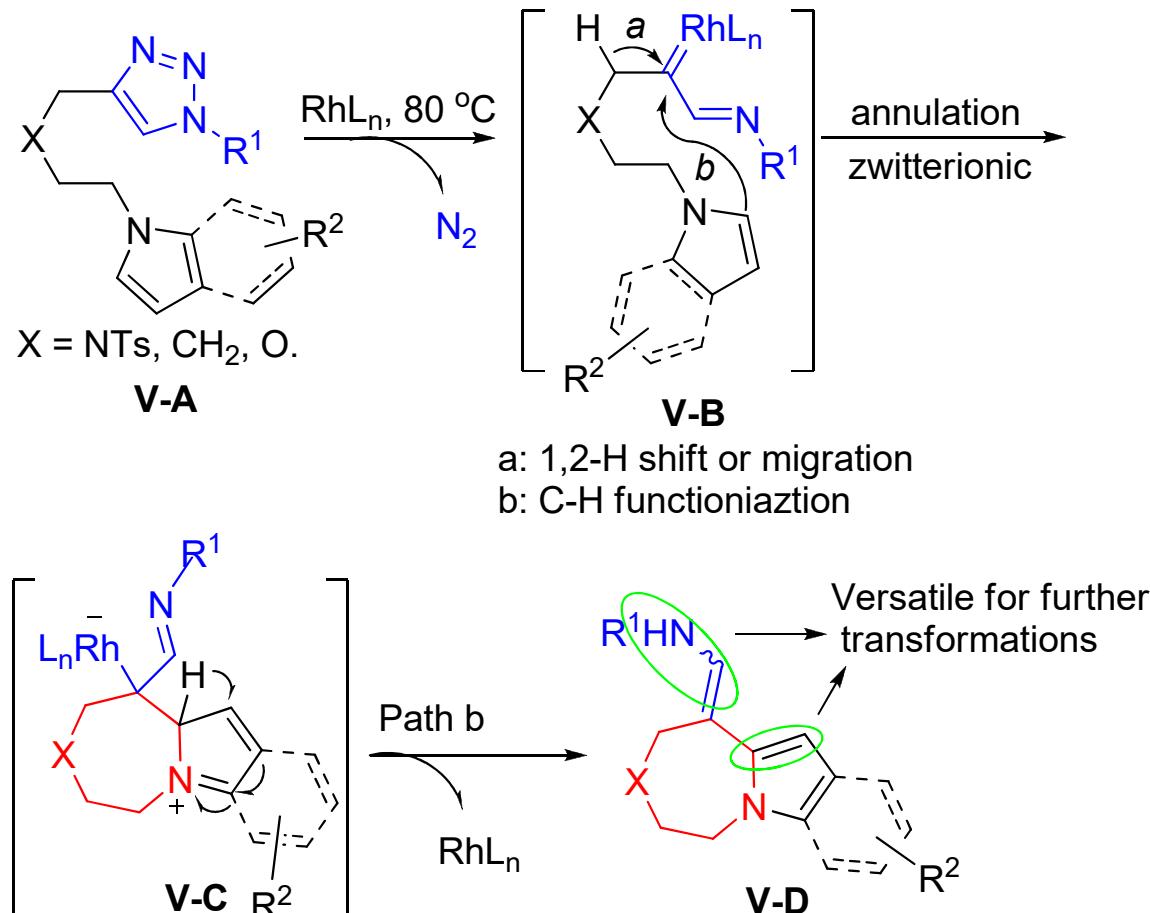
*Intermolecular annulation of indoles (Davies's work)*



Spangler, J. E.; Davies, H. M. L. *J. Am. Chem. Soc.* **2013**, *135*, 6802-6805.

# 铑卡宾参与的杂环合成

*Intramolecular annulation of pyrroles and indoles (This work)*

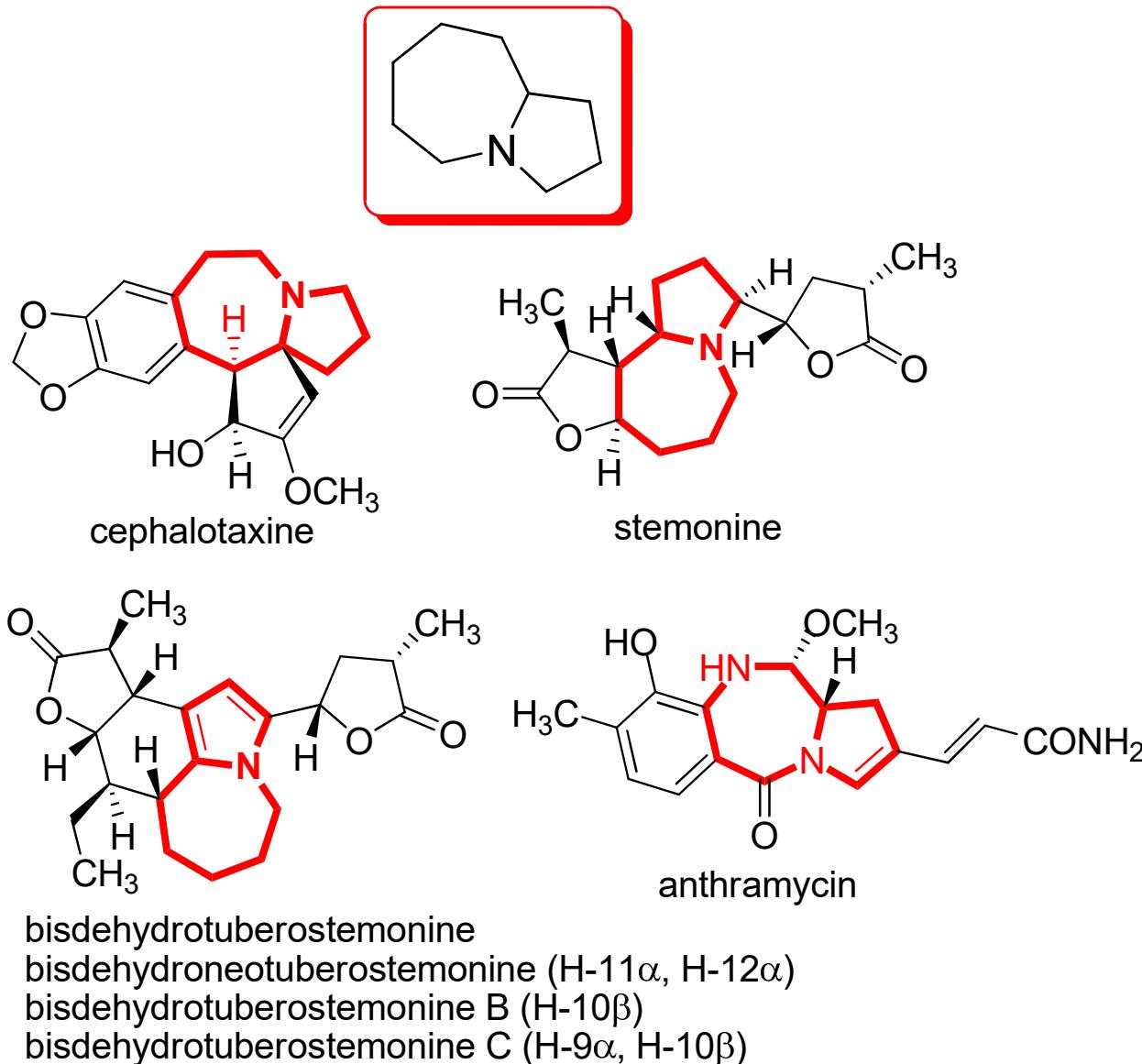


a: 1,2-H shift or migration  
b: C-H functioniaztion

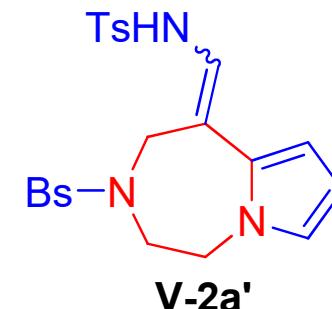
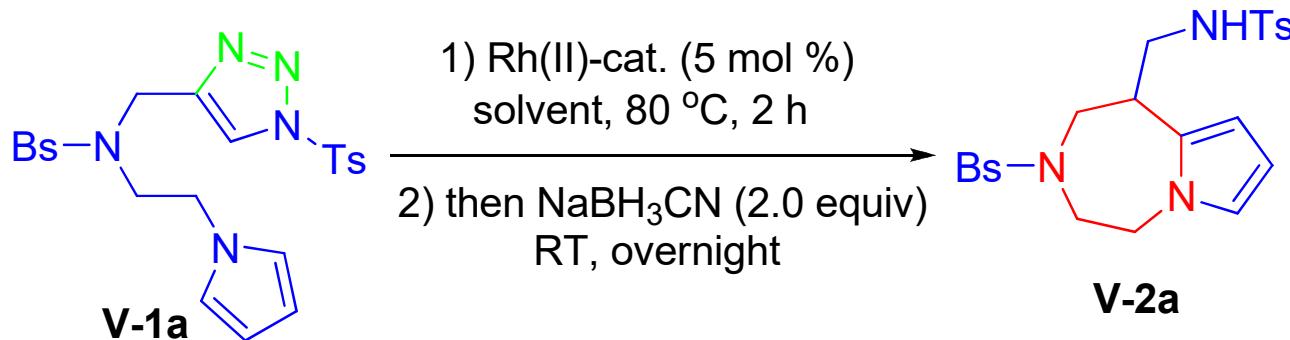
High chemoselectivity (path b only)!  
Easy-to-handle functional groups!  
Wide azepine ring flexibility (N, O, C)!

Yang, J.-M.; Zhu, C.-Z.; Tang, X.-Y.\*; Shi, M.\* *Angew. Chem. Int. Ed.* **2014**, *53*, 5142-5146.

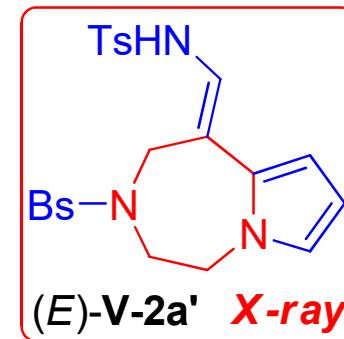
# 铑卡宾参与的杂环合成



# 反应条件优化

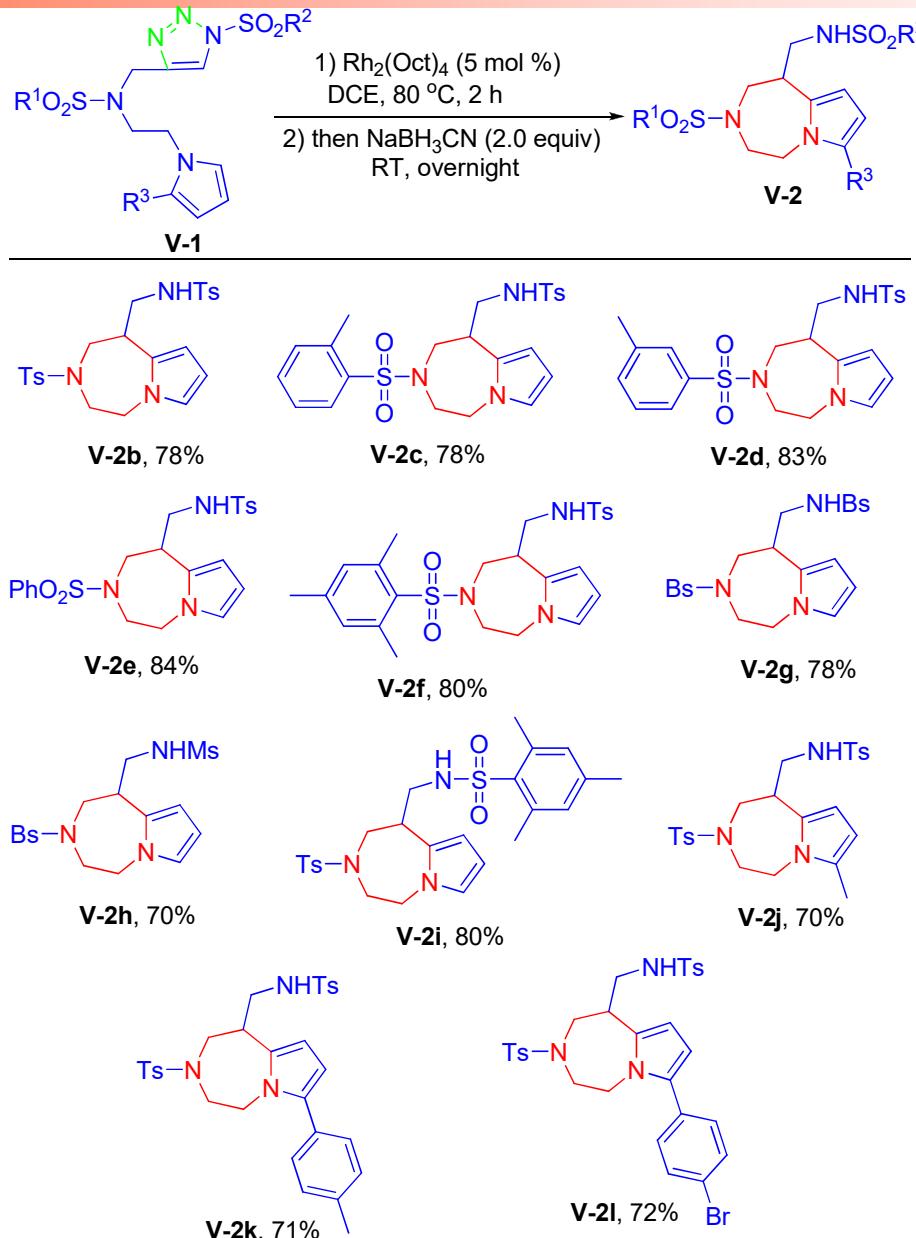


entry <sup>a</sup>	Rh(II)-cat.	solvent	yield (%) <sup>b</sup>
1	<b><i>Rh<sub>2</sub>(Oct)<sub>4</sub></i></b>	<b>DCE</b>	<b>86</b>
2	Rh <sub>2</sub> (Piv) <sub>4</sub>	DCE	80
3	Rh <sub>2</sub> (esp) <sub>2</sub>	DCE	77
4	Rh <sub>2</sub> (OAc) <sub>4</sub>	DCE	78
5	Rh <sub>2</sub> (Adc) <sub>4</sub>	DCE	80
6	Rh <sub>2</sub> (tfa) <sub>4</sub>	DCE	0
7	Rh <sub>2</sub> (S-NTTL) <sub>4</sub>	DCE	70
8	Rh <sub>2</sub> (Oct) <sub>4</sub>	toluene	78
9	Rh <sub>2</sub> (Oct) <sub>4</sub>	cyclohexane	- <sup>c</sup>
10	Rh <sub>2</sub> (Oct) <sub>4</sub>	CHCl <sub>3</sub>	- <sup>c</sup>



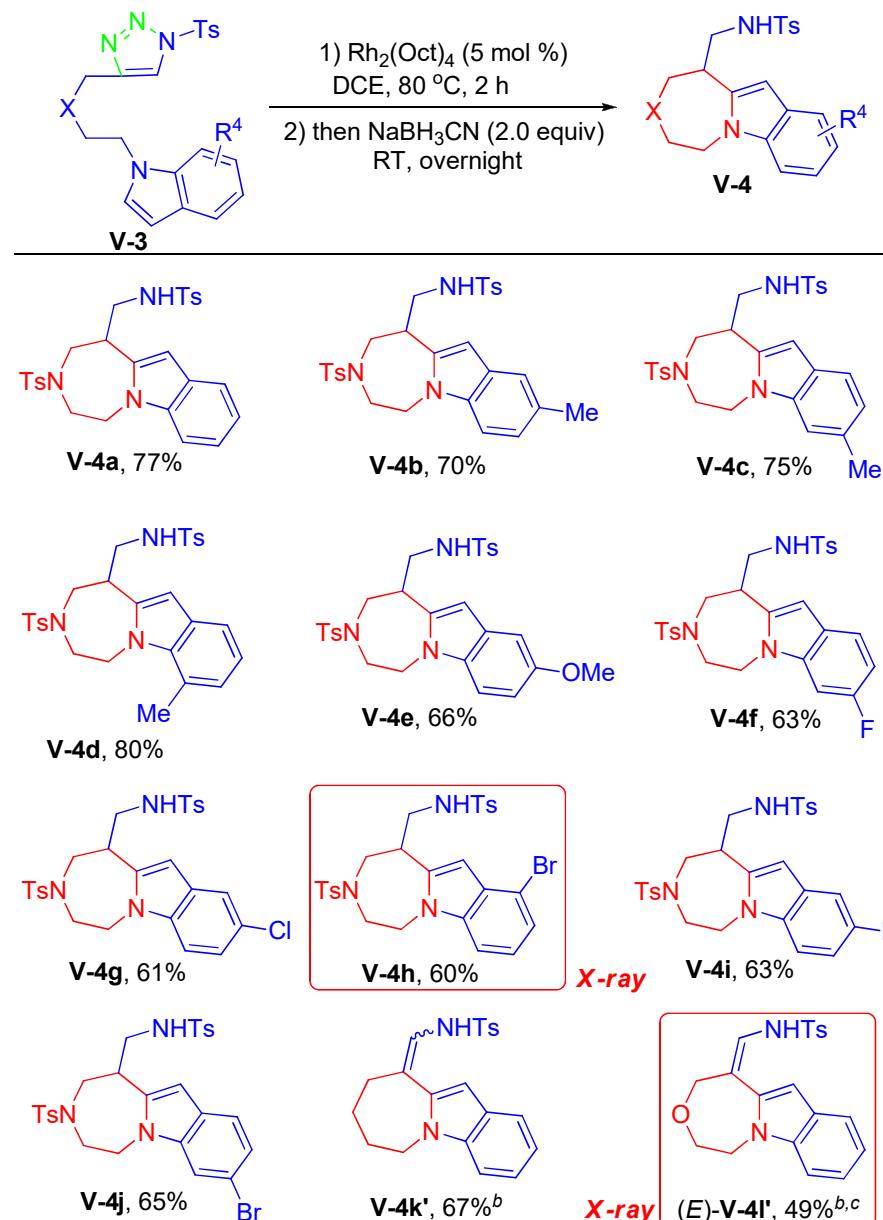
<sup>a</sup> Reaction conditions: 0.1 mmol of **V-1a**; 5 mol% of cat.; 1.0 mL of dry solvent. <sup>b</sup> Isolated yields. <sup>c</sup> undetermined. DCE =1,2-dichloroethane.

# 普适性考查



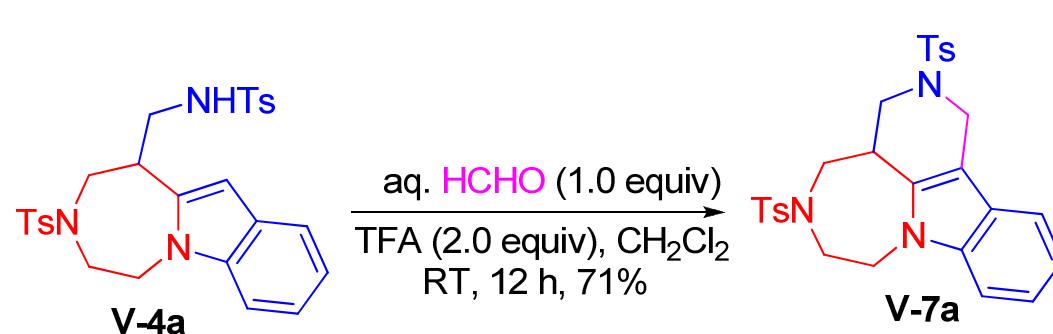
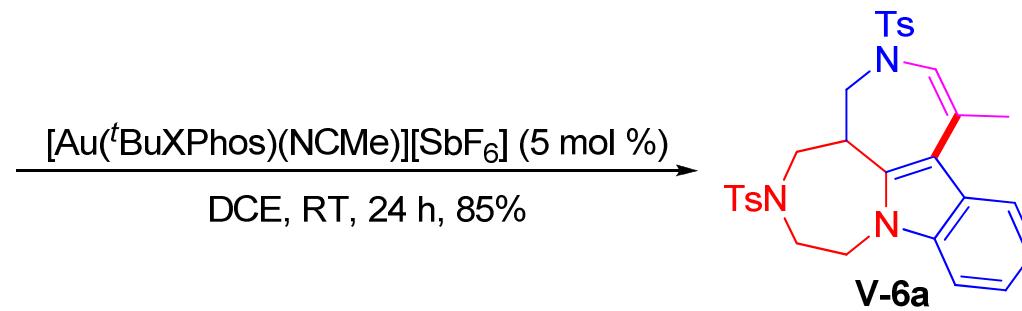
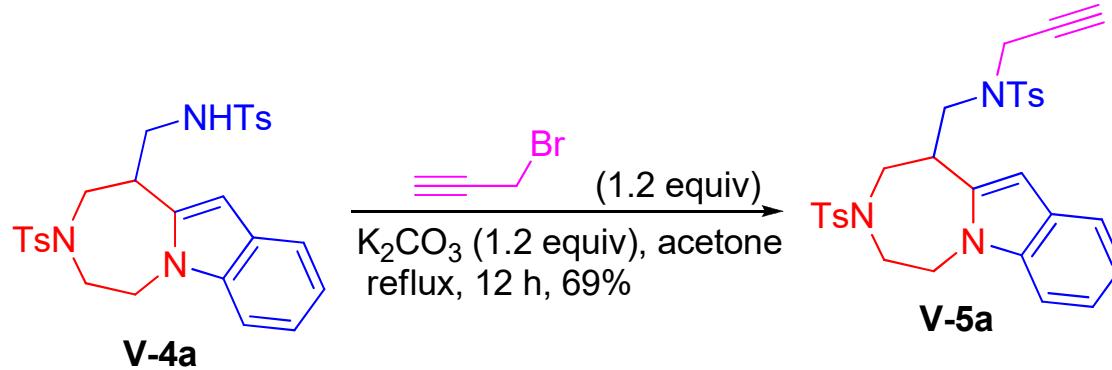
Reaction conditions: 0.1 mmol of **V-1**; 5 mol % of  $\text{Rh}_2(\text{Oct})_4$ ; 1.0 mL anhydrous DCE. Isolated yields.

# 普适性考查

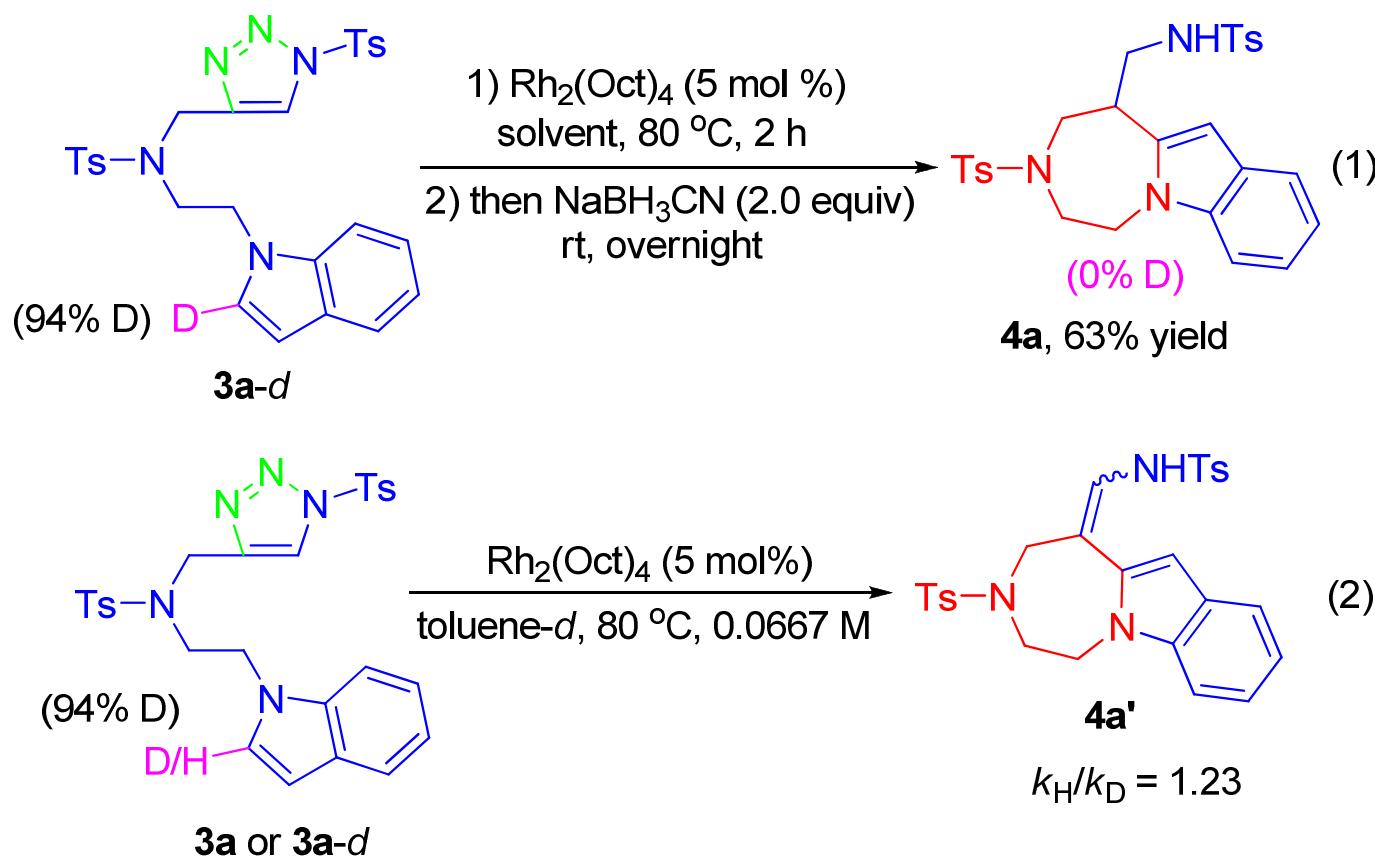


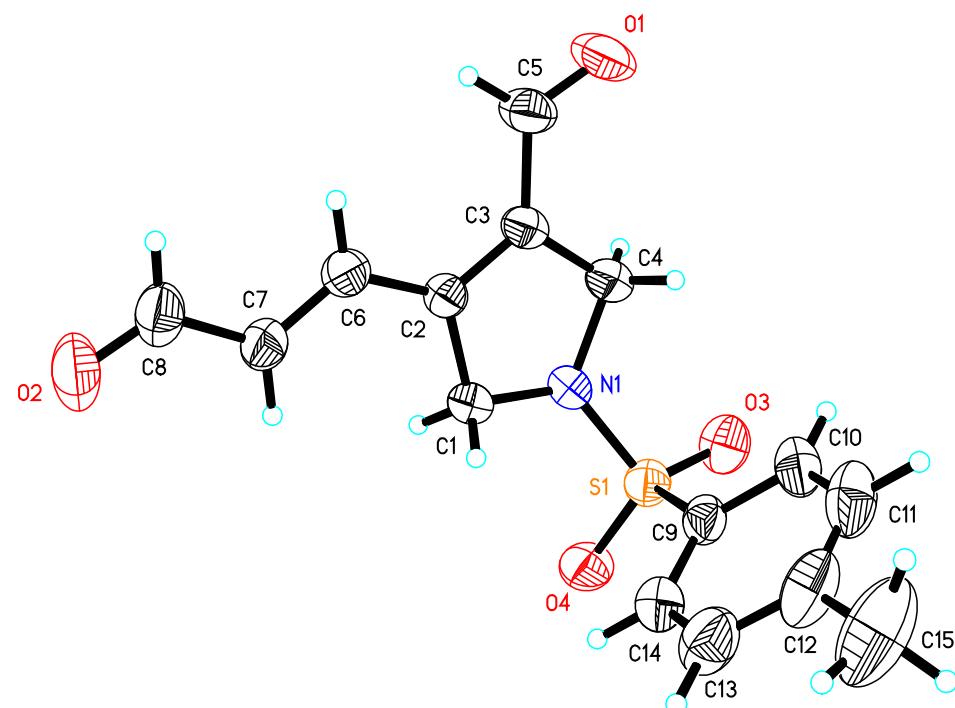
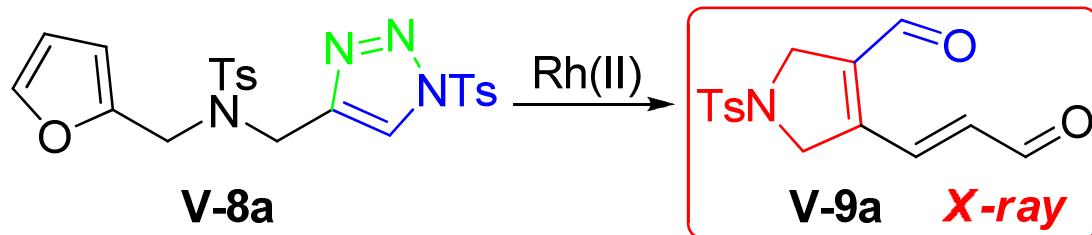
<sup>a</sup> Reaction conditions: 0.1 mmol of **V-3**; 5 mol % of  $\text{Rh}_2(\text{Oct})_4$ ; 1.0 mL anhydrous DCE. Isolated yields. <sup>b</sup> Substrates were performed only in the first step, and the two isomers were not reduced. <sup>c</sup> **V-4l''** was obtained in 43% yield.

## 产物转化

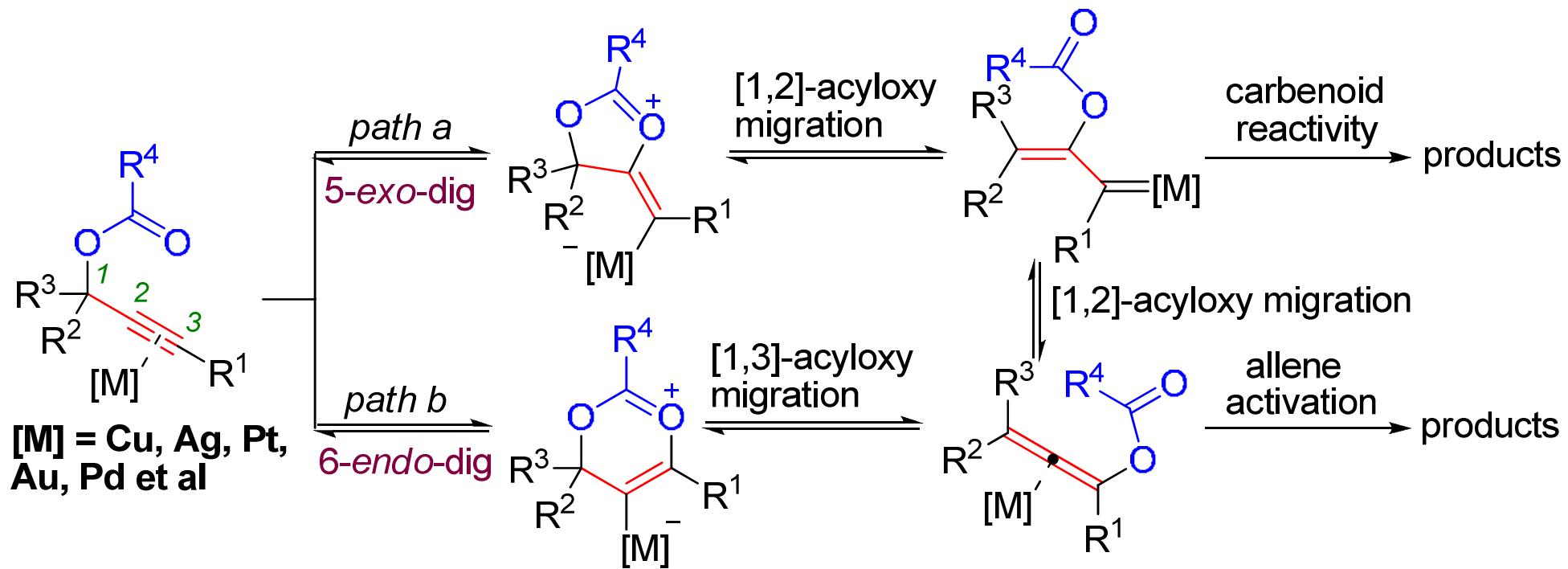


# 氘代及动力学同位素效应实验

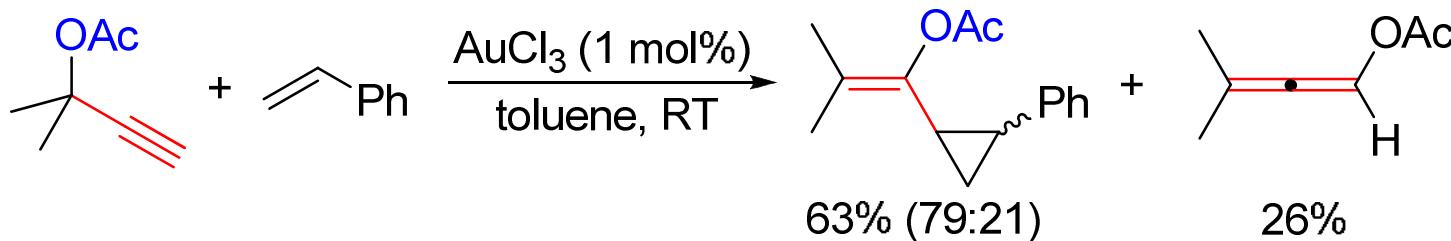




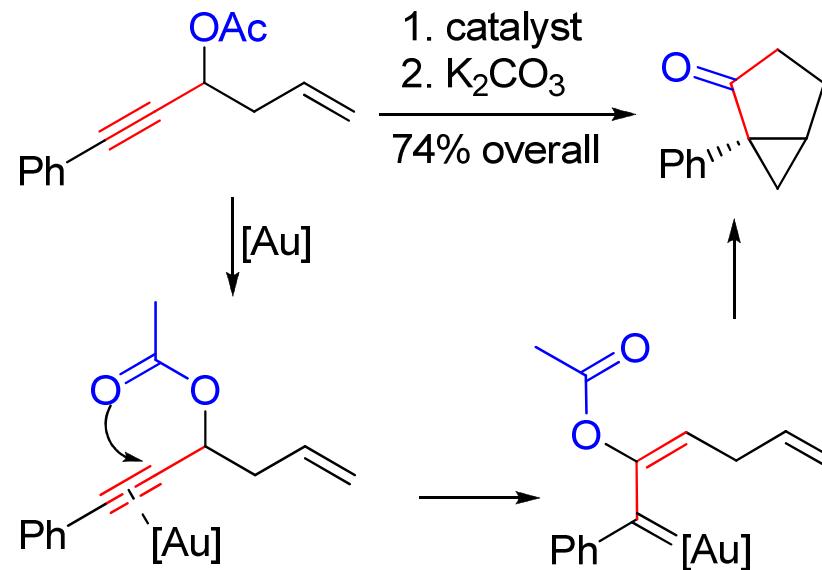
## 炔丙醇羧酸酯



# 炔丙醇羧酸酯[1,2]-迁移

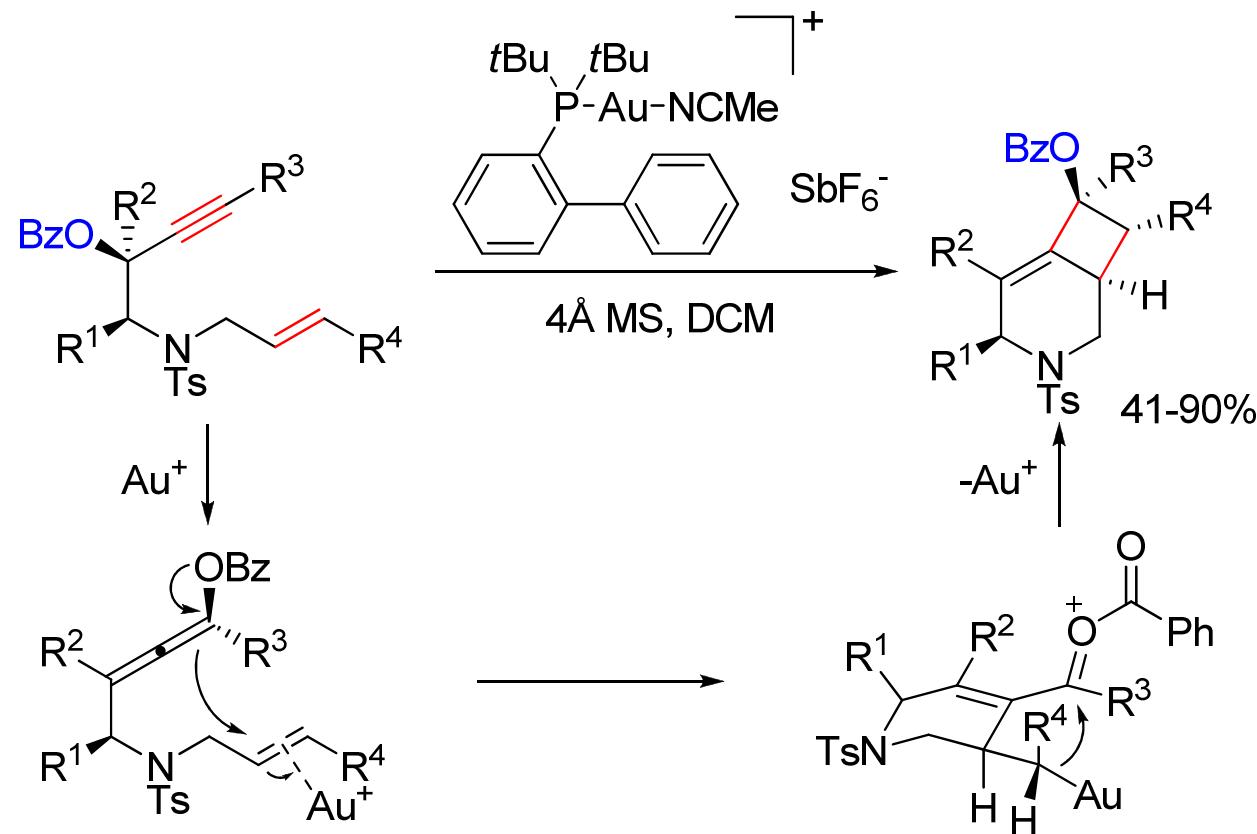


(a) Miki, K.; Ohe, K.; Uemura, S. *Tetrahedron Lett.* **2003**, *44*, 2019-2022. (b) Miki, K.; Ohe, K.; Uemura, S. *J. Org. Chem.* **2003**, *68*, 8505-8513.



Mamane, V.; Gress, T.; Krause, H.; Fürstner, *J. Am. Chem. Soc.* **2004**, *126*, 8654-8655.

# 炔丙醇羧酸酯[1,3]-迁移



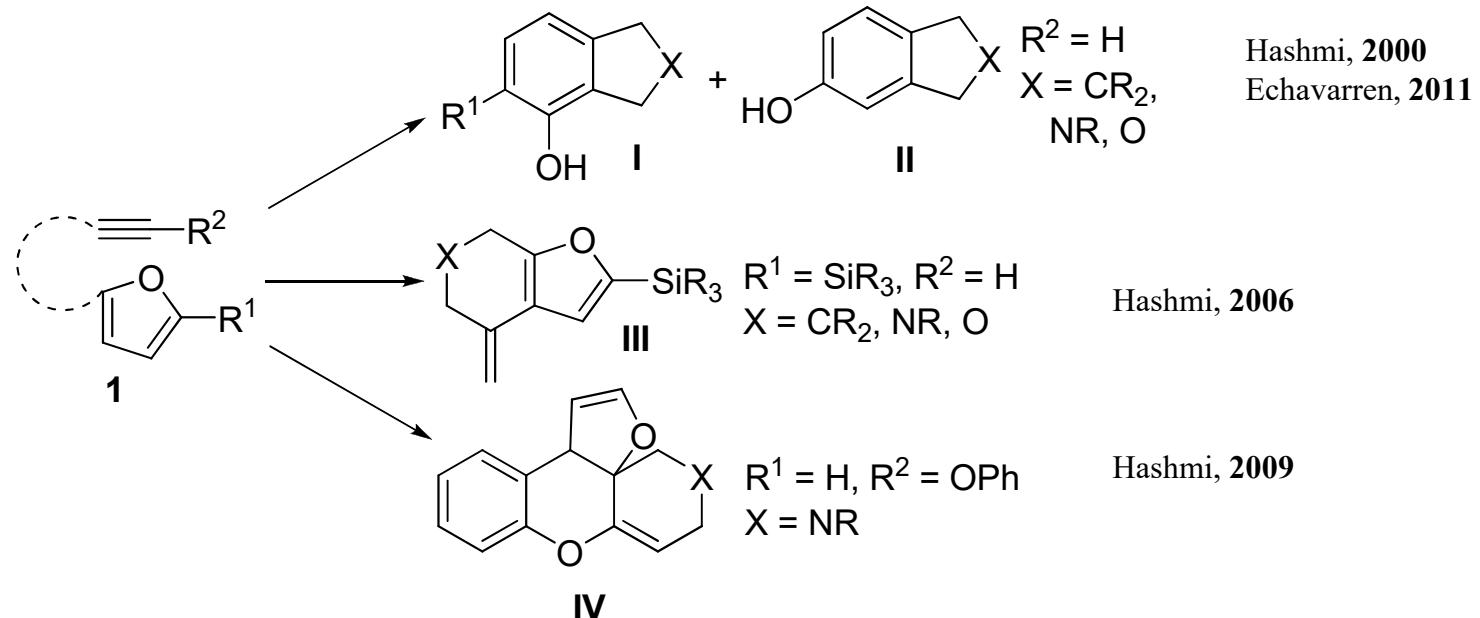
Rao, W.; Susanti, D.; Chan, P. W. H. *J. Am. Chem. Soc.* **2011**, *133*, 15248-15251.

## 二、金卡宾参与串联环化反应



Previous work

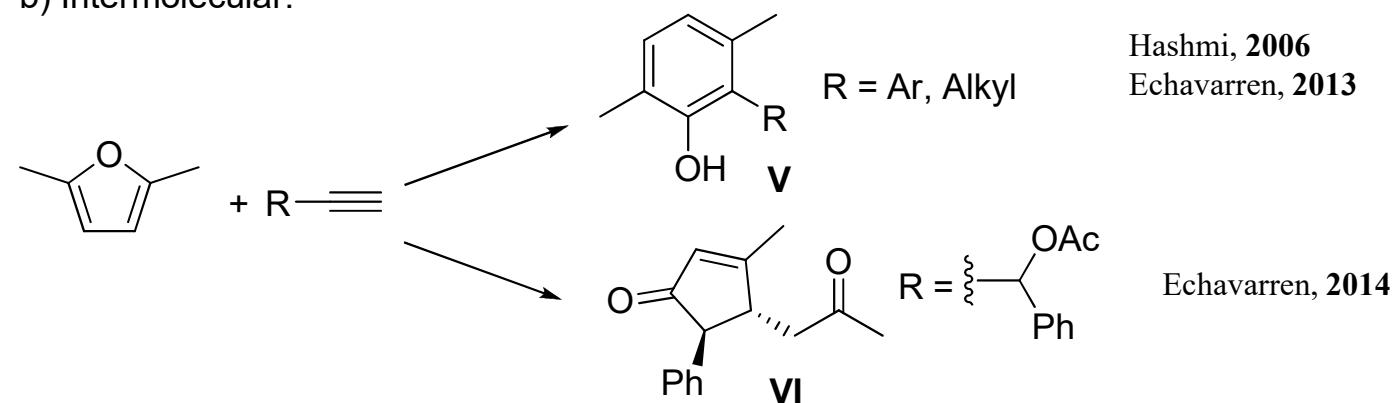
a) intramolecular:



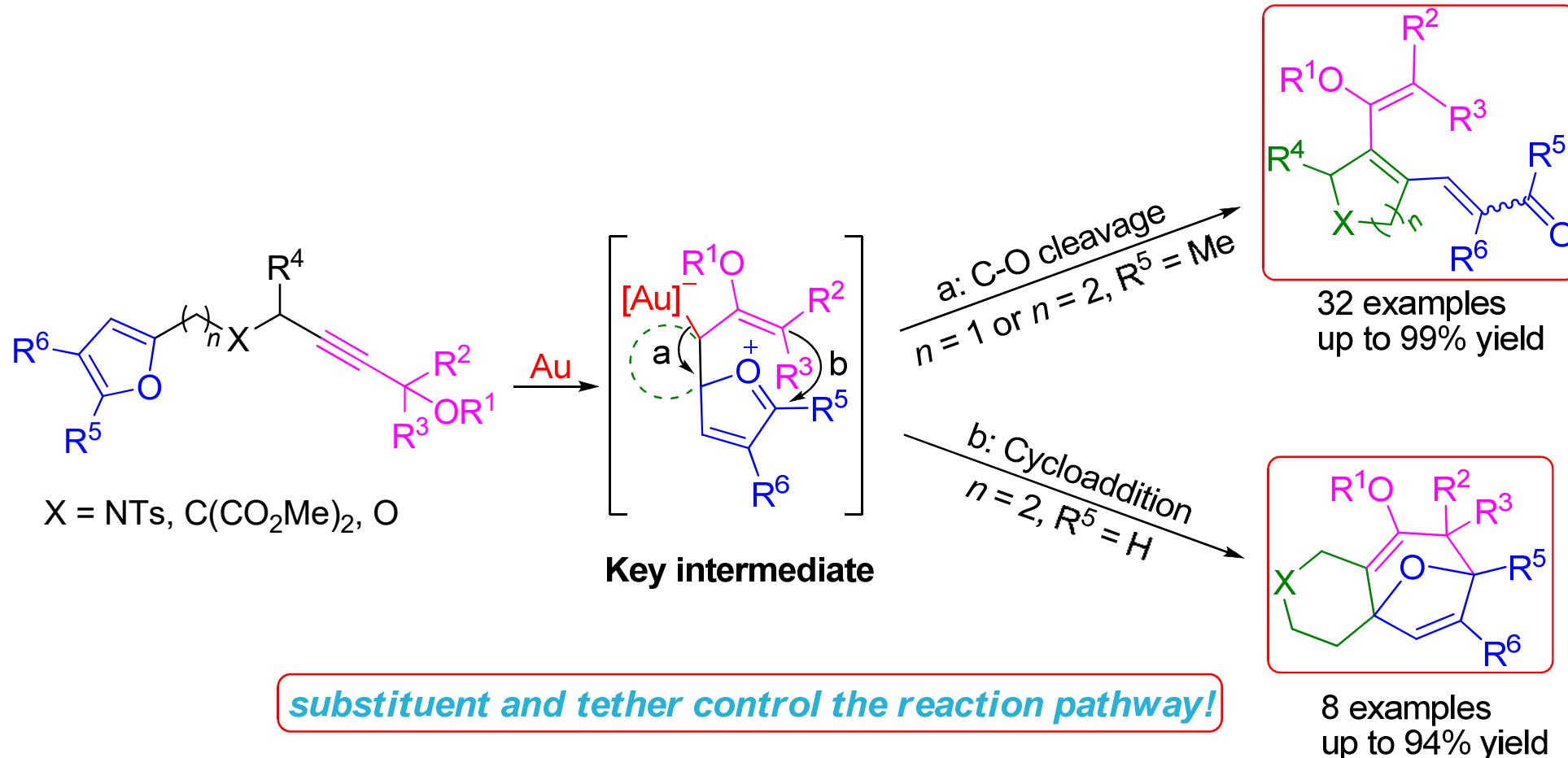
### Gold catalysis

反应条件温和  
操作简单  
反应时间短  
选择性高

b) Intermolecular:

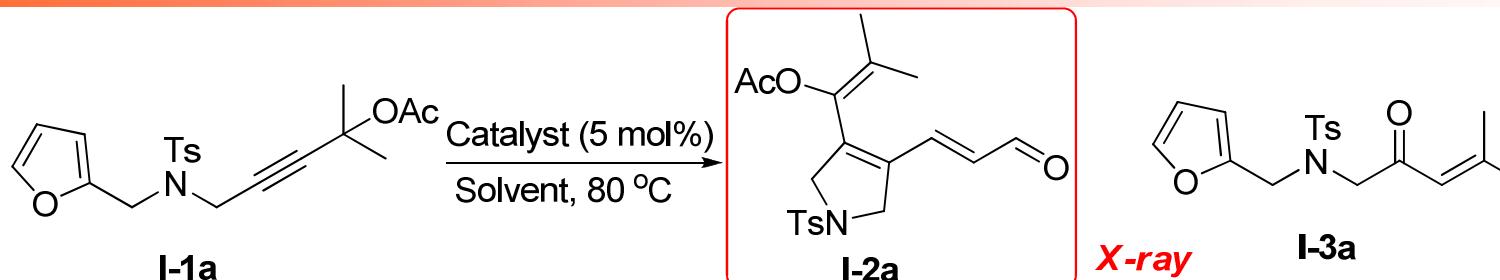


# 金催化炔丙醇羧酸酯和呋喃分子内的环异构化反应

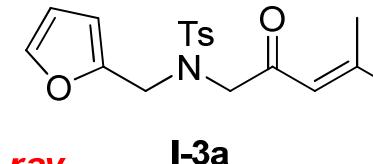


Yang, J.-M.; Tang, X.-Y.\*; Shi, M.\* *Chem. –Eur. J.* **2015**, *21*, 4534–4540.

# 反应条件优化

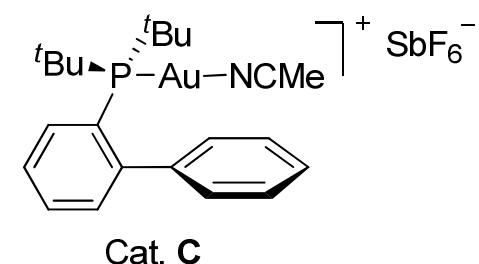
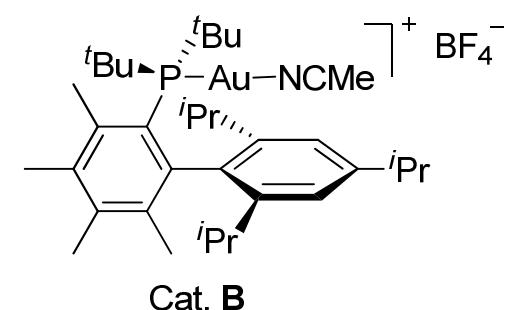
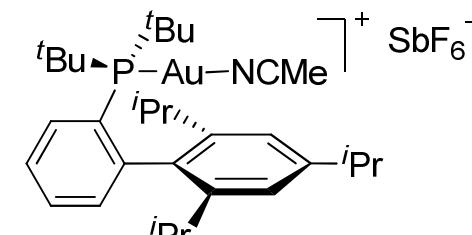


*X-ray*

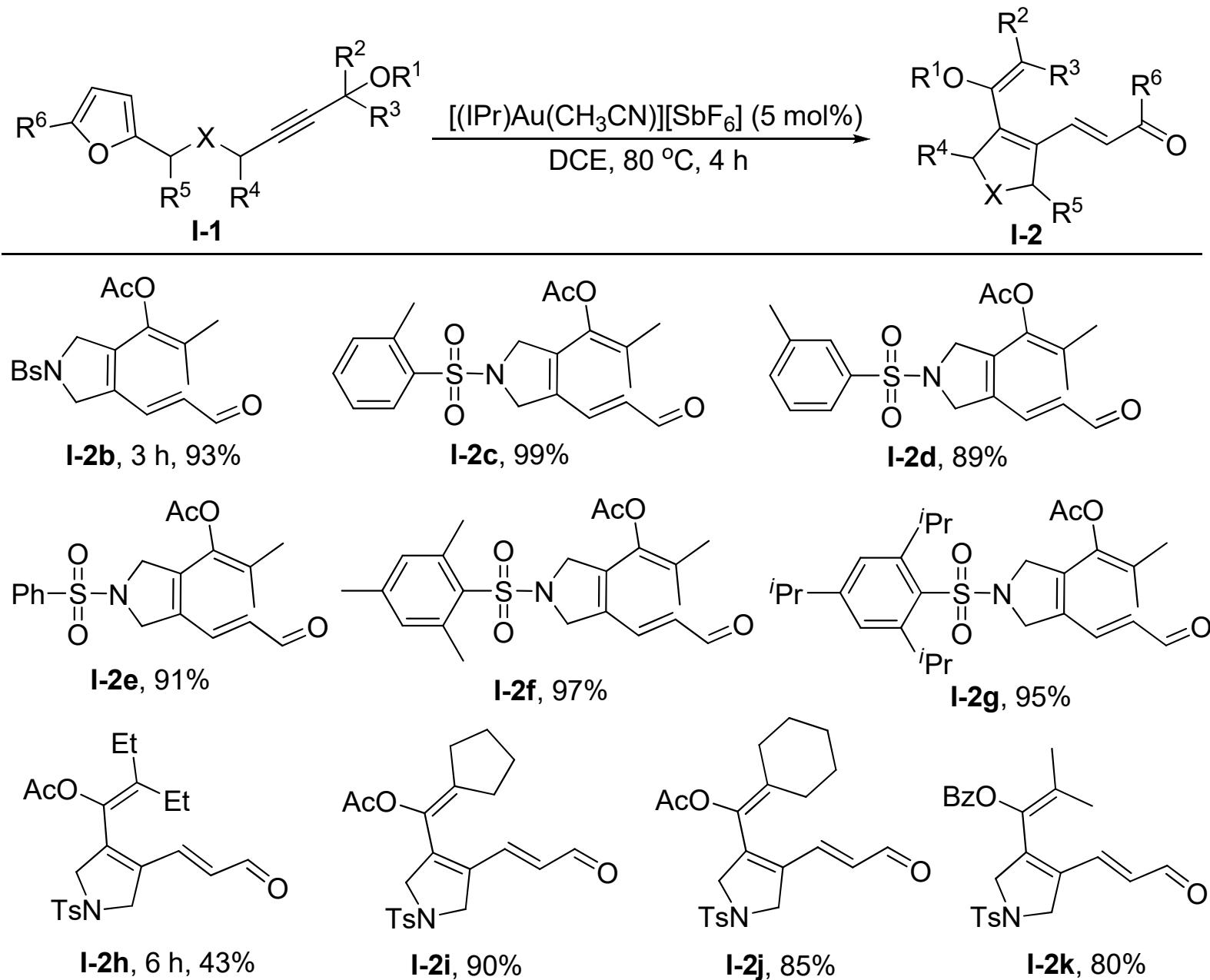


Entry <sup>a</sup>	Catalyst (5 mol%)	Time (h)	Solvent	Yield (%) <sup>b</sup>
1	[Ph <sub>3</sub> PAuCl]/AgSbF <sub>6</sub>	6	DCE	26
2	[IPrAuCl] <sup>c</sup> /AgNTf <sub>2</sub>	4	DCE	81
3	[Au <sub>2</sub> (CH <sub>3</sub> CN) <sub>2</sub> (dppm)][SbF <sub>6</sub> ] <sub>2</sub>	4	DCE	50 <sup>d</sup>
4	<b>A</b>	18	DCE	48
5	<b>B</b>	23	DCE	38
6	<b>C</b>	5.5	DCE	74
7	[Au(CH <sub>3</sub> CN)(Ph <sub>3</sub> P)][SbF <sub>6</sub> ]	6	DCE	88
8	[(IPr)Au][OTf]	4	DCE	76
9	[(IPr)Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ]	4	DCE	88
10	[(IPr)Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ]	6	toluene	72
11	[(IPr)Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ]	6	CH <sub>3</sub> CN	77
12	[(IPr)Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ]	4	1,4-dioxane	46
13 <sup>e</sup>	[(IPr)Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ]	4	DCE	85 <sup>f</sup>
14 <sup>g</sup>	[(IPr)Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ]	4	DCE	84

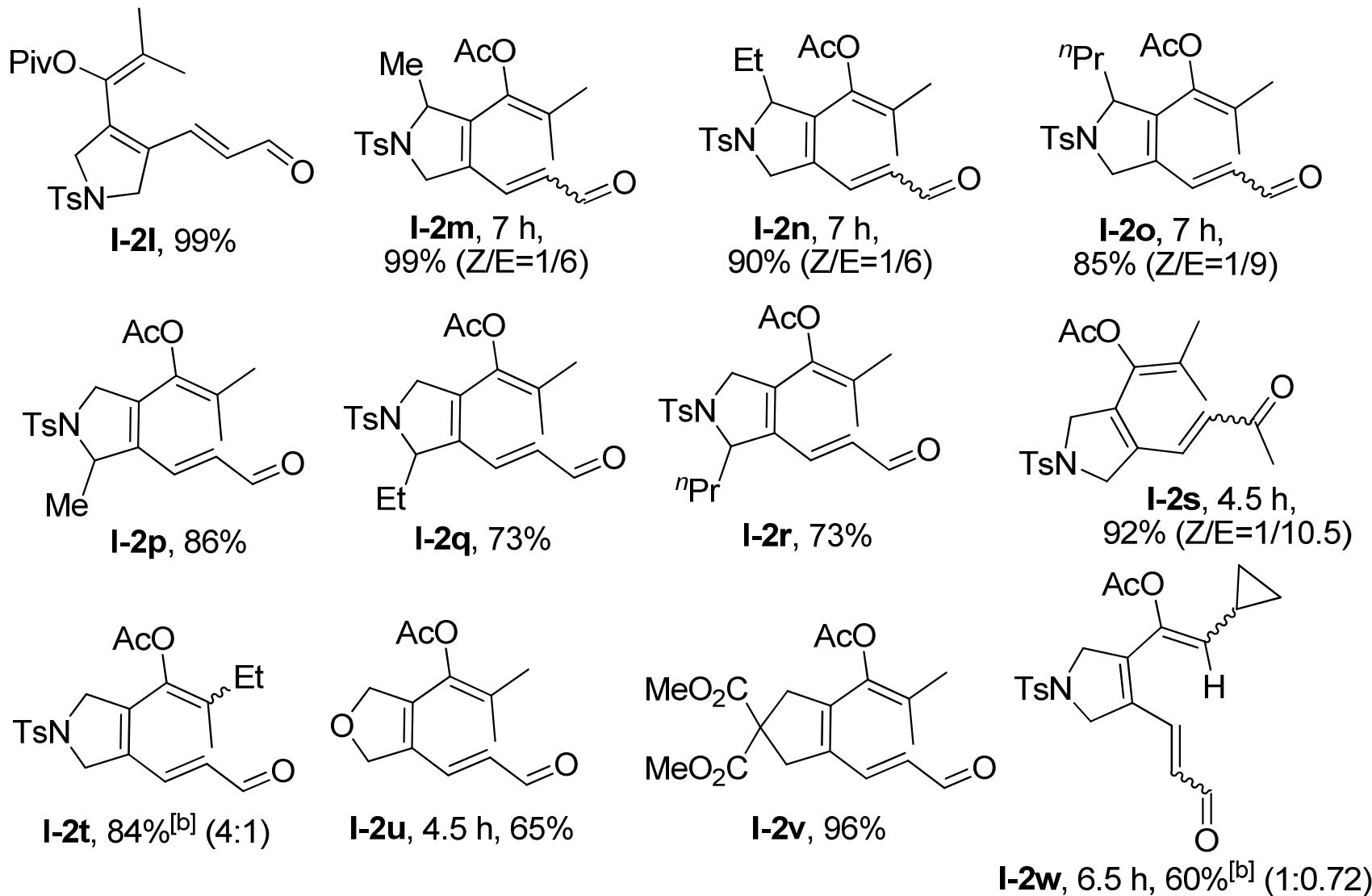
[a] Reaction conditions: **I-1a** (0.2 mmol); Au-cat. (5 mol%); dry solvent (1.0 mL). [b] Yield of isolated products. [c] IPr=[1,3-bis(2,6-diisopropylphenyl)imidazol-2-ylidene]. [d] **I-3a** was obtained in 44% yield. [e] Reaction performed at room temperature. [f] Z/E=1.5/1, determined by <sup>1</sup>H NMR. [g] 10 mol% of catalyst. DCE = 1,2-dichloroethane.



# 普适性考查

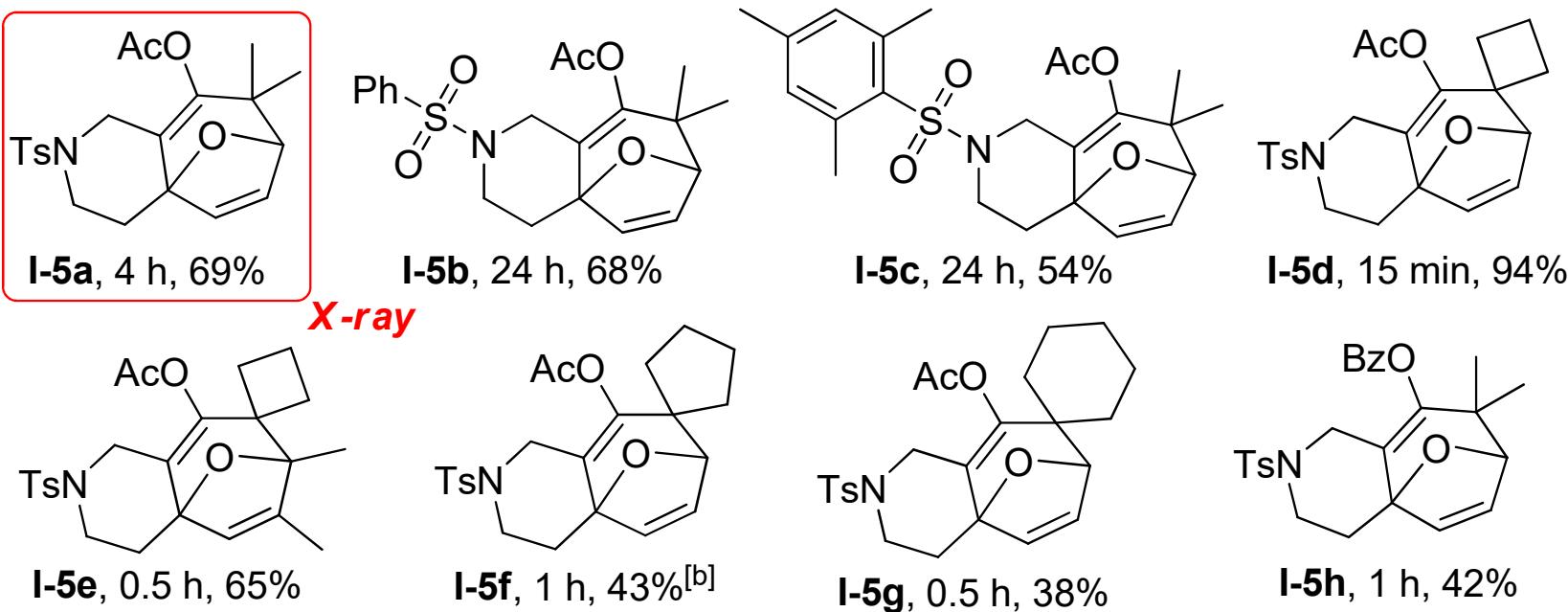
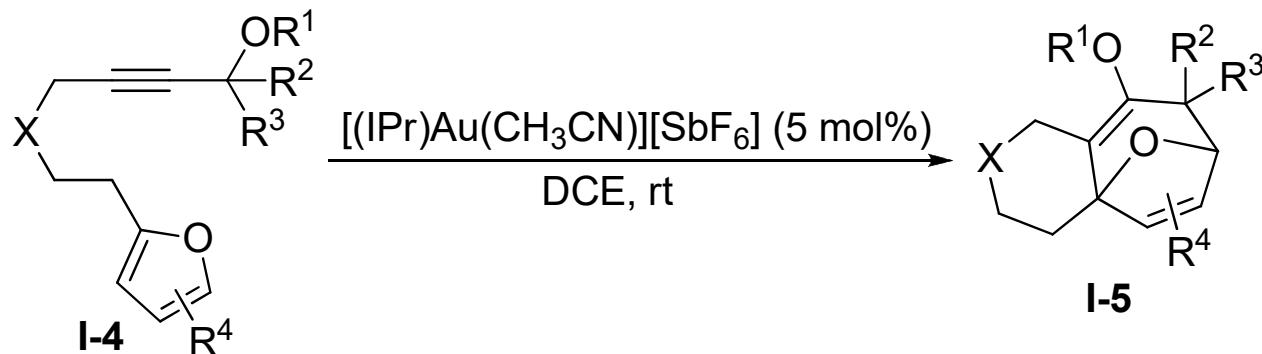


# 普适性考查



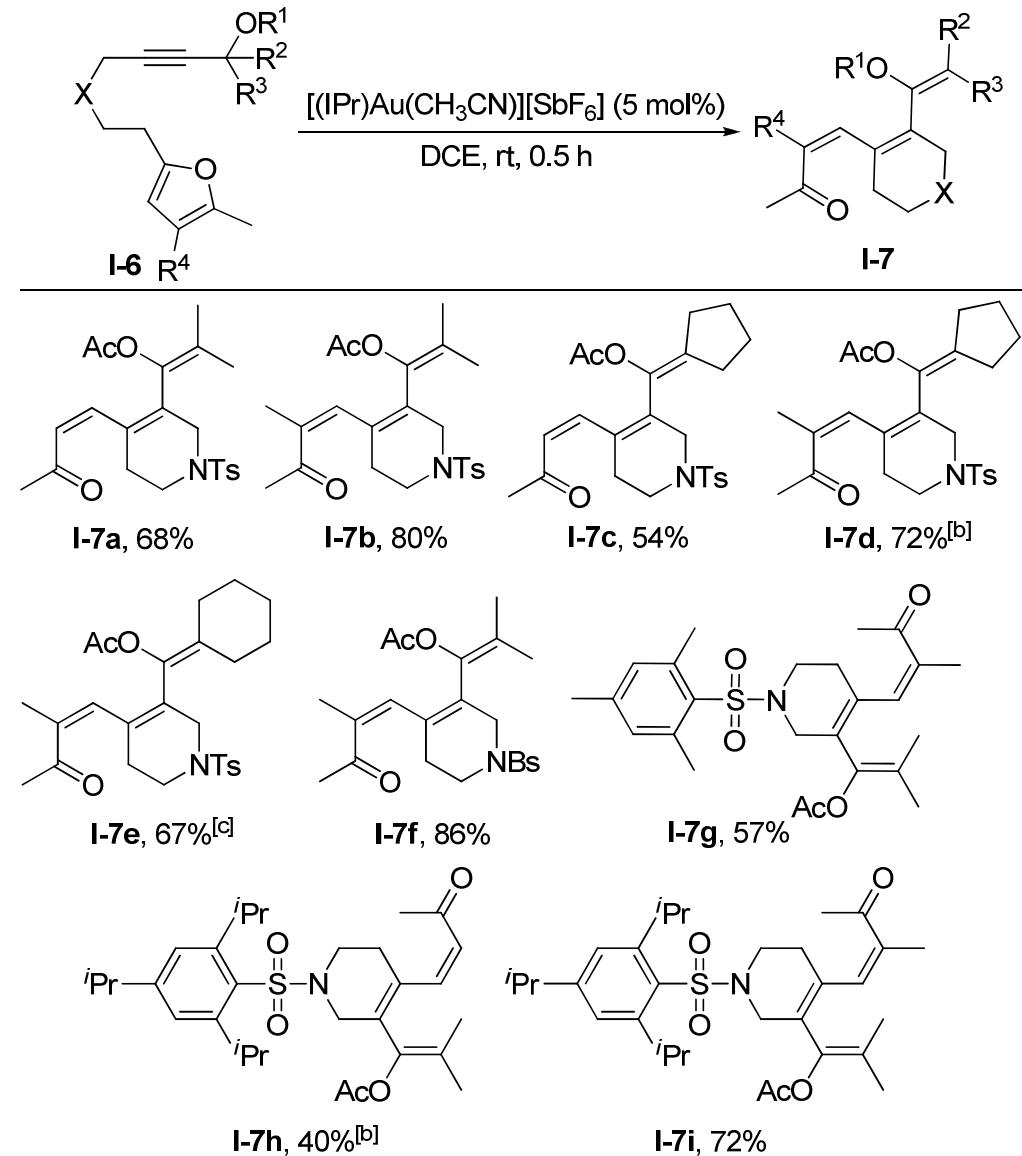
[a] Reaction conditions: **I-1** (0.2 mmol);  $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$  (5 mol%); anhydrous DCE (1.0 mL). Yields are those of the isolated products. [b] Mixture of *Z/E* isomers.

## 普适性考查



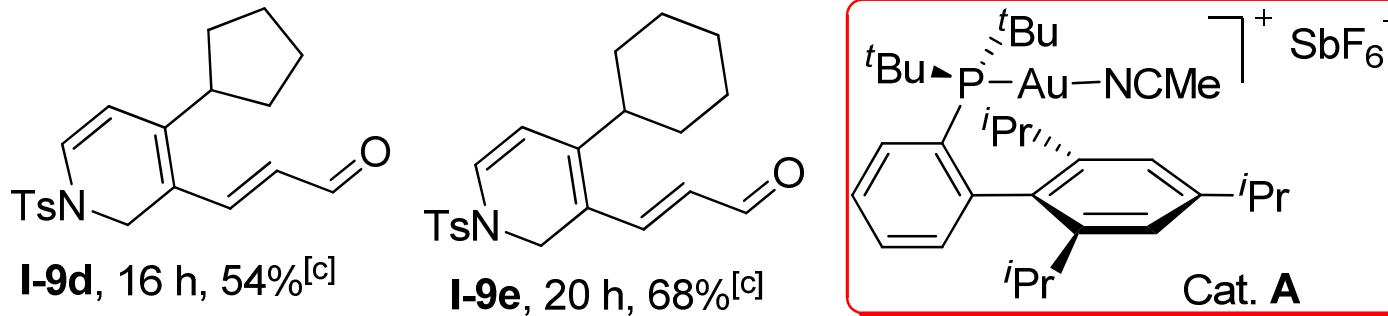
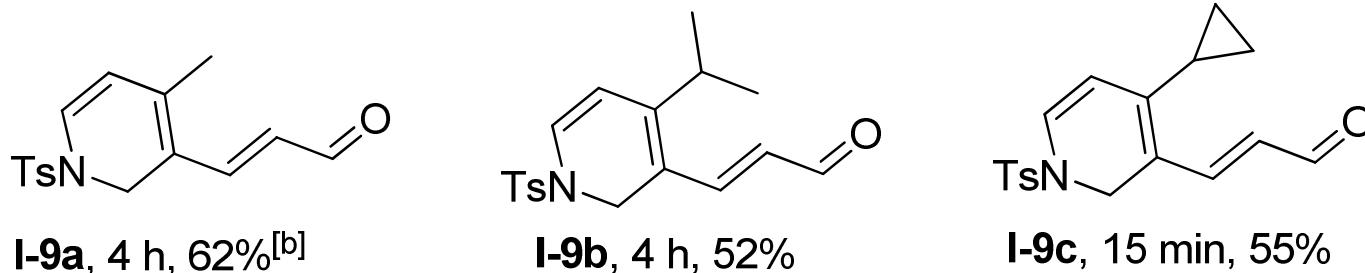
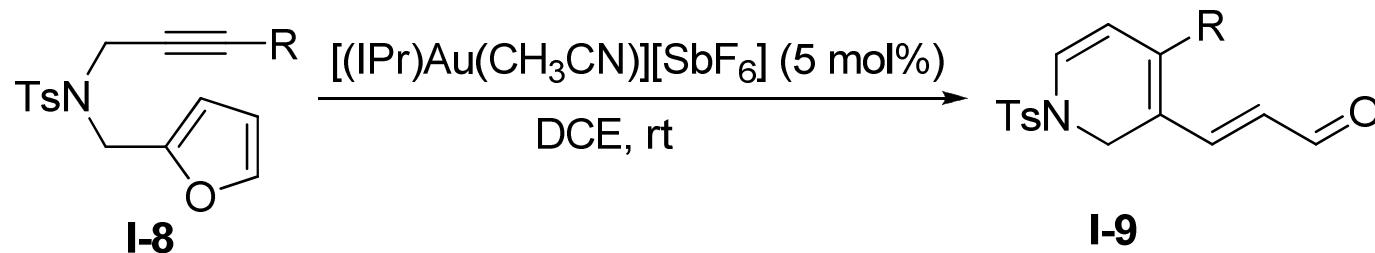
[a] Reaction conditions: I-4 (0.1-0.2 mmol); [(IPr)Au(CH<sub>3</sub>CN)][SbF<sub>6</sub>] (5 mol%); anhydrous DCE (1.0 mL). Yields are those of the isolated products. [b] Reaction performed at 80 °C.

# 普适性考查



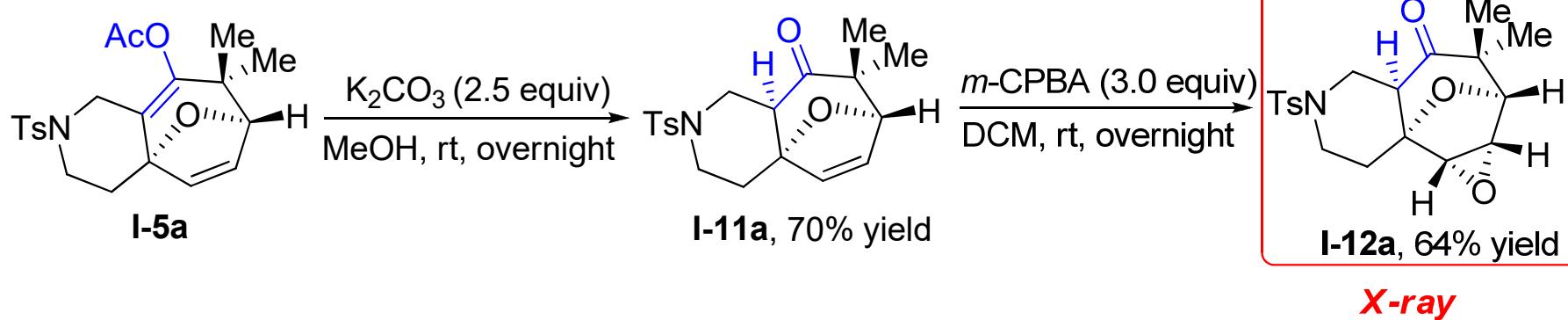
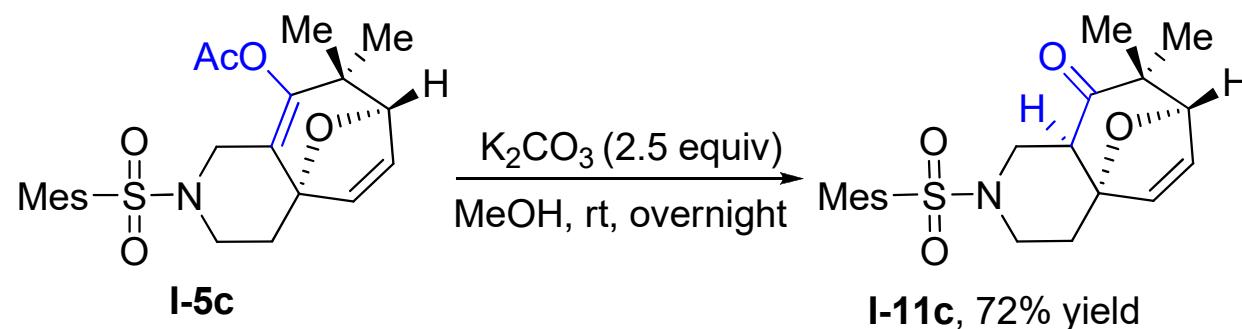
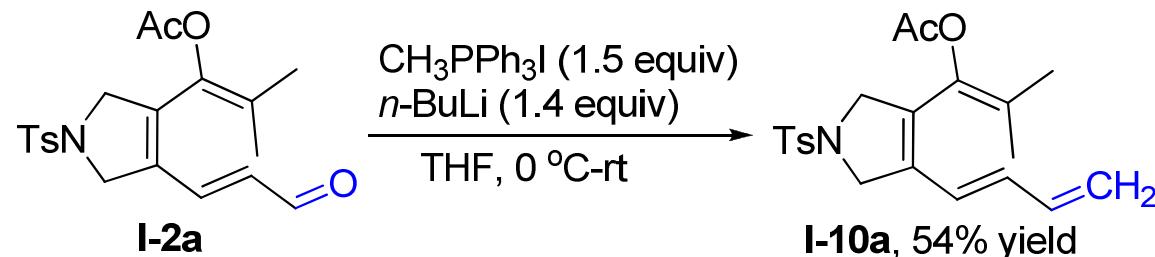
[a] Reaction conditions: **I-6** (0.1-0.2 mmol);  $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$  (5 mol%); anhydrous DCE (1.0 mL). Yields are those of the isolated products. [b] Reaction time: 1 h. [c] Reaction time: 2 h.

# 普适性考查

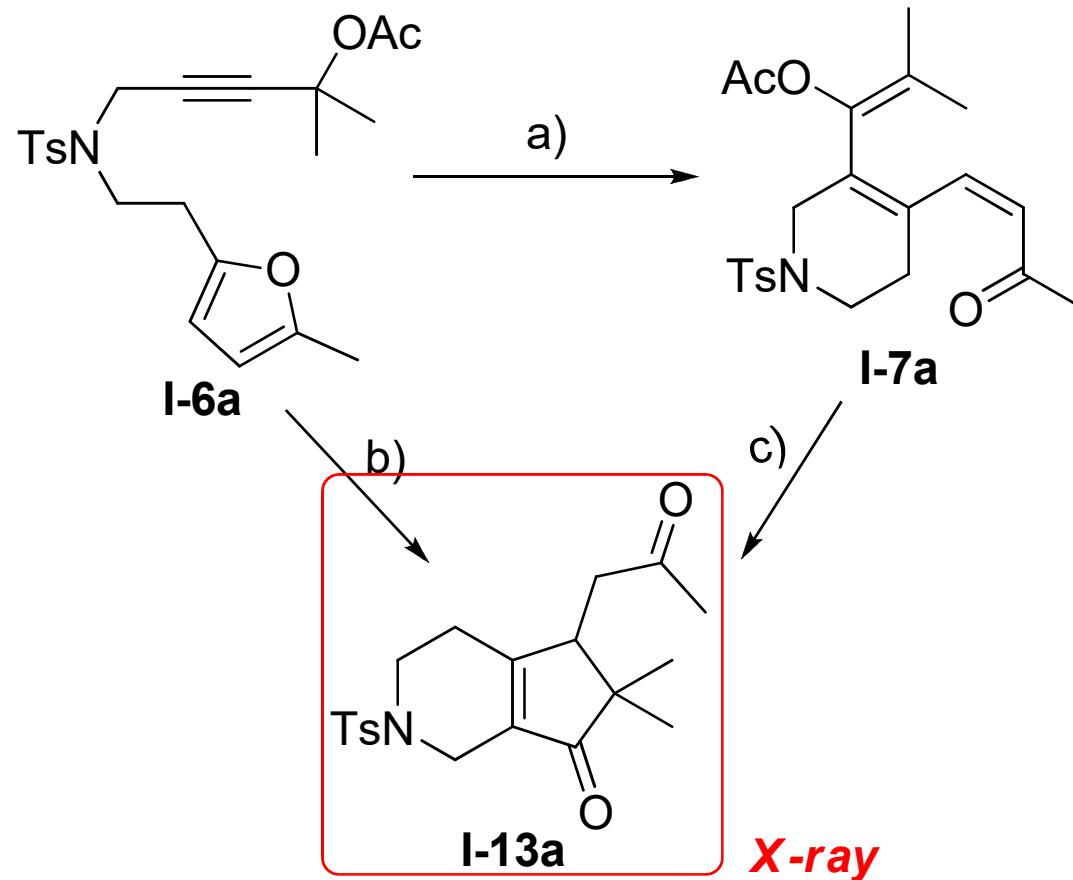


[a] Reaction conditions: **I-8** (0.1 mmol);  $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$  (5 mol%); anhydrous DCE (1.0 mL). Yields are those of the isolated products. [b] Reaction performed at 80 °C. [c] Cat. **A** was used.

## 产物转化



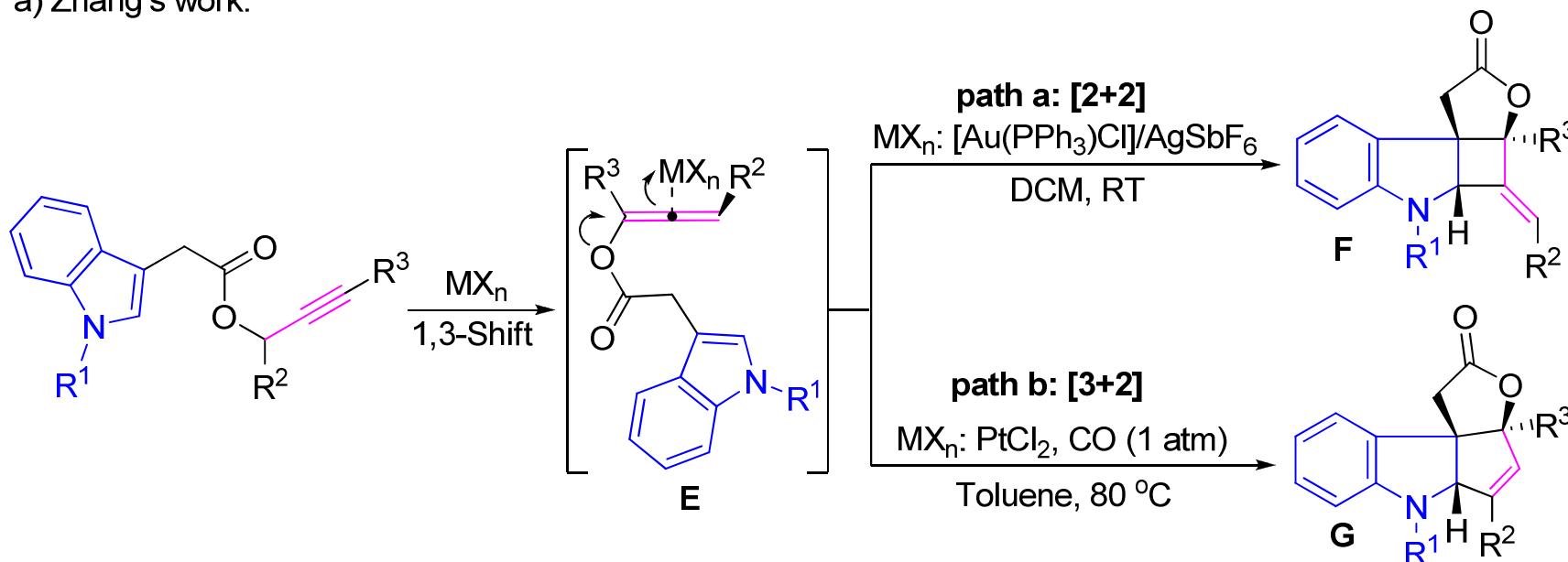
## 产物转化



- a)  $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$  (5 mol%), DCE, rt, 0.5 h, 68%;
- b)  $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$  (5 mol%), DCE, rt, 0.5 h;  
then  $\text{HOTf}$  (1.0 equiv), rt, 2 h, 80 %;
- c)  $\text{HOTf}$  (1.0 equiv), DCM, rt, 2 h, 91%.

# 金催化炔丙醇羧酸酯和吲哚分子内的环化反应

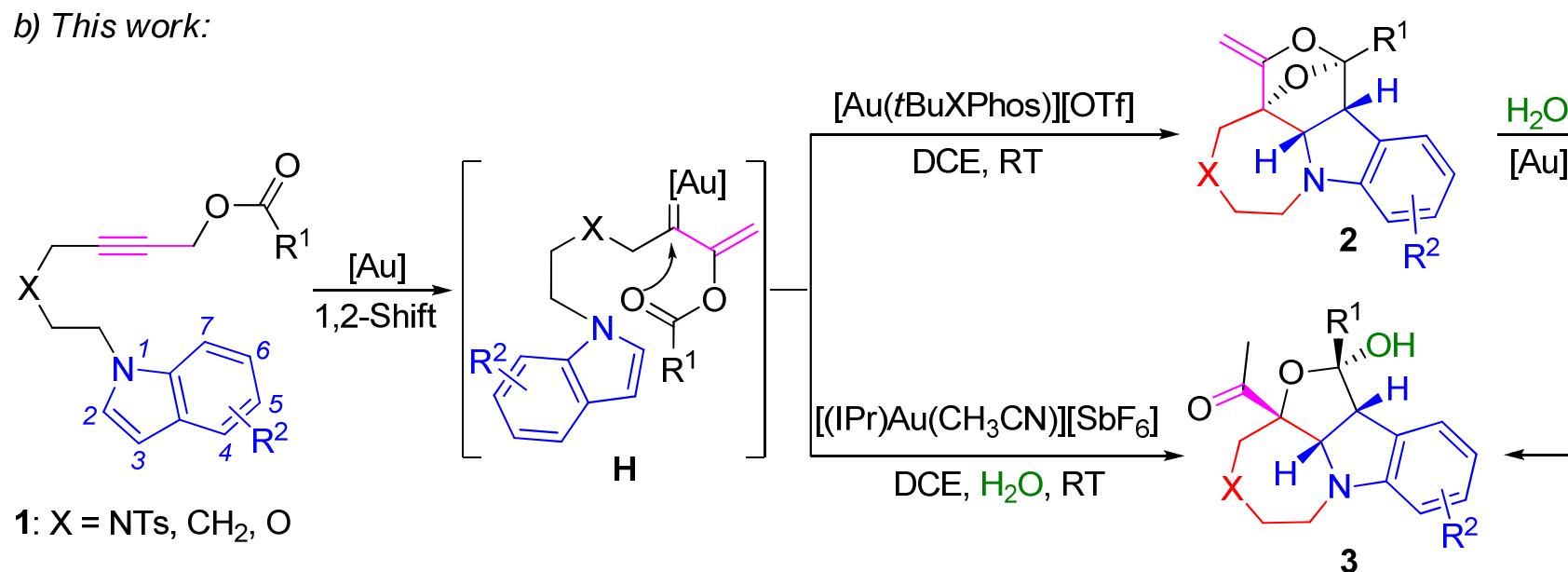
a) Zhang's work:



(a) Zhang, L. *J. Am. Chem. Soc.* **2005**, *127*, 16804-16805. (b) Zhang, G.; Catalano, V. J.; Zhang, L. *J. Am. Chem. Soc.* **2007**, *129*, 11358-11359.

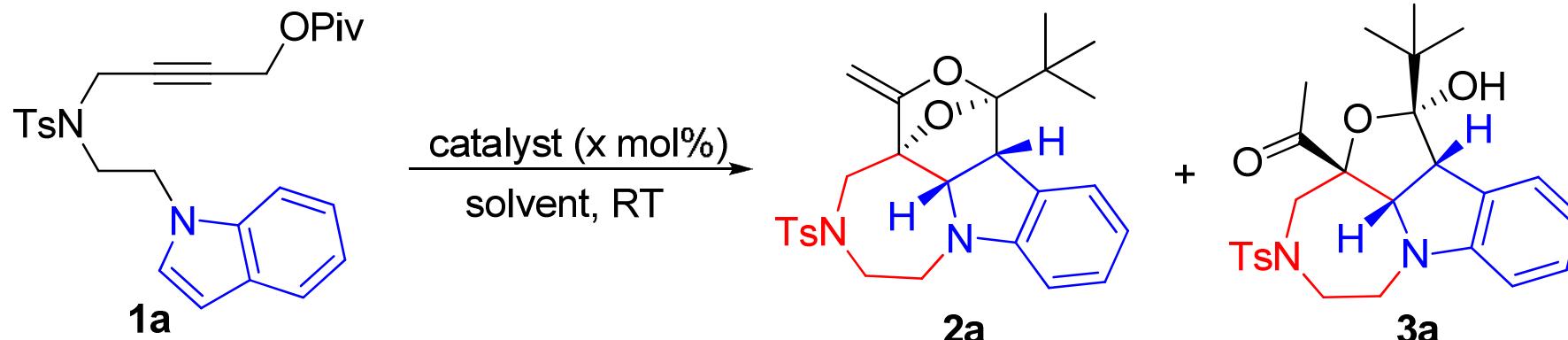
# 金催化炔丙醇羧酸酯和吲哚分子内的环化反应

b) This work:



Yang, J.-M.; Li, P.-H.; Wei, Y.; Tang, X.-Y.\*; Shi, M.\* *Chem. Commun.* **2016**, 52, 346-349.

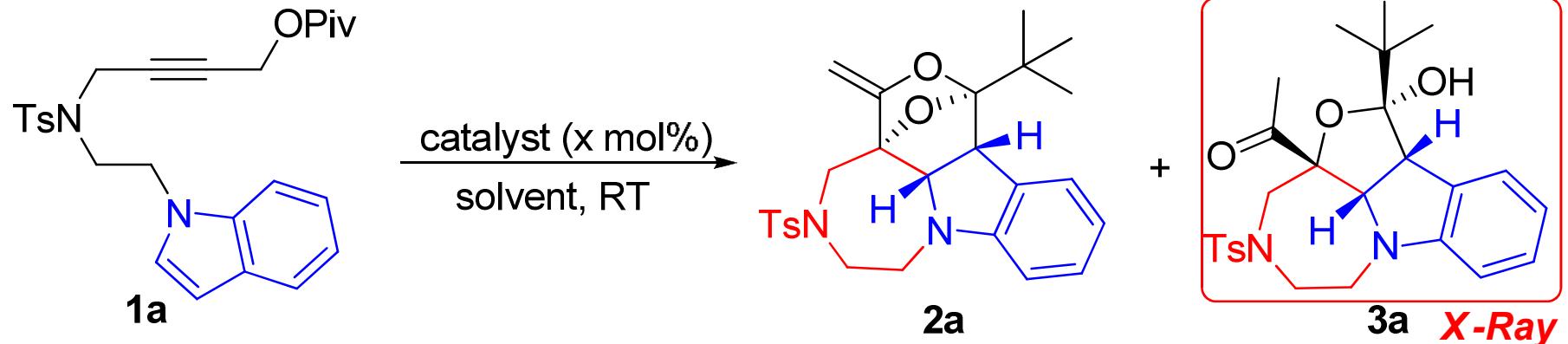
# 反应条件优化



entry <sup>[a]</sup>	catalyst (x mol%)	solvent	$\text{H}_2\text{O}$ (y eq.)	time	yield (%) <sup>[b]</sup>	
					<b>2a</b>	<b>3a</b>
<b>1</b>	<b><math>[\text{Au}(t\text{BuXPhos})][\text{OTf}]</math> (5)</b>	<b>DCE</b>	-	<b>2 h</b>	<b>95</b>	<b>0</b>
2	$[\text{Au}(\text{Me}_4t\text{BuXPhos})(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	-	6 h	0	0
3	$[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	-	2 h	0	70
4	$[\text{Au}(n\text{BuPAd}_2)(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	-	4 h	0	89
5	$[(\text{ArO})_3\text{PAu}][\text{NTf}_2]$ (5)	DCE	-	4 h	0	70
6	$[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	1.0	2 h	0	85
7	$[\text{Ph}_3\text{PAuCl}]$ (5)/ $\text{AgNTf}_2$ (5)	DCE	1.0	1 h	0	74
<b>8</b>	<b><math>[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]</math> (2.5)</b>	<b>DCE</b>	<b>1.0</b>	<b>5 h</b>	<b>0</b>	<b>86</b>

[a] All reactions were carried out using **1a** (0.1 mmol) in the presence of catalyst (x mol%) in DCE (1.0 mL) unless otherwise specified. [b] Isolated yields. Ar = 2,4-di-tert-butylphenyl. DCE = 1,2-dichloroethane.

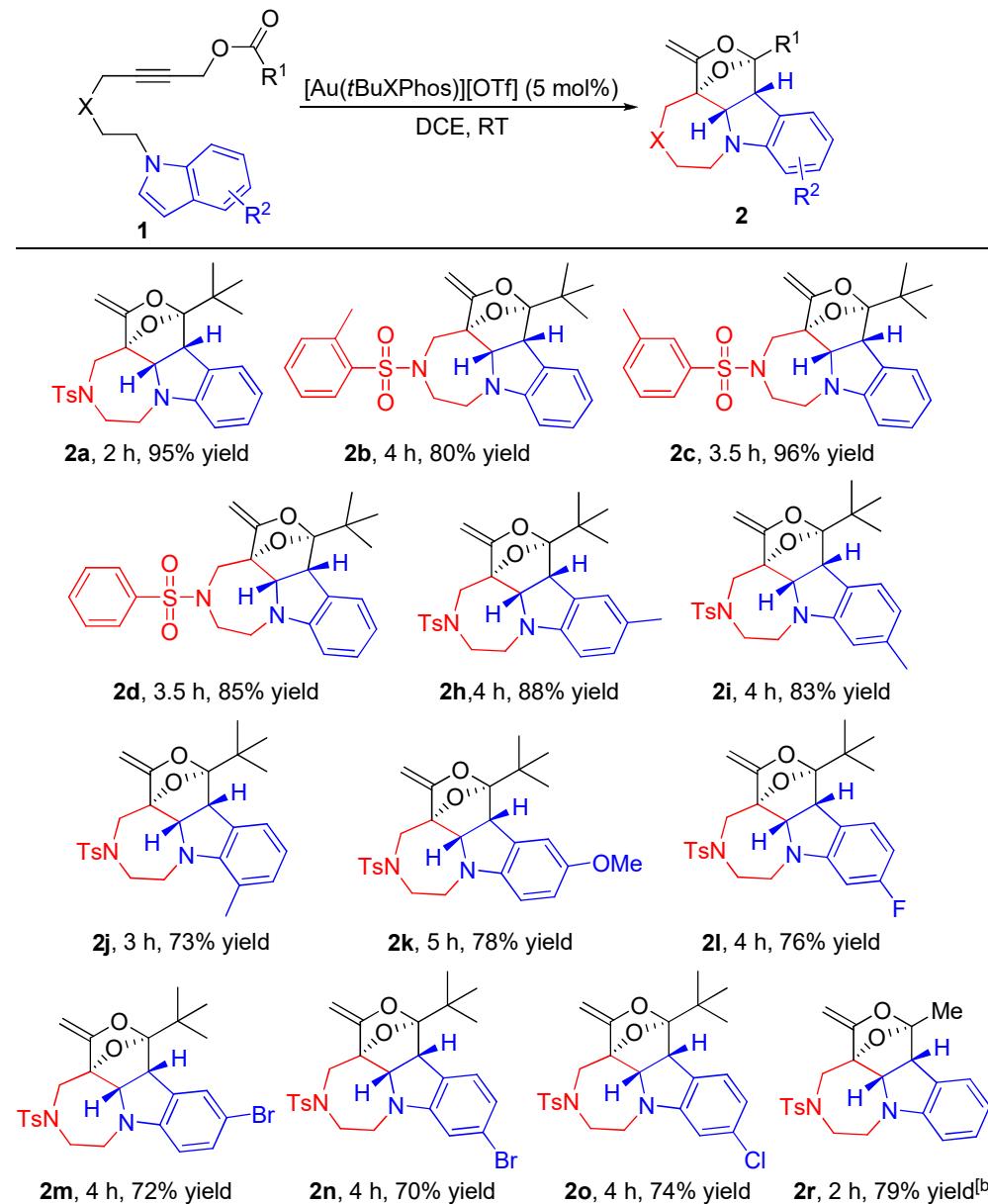
# 反应条件优化



entry <sup>[a]</sup>	catalyst (x mol%)	solvent	$\text{H}_2\text{O}$ (y eq.)	time	yield (%) <sup>[b]</sup>	
					2a	3a
1	<b><math>[\text{Au}(t\text{BuXPhos})][\text{OTf}]</math> (5)</b>	<b>DCE</b>	-	<b>2 h</b>	<b>95</b>	<b>0</b>
2	$[\text{Au}(\text{Me}_4t\text{BuXPhos})(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	-	6 h	0	0
3	$[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	-	2 h	0	70
4	$[\text{Au}(n\text{BuPAd}_2)(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	-	4 h	0	89
5	$[(\text{ArO})_3\text{PAu}][\text{NTf}_2]$ (5)	DCE	-	4 h	0	70
6	$[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]$ (5)	DCE	1.0	2 h	0	85
7	$[\text{Ph}_3\text{PAuCl}]$ (5)/ $\text{AgNTf}_2$ (5)	DCE	1.0	1 h	0	74
8	<b><math>[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})][\text{SbF}_6]</math> (2.5)</b>	<b>DCE</b>	<b>1.0</b>	<b>5 h</b>	<b>0</b>	<b>86</b>

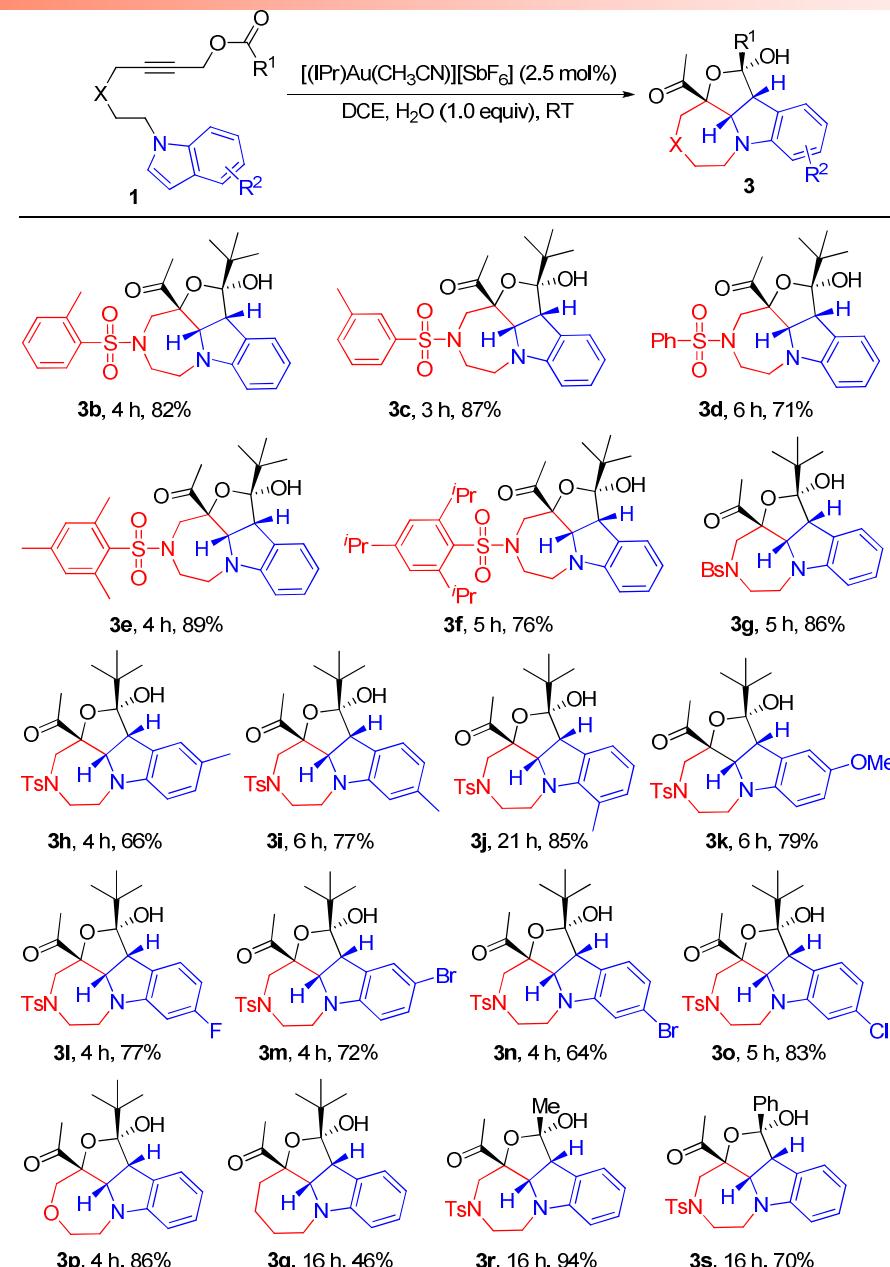
[a] All reactions were carried out using **1a** (0.1 mmol) in the presence of catalyst (x mol%) in DCE (1.0 mL) unless otherwise specified. [b] Isolated yields. Ar = 2,4-di-tert-butylphenyl. DCE = 1,2-dichloroethane.

# 普适性考查



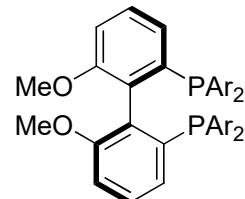
[a] Reaction conditions: **1** (0.1 mmol);  $[\text{Au}(t\text{BuXPhos})]\text{[OTf]} (5 \text{ mol}\%)$ ; anhydrous DCE (1.0 mL). Yields are those of the isolated yields. [b] 2.5 mol% of  $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})]\text{[SbF}_6]$  was used as the catalyst.

# 普适性考查

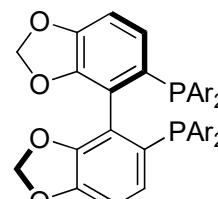


Reaction conditions: **1** (0.1 mmol);  $[(\text{IPr})\text{Au}(\text{CH}_3\text{CN})]\text{[SbF}_6\text{]} (2.5 \text{ mol\%})$ ;  $\text{H}_2\text{O}$  (1.0 equiv); anhydrous DCE (1.0 mL). Yields are those of the isolated yields. DCE = 1,2-dichloroethane.

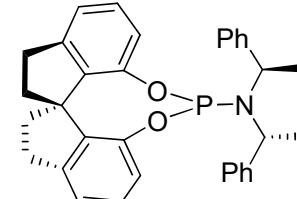
# 不对称优化条件使用的配体



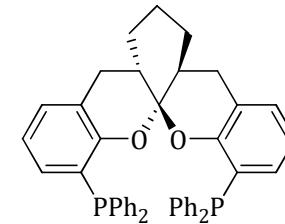
Ar = 3,5-(tBu)<sub>2</sub>-4-MeOC<sub>6</sub>H<sub>2</sub>;  
(R)-MeO-DTBM-BIPHEP (**L1**)



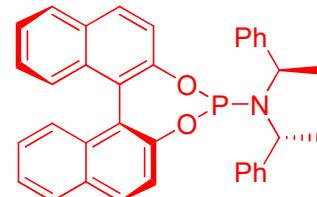
Ar = 3,5-(Me)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>;  
(R)-DM-SEGPHOS (**L2**)



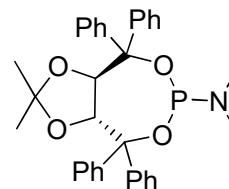
**L3**



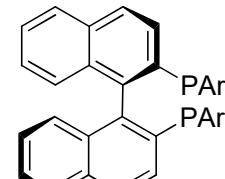
**L4**



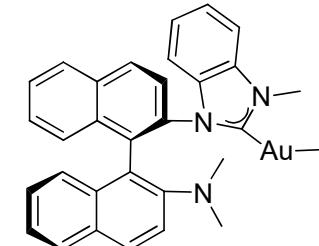
**L5**



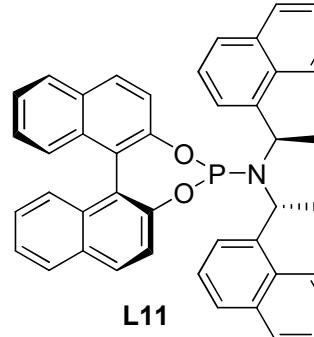
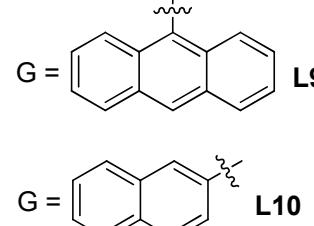
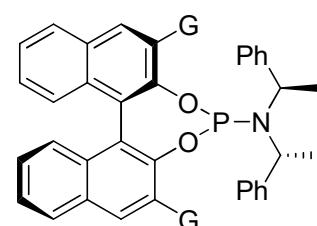
**L6**



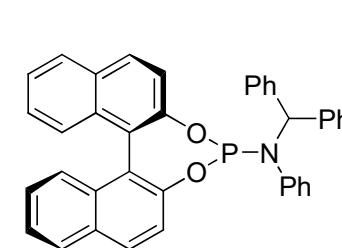
Ar = 3,5-(Me)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>;  
(R)-Xylyl-BINAP (**L7**)



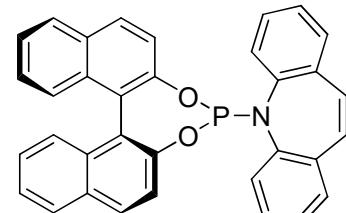
(aR)-8



**L11**

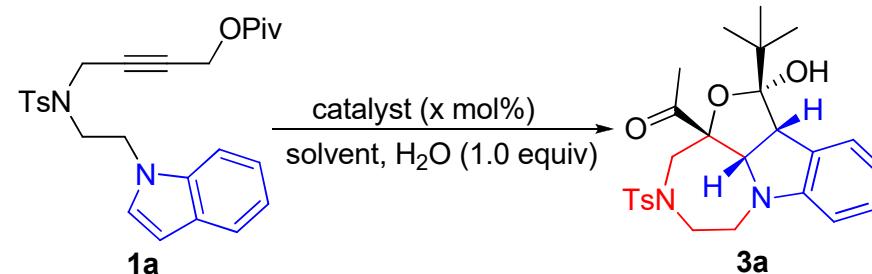


**L12**



**L13**

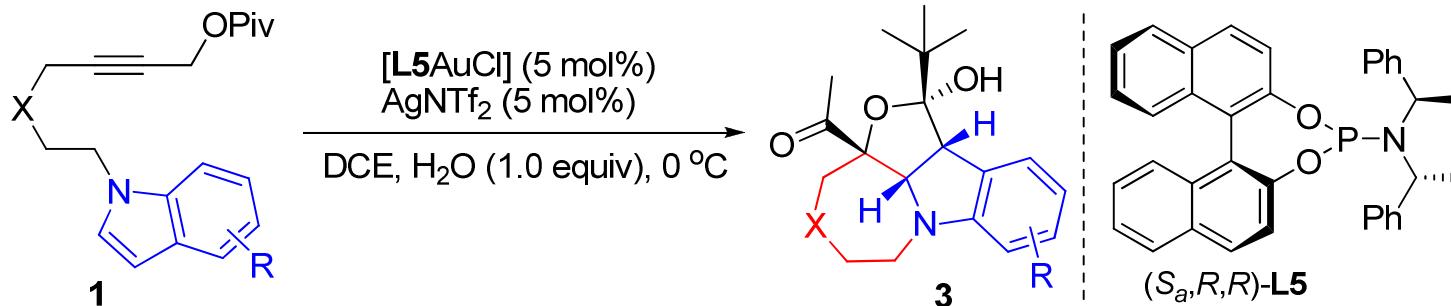
# 反应条件优化



entry <sup>[a]</sup>	catalyst (x mol%)	solvent	T (°C)	time	yield (%) <sup>[b]</sup>	ee (%) <sup>[c]</sup>
1	[L1AuCl] (5)/AgSbF <sub>6</sub> (5)	DCE	RT	1 h	74	0
2	[L2Au <sub>2</sub> Cl <sub>2</sub> ] (5)/AgOPNB (5)	DCE	RT	2 d	trace	-[d]
3	[L2Au <sub>2</sub> Cl <sub>2</sub> ] (5)/AgSbF <sub>6</sub> (5)	DCE	RT	2 h	72	0
4	[L3Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ] (5)	DCE	RT	3 h	85	40
5	[L4Au <sub>2</sub> (CH <sub>3</sub> CN) <sub>2</sub> ][(SbF <sub>6</sub> ) <sub>2</sub> ] (5)	DCE	RT	3 d	67	15
6	[L5Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ] (5)	DCE	RT	1 h	86	71
7	[L5AuCl] (5)/AgNTf <sub>2</sub> (5)	DCE	RT	3.5 h	79	71
8	[L5AuCl] (5)/AgNTf <sub>2</sub> (5)	Toluene	RT	4 h	23	47
9	[L5Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ] (5)	DCM	RT	1 h	87	67
10	[L5Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ] (5)	CHCl <sub>3</sub>	RT	5 h	85	50
11	[L5AuCl] (5)/AgBF <sub>4</sub> (5)	DCE	RT	45 min	65	31
12	[L5AuCl] (5)/AgSbF <sub>6</sub> (5)	DCE	RT	4 h	77	14
13	[L5AuCl] (5)/AgOTf (5)	DCE	RT	4 h	trace	-[d]
14	[L5AuCl] (5)/AgOONB (5)	DCE	RT	4 d	trace	-[d]
15	[L6Au(CH <sub>3</sub> CN)][SbF <sub>6</sub> ] (5)	DCE	RT	6 h	55	-7
16	[L7Au <sub>2</sub> Cl <sub>2</sub> ] (5)/AgSbF <sub>6</sub> (10)	DCE	RT	5 h	67	2
17	(aR)-8 (5)/AgSbF <sub>6</sub> (5)	DCE	RT	30 min	79	0
18	[L9AuCl] (5)/AgNTf <sub>2</sub> (5)	DCE	RT	4 h	78	-17
19	[L10AuCl] (5)/AgNTf <sub>2</sub> (5)	DCE	RT	22 h	77	43
20	[L11AuCl] (5)/AgNTf <sub>2</sub> (5)	DCE	RT	2 h	84	52
21	[L12AuCl] (5)/AgNTf <sub>2</sub> (5)	DCE	RT	18 h	69	-46
22	[L13AuCl] (5)/AgNTf <sub>2</sub> (5)	DCE	RT	1.5 h	72	27
<b>23</b>	<b>[L5AuCl] (5)/AgNTf<sub>2</sub> (5)</b>	<b>DCE</b>	<b>0</b>	<b>16 h</b>	<b>72</b>	<b>77</b>

[a] All reactions were carried out using **1a** (0.1 mmol) in the presence of catalyst (x mol%) in various solvents (1.0 mL) unless otherwise specified. [b] Yield of isolated product. [c] Determined by HPLC on a chiral stationary phase. [d] Not determined.  
OPNB = *p*-nitrobenzoate, OONB = *o*-nitrobenzoate

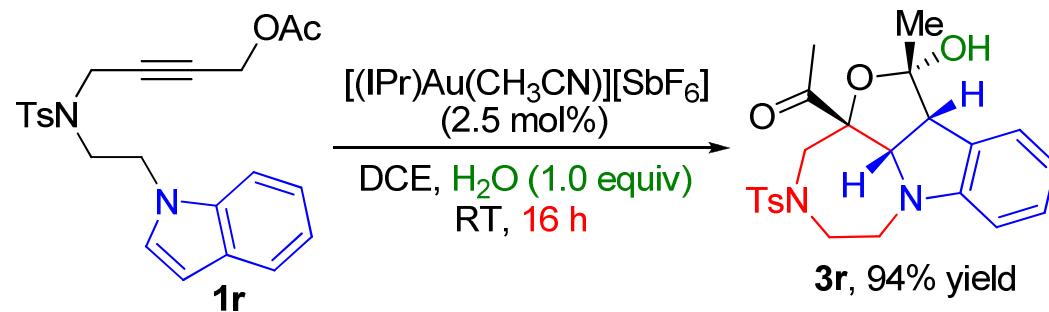
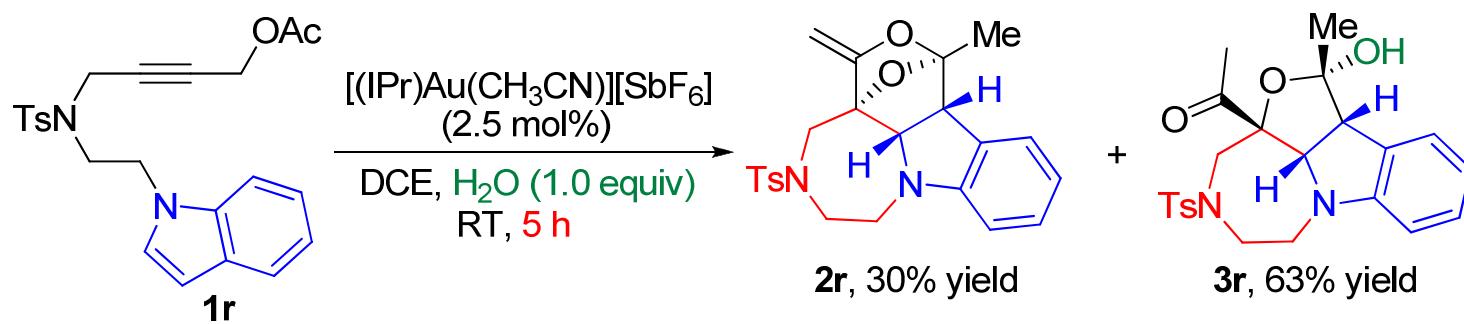
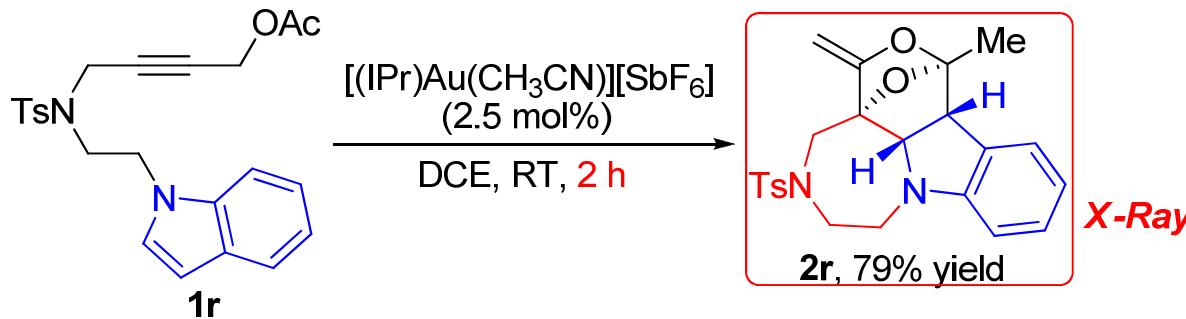
# 普适性考查



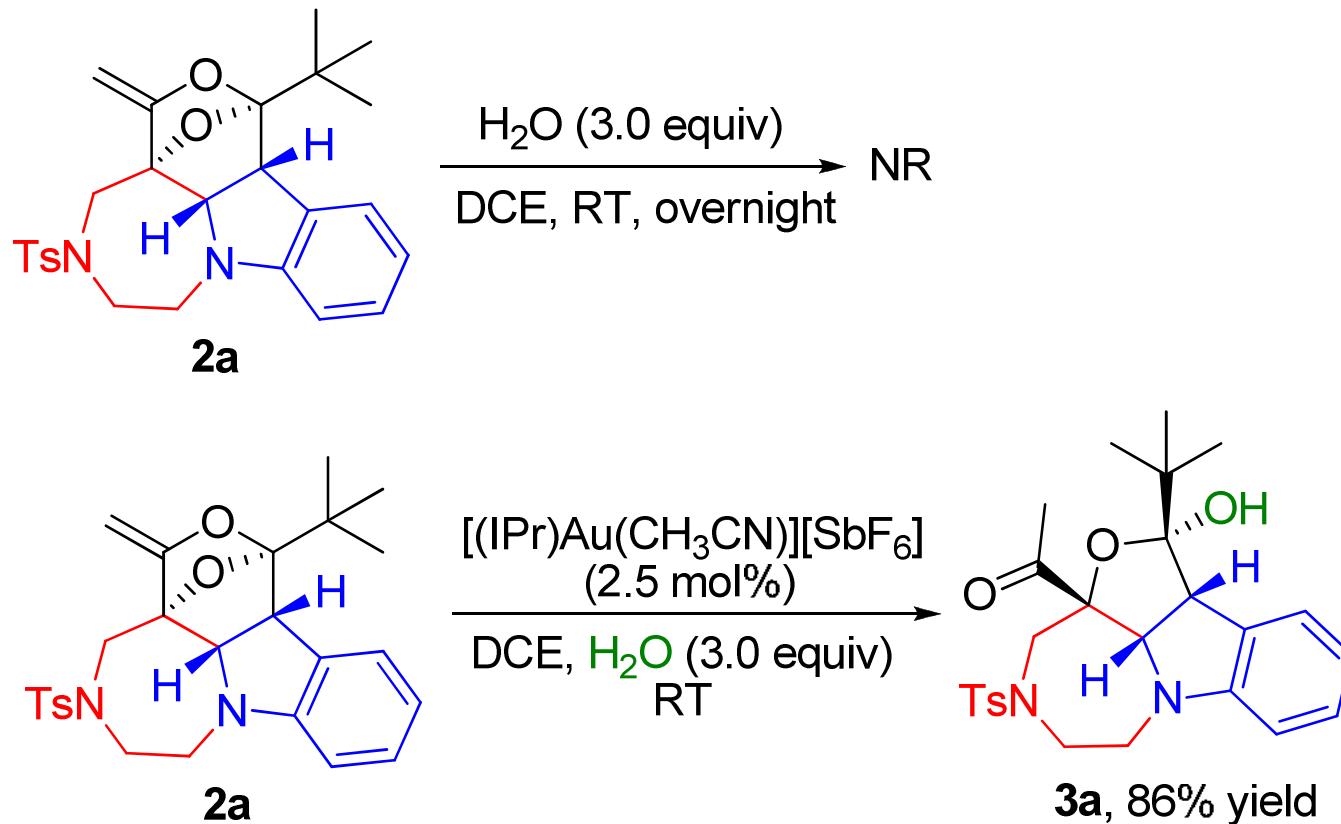
entry <sup>[a]</sup>	1	X	R	time [days]	3	yield [%] <sup>[b]</sup>	ee [%] <sup>[c]</sup>
1	<b>1b</b>	2-MeC <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> N	H	3	<b>3b</b>	30	74
2	<b>1c</b>	3-MeC <sub>6</sub> H <sub>4</sub> SO <sub>2</sub> N	H	2	<b>3c</b>	60	80
3	<b>1d</b>	PhSO <sub>2</sub> N	H	2	<b>3d</b>	43	90
4	<b>1e</b>	MesSO <sub>2</sub> N	H	2	<b>3e</b>	51	83
5	<b>1f</b>	2,4,6-iPr <sub>3</sub> C <sub>6</sub> H <sub>2</sub> SO <sub>2</sub> N	H	2	<b>3f</b>	40	91
6	<b>1h</b>	TsN	5-Me	2	<b>3h</b>	50	74
7	<b>1i</b>	TsN	6-Me	0.5	<b>3i</b>	46	71
8	<b>1j</b>	TsN	7-Me	2	<b>3j</b>	88	81
9	<b>1k</b>	TsN	5-OMe	2	<b>3k</b>	62	82
10	<b>1l</b>	TsN	6-F	0.5	<b>3l</b>	72	82
11 <sup>[d]</sup>	<b>1m</b>	TsN	5-Br	2	<b>3m</b>	66	62
12	<b>1n</b>	TsN	6-Br	1	<b>3n</b>	85	82
13	<b>1o</b>	TsN	6-Cl	3		[e]	[e]
14	<b>1p</b>	O	H	3	<b>3p</b>	30	72
15	<b>1q</b>	CH <sub>2</sub>	H	3	<b>3q</b>	25	60

[a] Reaction conditions: **1** (0.1 mmol), **[L5AuCl]** (5 mol%), AgNTf<sub>2</sub> (5 mol%), H<sub>2</sub>O (1.0 equiv), anhydrous DCE (1.0 mL). [b] Yields are those of the isolated yields. [c] Determined by HPLC on a chiral stationary phase. [d] Reaction performed at room temperature. [e] Complex mixtures, not determined. DCE=1,2-dichloroethane.

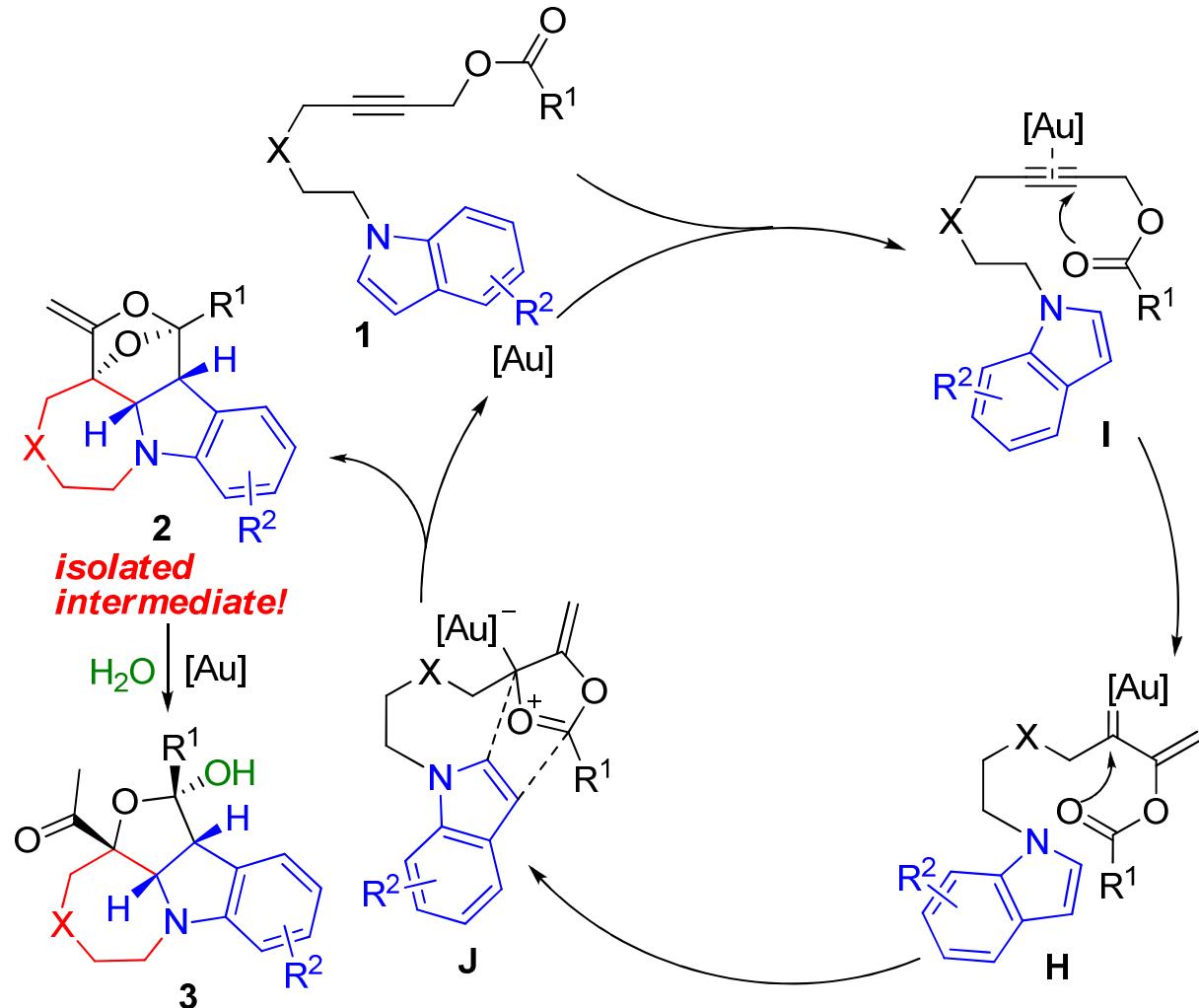
## 控制实验



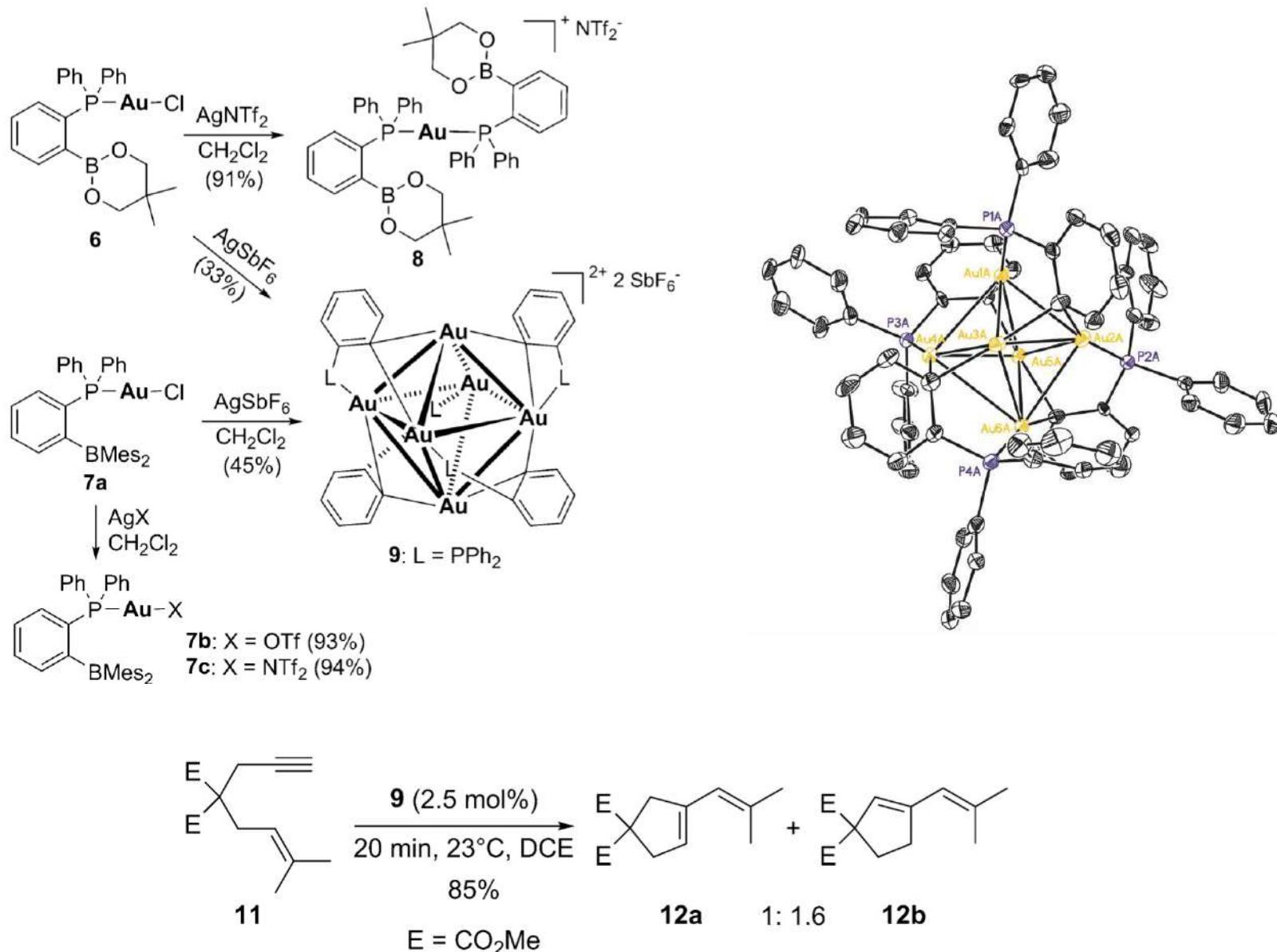
## 控制实验

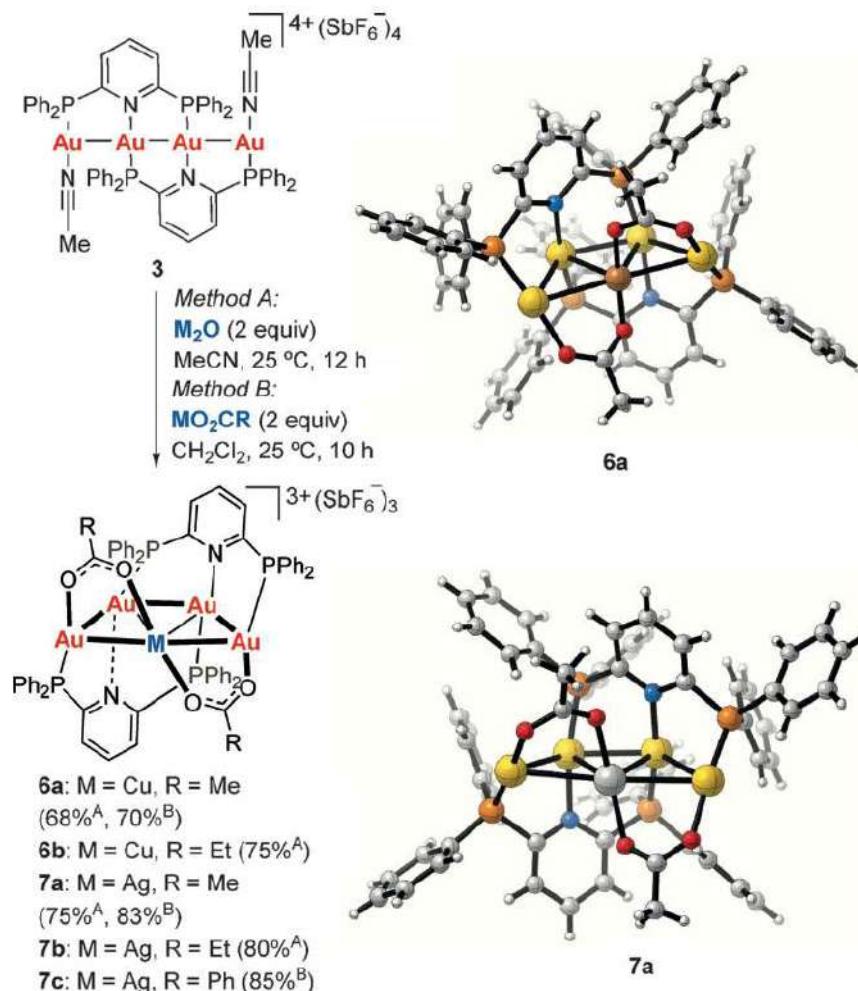
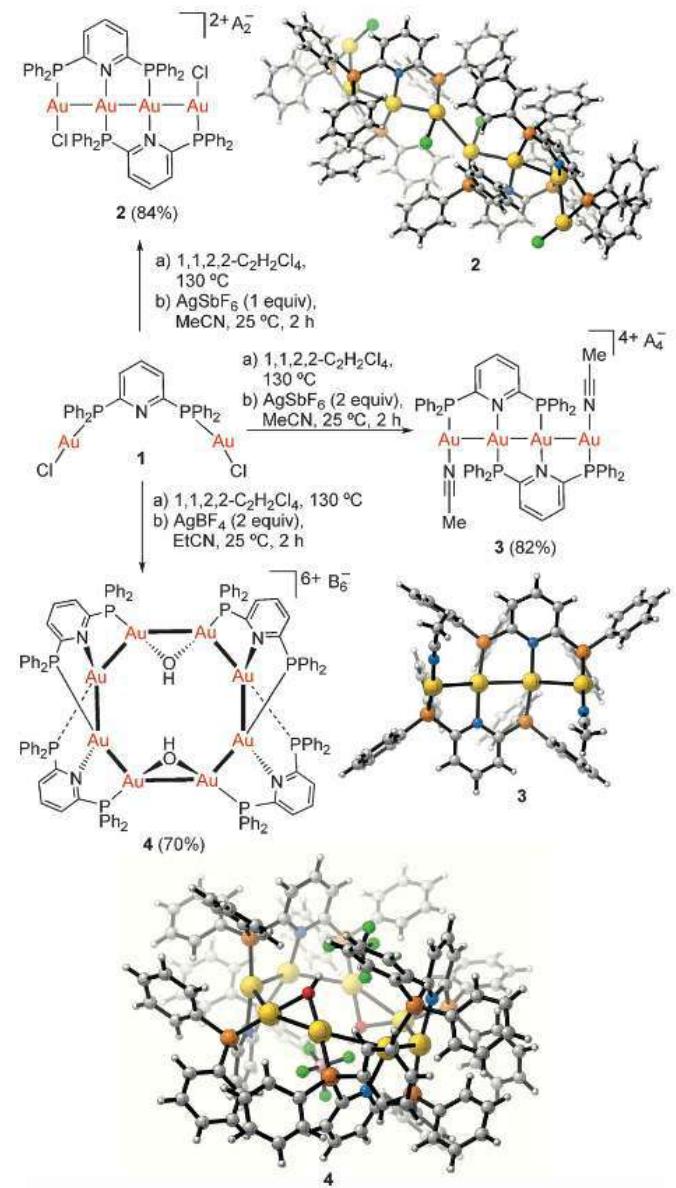


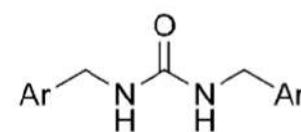
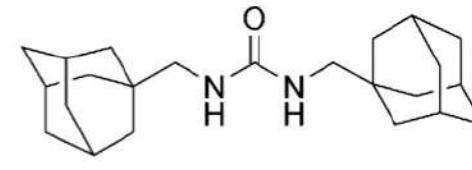
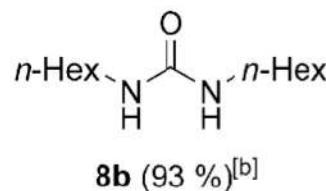
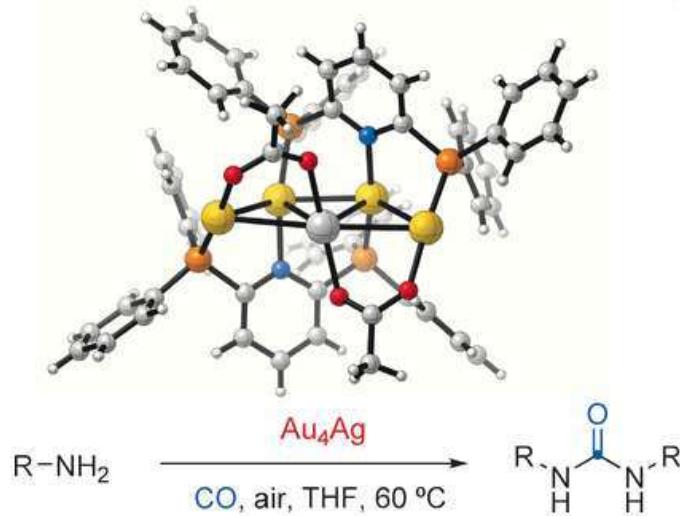
# 反应机理



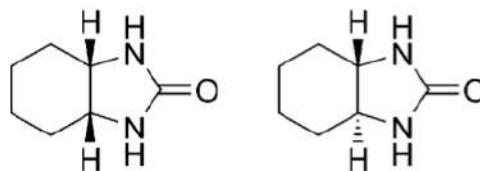
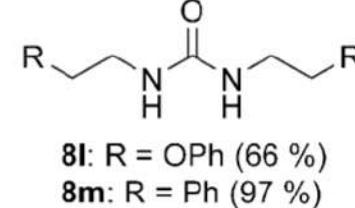
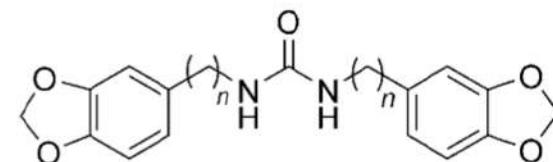
### 三、多金簇合物的合成、表征以及催化活性研究



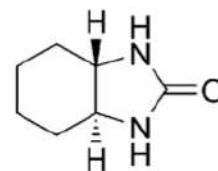




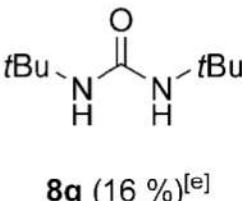
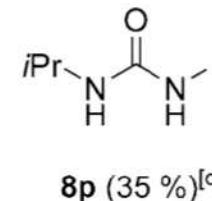
- 8d:** Ar = Ph ( $97\%$ )  
**8e:** Ar = *p*-MeOC<sub>6</sub>H<sub>4</sub> ( $55\%$ )  
**8f:** Ar = *p*-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub> ( $61\%$ )  
**8g:** Ar = *o*-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub> ( $98\%$ )  
**8h:** Ar = *o*-FC<sub>6</sub>H<sub>4</sub> ( $97\%$ )  
**8i:** Ar = 2-Furyl ( $93\%$ )



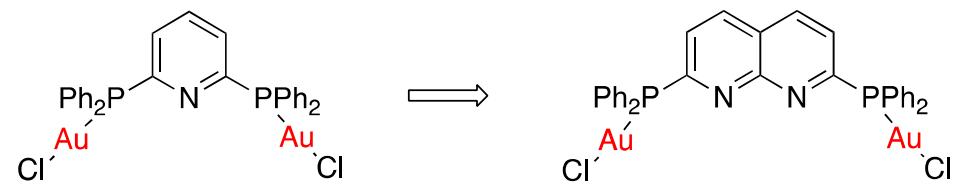
**8n** ( $89\%$ )



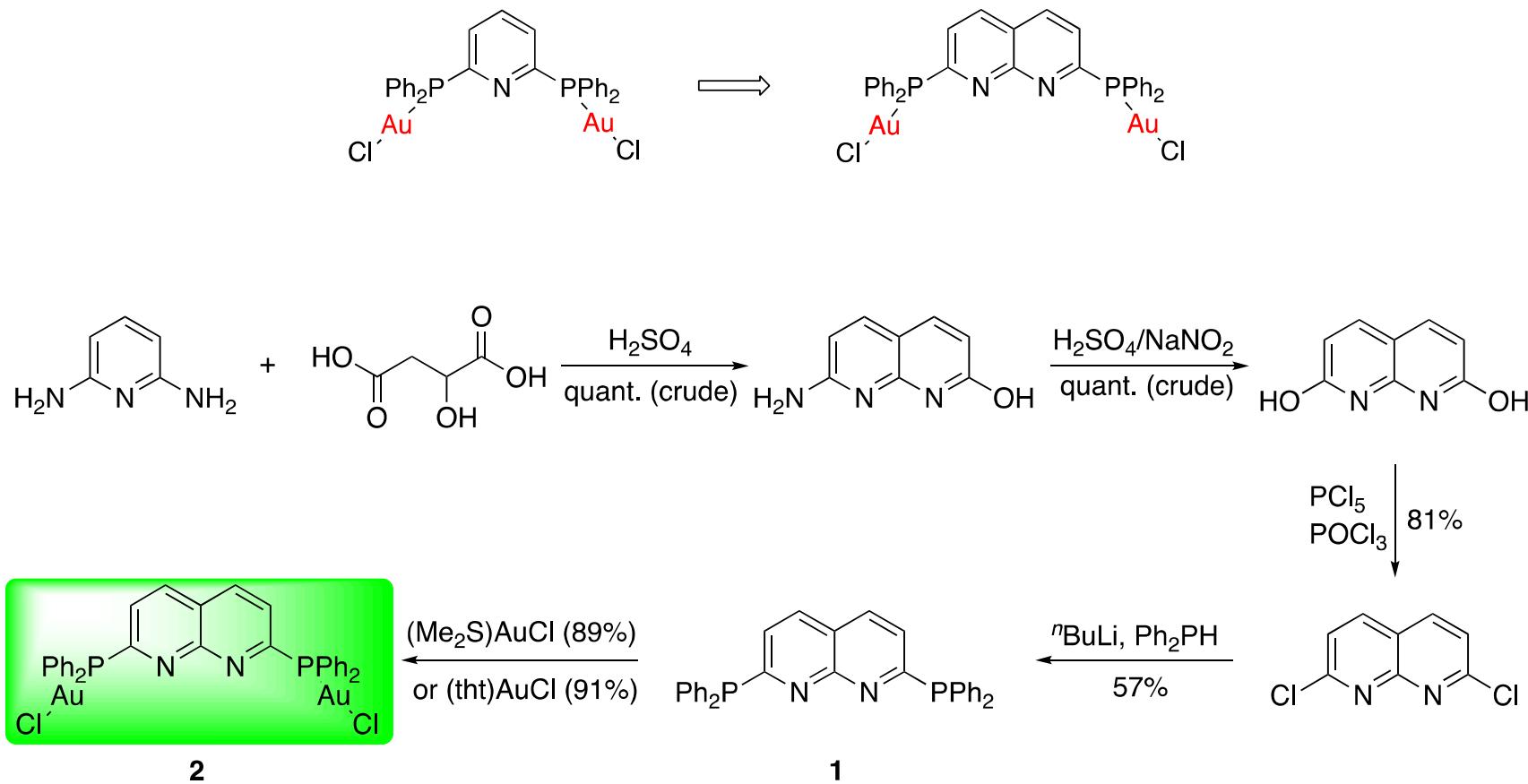
**8o** ( $92\%$ )



# PNNP类型配体的合成

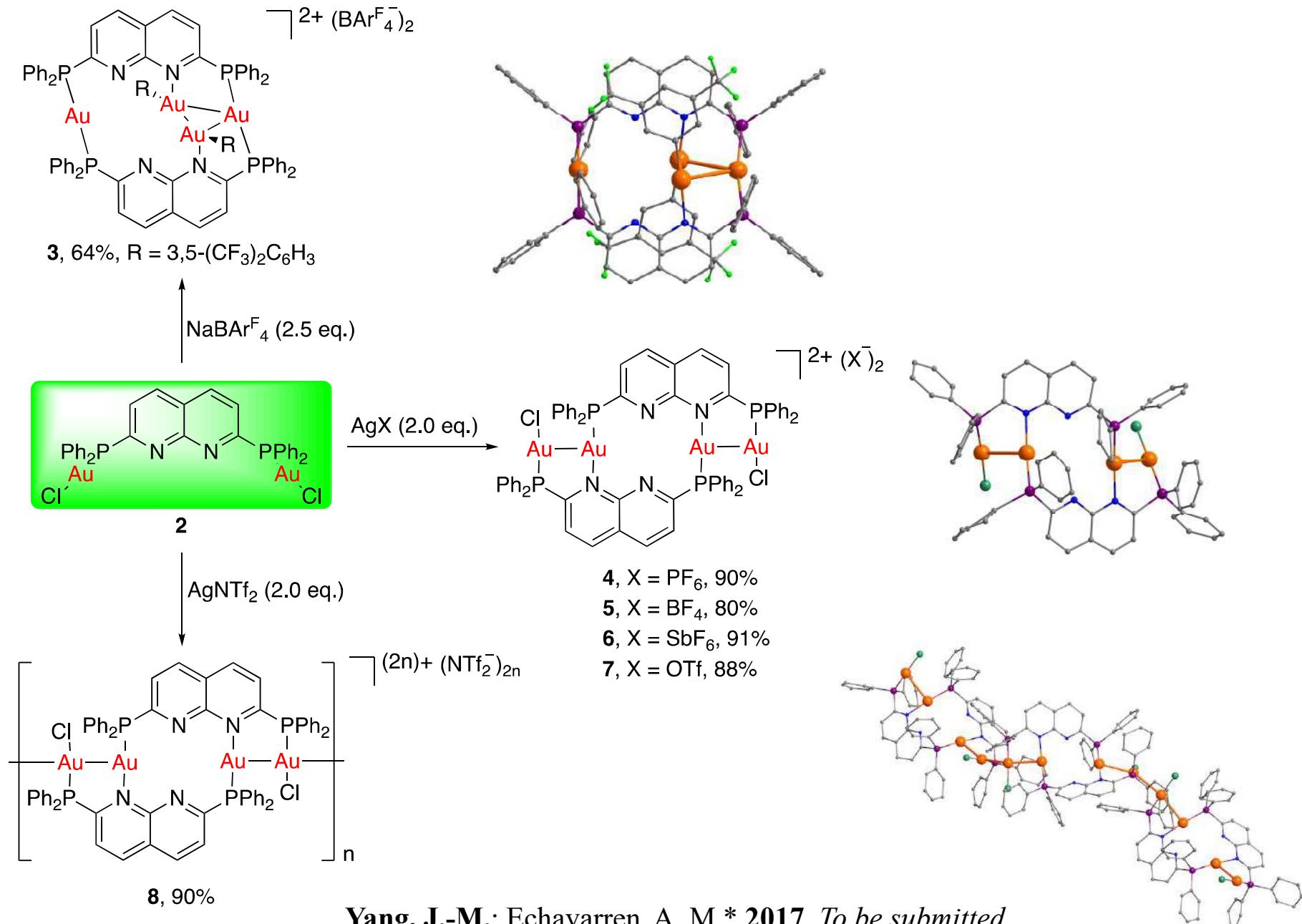


# PNNP类型配体的合成

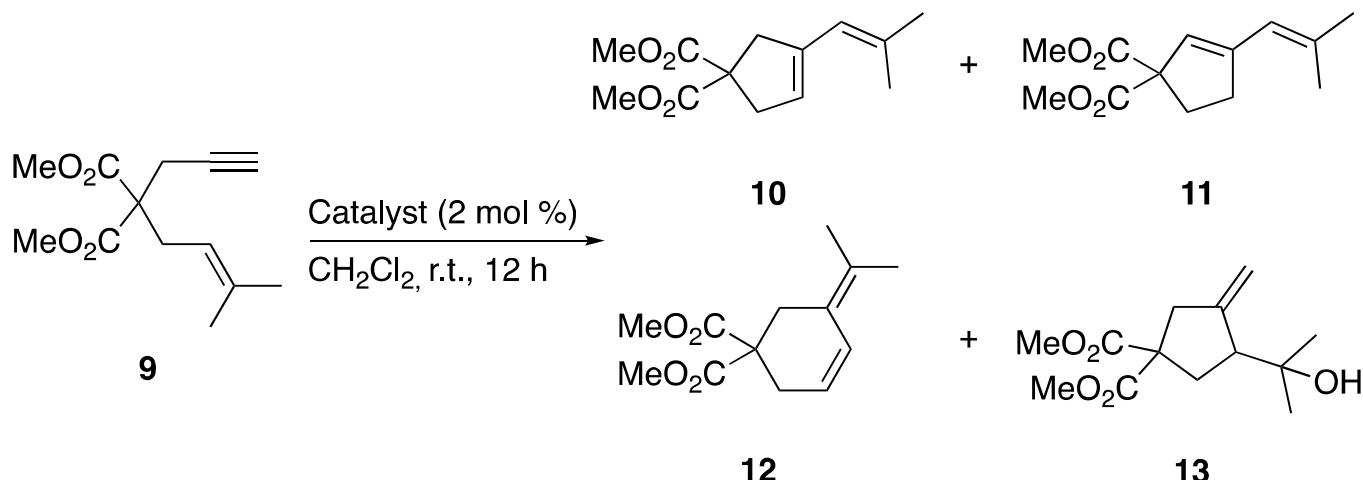


(a) Ziessel, R. *Tetrahedron Lett.* **1989**, *30*, 463-466. (b) Catalano, V. J.; Kar, H. M.; Bennett, B. L. *Inorg. Chem.* **2000**, *39*, 121-127.

# 多金簇合物的合成与表征



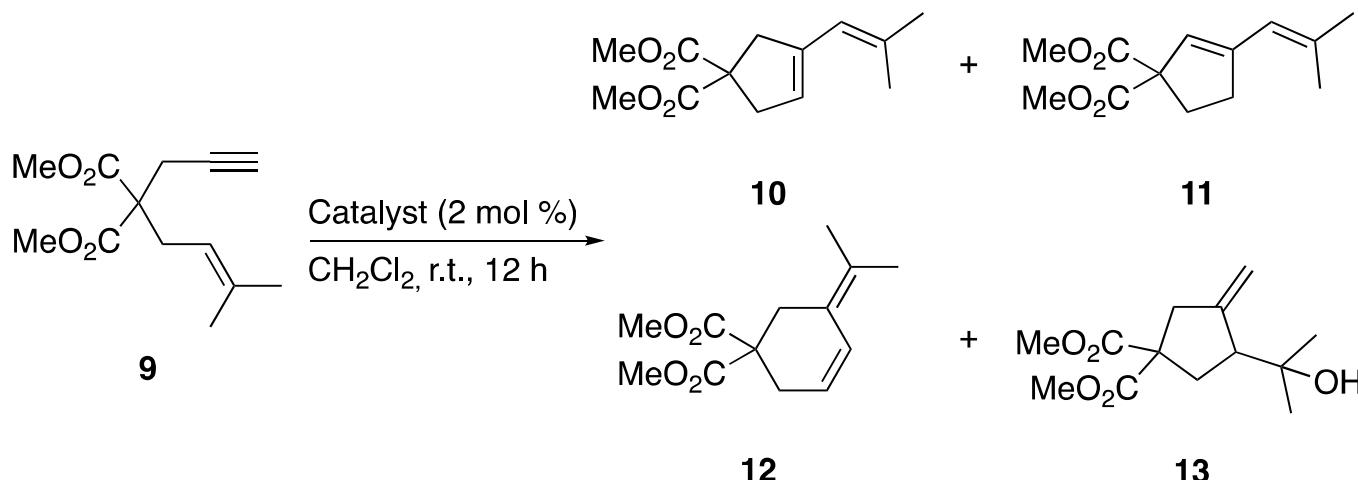
## 催化活性研究



Entry <sup>[a]</sup>	Catalyst	Conv. [%] <sup>[b]</sup>	10	11	12	13
1	<b>3, L<sub>2</sub>Au<sub>4</sub>(BAr<sup>F</sup><sub>4</sub>)<sub>4</sub></b>	100	85 (83)	0	10	1
2 <sup>[c]</sup>	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	35	21	0	5	1
3	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	100	58	0	18	0
4	<b>5, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(BF<sub>4</sub>)<sub>2</sub></b>	42	28	0	3	9
5	<b>6, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(SbF<sub>6</sub>)<sub>2</sub></b>	100	74	0	17	4
6	<b>7, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(OTf)<sub>2</sub></b>	100	22	0	23	0
7	<b>8, [L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(NTf<sub>2</sub>)<sub>2</sub>]<sub>n</sub></b>	100	14	27	29	0

[a] Reaction conditions : **9** (0.1 mmol), cat. (2 mol %),  $\text{CH}_2\text{Cl}_2$  (1.0 mL). [b] Conversion and yields were determined by  $^1\text{H}$  NMR spectroscopy using 1,4-diacetylbenzene as internal standard. Value within parentheses is that of the yield of the isolated product after column chromatography. [c] 1 mol % of **4**, reaction time: 3 h. L = 2,9-bis(diphenylphosphino)-1,8-naphthyridine (dppn).

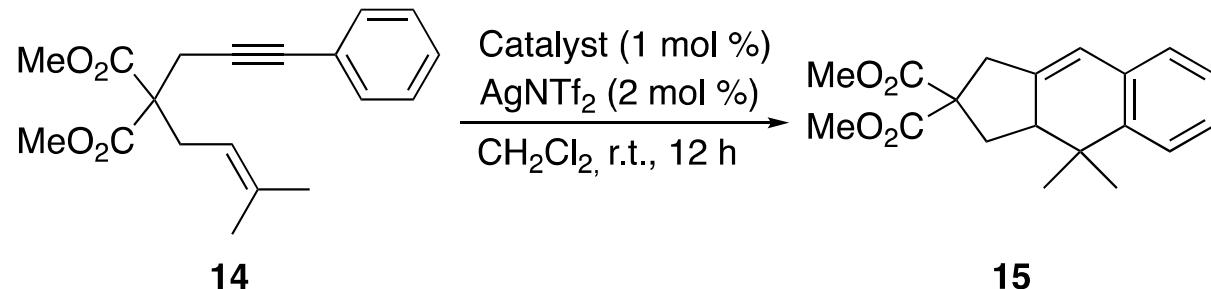
## 催化活性研究



Entry <sup>[a]</sup>	Catalyst	Conv. [%] <sup>[b]</sup>	10	11	12	13
1	<b>3, L<sub>2</sub>Au<sub>4</sub>(BAr<sup>F</sup><sub>4</sub>)<sub>4</sub></b>	100	85 (83)	0	10	1
2 <sup>[c]</sup>	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	35	21	0	5	1
3	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	100	58	0	18	0
4	<b>5, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(BF<sub>4</sub>)<sub>2</sub></b>	42	28	0	3	9
5	<b>6, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(SbF<sub>6</sub>)<sub>2</sub></b>	100	74	0	17	4
6	<b>7, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(OTf)<sub>2</sub></b>	100	22	0	23	0
7	<b>8, [L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(NTf<sub>2</sub>)<sub>2</sub>]<sub>n</sub></b>	100	14	27	29	0

[a] Reaction conditions : **9** (0.1 mmol), cat. (2 mol %), CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL). [b] Conversion and yields were determined by <sup>1</sup>H NMR spectroscopy using 1,4-diacetylbenzene as internal standard. Value within parentheses is that of the yield of the isolated product after column chromatography. [c] 1 mol % of **4**, reaction time: 3 h. L = 2,9-bis(diphenylphosphino)-1,8-naphthyridine (dppn).

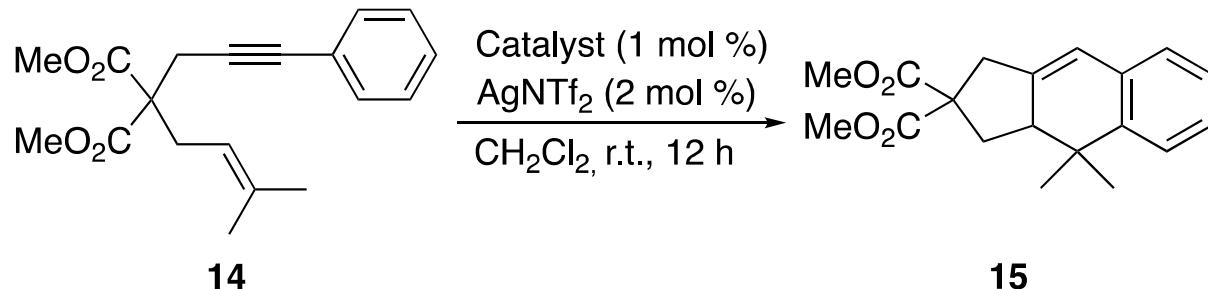
## 催化活性研究



Entry <sup>[a]</sup>	Catalyst	Conv. [%] <sup>[b]</sup>	<b>15</b> <sup>[b]</sup>
1	<b>3, L<sub>2</sub>Au<sub>4</sub>(BAr<sup>F</sup><sub>4</sub>)<sub>4</sub></b>	35	23
2 <sup>[c]</sup>	<b>3, L<sub>2</sub>Au<sub>4</sub>(BAr<sup>F</sup><sub>4</sub>)<sub>4</sub></b>	0	0
3 <sup>[d]</sup>	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	0	0
4	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	100	83 (77)
5	<b>5, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(BF<sub>4</sub>)<sub>2</sub></b>	100	81 (76)
6	<b>6, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(SbF<sub>6</sub>)<sub>2</sub></b>	100	75
7	<b>7, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(OTf)<sub>2</sub></b>	100	81 (74)
8	<b>8, [L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(NTf<sub>2</sub>)<sub>2</sub>]<sub>n</sub></b>	100	68
9	-	39 <sup>[e]</sup>	0

[a] Reaction conditions : **14** (0.1 mmol), cat. (1 mol %), AgNTf<sub>2</sub> (2 mol %), CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL). [b] Conversion and yields were determined by <sup>1</sup>H NMR spectroscopy using 1,4-diacetylbenzene as internal standard. Value within parentheses is that of the yield of the isolated product after column chromatography. [c] NaBAr<sup>F</sup><sub>4</sub> (2 mol %) was added instead of AgNTf<sub>2</sub>. [d] Absence of AgNTf<sub>2</sub>. [e] Partial decomposition of **14**. L = 2,9-bis(diphenylphosphino)-1,8-naphthyridine (dppn).

## 催化活性研究



Entry <sup>[a]</sup>	Catalyst	Conv. [%] <sup>[b]</sup>	<b>15<sup>[b]</sup></b>
1	<b>3, L<sub>2</sub>Au<sub>4</sub>(BAr<sup>F</sup><sub>4</sub>)<sub>4</sub></b>	35	23
2 <sup>[c]</sup>	<b>3, L<sub>2</sub>Au<sub>4</sub>(BAr<sup>F</sup><sub>4</sub>)<sub>4</sub></b>	0	0
3 <sup>[d]</sup>	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	0	0
<b>4</b>	<b>4, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub></b>	100	83 (77)
5	<b>5, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(BF<sub>4</sub>)<sub>2</sub></b>	100	81 (76)
6	<b>6, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(SbF<sub>6</sub>)<sub>2</sub></b>	100	75
7	<b>7, L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(OTf)<sub>2</sub></b>	100	81 (74)
8	<b>8, [L<sub>2</sub>Au<sub>4</sub>Cl<sub>2</sub>(NTf<sub>2</sub>)<sub>2</sub>]<sub>n</sub></b>	100	68
9	-	39 <sup>[e]</sup>	0

[a] Reaction conditions : **14** (0.1 mmol), cat. (1 mol %), AgNTf<sub>2</sub> (2 mol %), CH<sub>2</sub>Cl<sub>2</sub> (1.0 mL). [b] Conversion and yields were determined by <sup>1</sup>H NMR spectroscopy using 1,4-diacetylbenzene as internal standard. Value within parentheses is that of the yield of the isolated product after column chromatography. [c] NaBAr<sup>F</sup><sub>4</sub> (2 mol %) was added instead of AgNTf<sub>2</sub>. [d] Absence of AgNTf<sub>2</sub>. [e] Partial decomposition of **14**. L = 2,9-bis(diphenylphosphino)-1,8-naphthyridine (dppn).

- 发展了一系列过渡金属催化的串联环化反应合成方法学，构建了多种杂环和稠环化合物，并对反应机理进行了探讨，提出了合理的反应机理。
- 主要包括：1) 铑催化的三氮唑与吡咯和吲哚的分子内C-H键官能团化反应（发表于*Angew. Chem. Int. Ed.*，高引用论文，Top 1%，截至2017年8月，106次引用）； 2) 金或银催化的炔丙醇羧酸酯与呋喃、吲哚、氮杂环丙烷的分子内串联环化反应；3) 以PNNP类型配体为原料，合成并表征了一系列多金簇合物，并对其催化活性进行了研究。
- 已发表SCI论文8篇，JCR化学大类一区3篇 (*Angew. Chem. Int. Ed.*, *Chem. Commun.*, *Adv. Synth. Catal.* ), 二区2篇 (*Chem. Eur. J.*, *Organometallics*), 三区3篇 (*Tetrahedron Lett.*, *Tetrahedron Asymmetry*, *ChemistryOpen*)。其中第一作者6篇，第二作者2篇，论文总影响因子40.375，总引用次数216次，他引209次。

## 致谢



Prof. Dr. Min Shi

Prof. Dr. Antonio M Echavarren



European Research Council  
Established by the European Commission



国家自然科学  
基金委员会  
National Natural Science  
Foundation of China

